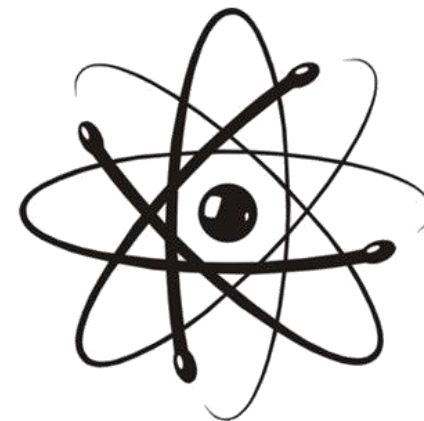


# Mistura de Soluções

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Química



# Mistura de Soluções

A concentração final, resultante da mistura de soluções do mesmo soluto, será uma \_\_\_\_\_ em relação às concentrações das soluções iniciais.

The diagram illustrates the process of mixing two solutions of the same solute (NaCl). On the left, two round-bottom flasks are shown. The first flask contains a solution with a concentration of  $2,4 \text{ mol/L}$  and a volume of  $300 \text{ mL}$ . The second flask contains a solution with a concentration of  $0,8 \text{ mol/L}$  and a volume of  $200 \text{ mL}$ . Both flasks are labeled "NaCl". A plus sign (+) is placed between the two flasks. To the right of the plus sign is an equals sign (=). On the far right, a larger flask is shown, representing the mixture. This flask contains a green liquid and has a label with "C =" and "V =".

# Mistura de Soluções

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**Dados:**

**Solução A**

**M: 2,4 mol/L**

**V: 300 mL**

**Solução B**

**M: 0,8 mol/L**

**V: 200 mL**

Módulo 71

02, 03, 05,

06, 10.

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