

ARSINE

CAS number:	7784-42-1	
Synonyms:	Arsenic hydride, arsenic trihydride, arsenous hydride, hydrogen arsenide, arseniuretted hydrogen	
Chemical formula:	AsH ₃	
Workplace expos	ure standard (interim)	
TWA:	0.005 ppm (0.016 mg/m³)	
STEL:	_	
Peak limitation:	_	
Notations:	Carc. 1A	
IDLH:	3 ppm (9.6 mg/m³)	
Sampling and analysis:	There is uncertainty regarding quantification of the recommended value with currently available sampling and/or analysis techniques.	

Recommendation and basis for workplace exposure standard

An interim TWA of 0.005 ppm (0.016 mg/m³) is recommended to protect for adverse effects to the peripheral nervous system and haemopoietic system in exposed workers.

A review of additional data sources is recommended at the next scheduled review due to the similarity in metabolism of arsine and inorganic arsenic compounds. The review should consider a detailed examination of the carcinogenicity classification and the suitability of the interim TWA.

Discussion and conclusions

Arsine is used in the semiconductor industry and in organoarsenic production. It is a gas under standard conditions and is rapidly absorbed upon inhalation, metabolised and excreted; in contrast to particulate inorganic arsenic compounds, which accumulate in the lungs (ACGIH, 2007; HCTON, 2008). It is unclear if arsine has the same carcinogenic potential as inorganic arsenic compounds, despite the production of similar or identical metabolites upon absorption (ACGIH, 2007; HCOTN, 2008; DFG, 1999). The low pulmonary retention of arsine has been discussed as a property that distinguishes it from other inorganic arsenic compounds (ACGIH, 2017; HCOTN, 2008).

The recommended interim TWA is based on urinary excretion levels that relate to arsine air concentrations below which systemic health effects are not observed in arsenic-exposed workers. The TWA is supported by a haematological NOAEL in rats exposed to arsine (0.025 ppm) to which an uncertainty factor of 5 was applied using a human regression model.

Recommendation for notations

Classified as a category 1A carcinogen according to the Globally Harmonized System of Classification and Labelling of Chemicals (GHS).

Not classified as a skin sensitiser or respiratory sensitiser according to the GHS.



Insufficient data available to recommend a skin notation.

APPENDIX

Primary sources with reports

Source	Year set	Standard					
SWA	Year	TWA: 0.05 ppm (0.16 mg/m³)					
ACGIH	2007	TLV-TWA: 0.005 ppm (0.016 mg/m³)					
the spleen,	liver and kidne s may not apply	to prevent increased risk of haemolysis and haemolysis-related effects of eys. Relationship of lung cancer risk and exposure established for ly to AsH ₃ exposure due to lower pulmonary retention and rapid					
exposed to	airborne partic	combination of studies of urinary excretion levels in workers chronically culate inorganic arsenic compounds, and a haematological NOAEL in rate nty factor of 5 was applied, derived from a human regression model.					
Insufficient	data to derive a	a TLV-STEL, or notations for carcinogenicity, skin, or sensitisation.					
Summary c	of data:						
Human dat							
• No	data confirming	ng carcinogenicity					
 Let 	hal dose: 250 p	ppm (30 min); symptoms of poisoning at 1–3.3 ppm (>1 h)					
		exposure of 0.4 ppm over 4 h non-fatal (adverse effects not specified) in es (1 mg As excreted in urine over 6 d)					
	• Severe anaemia in chronically exposed persons (duration and concentration unspecified); max. average urinary As level 2.3 mg/L, decreased to 0.66 mg/L after 3 d						
	orkers exposed 3 µg/L	to 0–0.015 ppm in breathing zone had mean urinary As levels of					
• Tot Asl adv olig	AsH ₃ exposure tal body As cont H ₃ poisoning in verse effects inc	re of 0.005 ppm correlates with urinary As level of 48 µg/L ntent between 100–150 mg estimated in clinical study of simultaneous a metal refinery workers (14 cases); none of these cases were fatal; included haemolysis and renal damage and oliguria; recovery in all cases cases lasted for 40 d, long-term but reversible renal damage noted in 4					
• Re	maining review	ved studies are of inorganic As exposure; AsH_3 exposure is expected to al effects due to metabolic generation of As from AsH_3					
0	slight reduction of 300 µg As mean urinary	ion in peripheral nerve conduction in workers with mean daily absorption s versus control group. y levels in exposed workers was 71 μg/L compared with 7 μg/L in					
0		renal disease increased in exposed workers with urinary As levels of estimated inhalational exposure and duration not specified).					
Animal data							
		targets haemopoletic system					
	d); some haem	Im for haematological changes (rats, mice, and hamsters, inhalation, natological changes were observed in rats at this concentration, these transient and did not constitute a LOAEL.					

ional data: MAK of 0.05 ppm animals or huma vailable for acute haematological l be a LOAEL due A NA A NA 008 — ional data: han air, decompos	e toxicity, effects on skin, allergenic effects or carcinogenicity NOAEL of 0.025 ppm (rats, mice) used to derive TLV-TWA by to measurable haematological changes at this concentration. 4 4					
MAK of 0.05 ppm animals or huma vailable for acute haematological l be a LOAEL due A NA A NA DOB — fonal data: han air, decompos	 ans. toxicity, effects on skin, allergenic effects or carcinogenicity NOAEL of 0.025 ppm (rats, mice) used to derive TLV-TWA by to measurable haematological changes at this concentration. 4 					
haematological l be a LOAEL due A NA A NA D08 — fonal data: han air, decompos	NOAEL of 0.025 ppm (rats, mice) used to derive TLV-TWA by to measurable haematological changes at this concentration.					
be a LOAEL due A NA A NA DOB — ional data: nan air, decompos	to measurable haematological changes at this concentration.					
A NA 2008 — Tonal data: han air, decompos	4					
008 — ional data: nan air, decompos	·					
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ional data: nan air, decompos						
an air, decompos	and an evenesure to light or mointure					
	and an averaging to light or mainture					
s rapidly in water	 Heavier than air, decomposes on exposure to light or moisture. 					
	r to other As compounds					
 No data available on carcinogenicity and genotoxicity in humans or animals 						
 Recommends classification as suspected human carcinogen due to known carcinogenicity of other As compounds in humans 						
• Evidence indicates As compounds can cause clastogenic effects in humans and animals, but results do not suggest mutagenic mechanism of action.						
 Inhalational study with exposed rabbits (20–26 mo) reported mesothelioma in one animal (no further information provided); study considered inadequate for evaluation due to insufficient reporting. 						
a r	al study with exp information prov					

Source		Year	Additional information
NICNAS	~	2013	 Substance should be considered carcinogenic to humans based on IARC (2012) evaluation No acute or chronic dermal toxicity data available.
IARC	✓	2012	 AsH₃ distinct from other As compounds in toxicology but sparsely discussed AsH₃ produces same metabolites as other As compounds Overall, As and inorganic As compounds are classified as carcinogenic to humans, regardless of their mechanism of carcinogenic action The mechanism of carcinogenicity of AsH₃ not discussed.
US EPA	√		 Clinical signs of acute exposures include abdominal pain, haematuria, jaundice, headache, malaise, weakness and GI distress accompanied by nausea and vomiting.

Source		Year	Additional information	
US NIOSH	✓	1994	•	IDLH based on acute inhalation toxicity data in humans.

Carcinogenicity — non-threshold based genotoxic carcinogens

Is the chemical mutagenic?

No

The chemical is not a non-threshold based genotoxic carcinogen.

Notations

Source	Notations
SWA	—
HCIS	Carcinogenicity – Category 1A
NICNAS	Carc. Cat. 1A
EU Annex	—
ECHA	NA
ACGIH	—
DFG	—
SCOEL	NA
HCOTN	Carcinogenicity – category 2
IARC	Carcinogenicity – Group 1
US NIOSH	_

NA = not applicable (a recommendation has not been made by this Agency); — = the Agency has assessed available data for this chemical but has not recommended any notations

Skin notation assessment

Calculation

Insufficient data available to assign a skin notation.

IDLH

Is there a suitable IDLH value available? Yes

Additional information

Molecular weight:	77.95
Conversion factors at 25°C and 101.3 kPa:	1 ppm = 3 µg/m³; 1 mg/m³ = 0.313 ppm
This chemical is used as a pesticide:	
This chemical is a biological product:	

Molecular weight:	77.95		
Conversion factors at 25°C and 101.3 kPa:	1 ppm = 3 µg/m³; 1 mg/m³ = 0.313 ppm		
This chemical is used as a pesticide:			
This chemical is a by-product of a process:	\checkmark		
A biological exposure index has been recommended by these agencies:	✓ ACGIH □ DFG □ SCOEL		

Workplace exposure standard history

Year	Standard	
Click here to enter year		

References

American Conference of Industrial Hygienists (ACGIH[®]) (2018) TLVs[®] and BEIs[®] with 7th Edition Documentation, CD-ROM, Single User Version. Copyright 2018. Reprinted with permission. See the <u>TLVs[®] and BEIs[®] Guidelines section</u> on the ACGIH website.

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