

# SODIUM HYDROXIDE

**CAS number:** 1310-73-2

Synonyms: Caustic soda, lye

Chemical formula: NaOH

Structural formula: —

Workplace exposure standard (retained)

TWA: — STEL: — Peak limitation: 2 mg/m<sup>3</sup> Notations: — IDLH: 10 mg/m<sup>3</sup>

**Sampling and analysis**: The recommended value is quantifiable through available sampling and analysis techniques.

## Recommendation and basis for workplace exposure standard

A peak limitation of 2 mg/m<sup>3</sup> is recommended to protect for acute irritation of the eyes, skin and mucous membranes in exposed workers.

# **Discussion and conclusions**

Sodium hydroxide is a soluble, strong base, used in numerous industries such as pulp and paper, soap and detergents, cellophane and textiles, etching and electroplating.

The critical effects are identified as irritation of the eyes, skin and mucous membranes (ACGIH, 2001).

Much of the data from humans generally relates to accidental or suicidal ingestion with effects include necrosis, principally of the mouth, oesophagus, gastric mucosa, hypersalivation, emesis, cardiovascular collapse, tracheal obstruction and dyspnoea, retching and severe pain (ACGIH, 2018). The main effects of exposure are local irritation and corrosion. Very limited inhalation data are available with irritant effects of caustic mists encountered at 1 to 40 mg/m<sup>3</sup>, with 2 mg/m<sup>3</sup> considered 'noticeably but not excessively' irritating (ACGIH, 2018).

The peak limitation of 2 mg/m<sup>3</sup> as assigned by SWA, ACGIH (2018) and HCOTN (2000) is recommended to be retained and is generally protective of irritation to eyes, skin and mucous membranes in exposed workers.

# **Recommendation for notations**

Not classified as a carcinogen according to the Globally Harmonized System of Classification and Labelling of Chemicals (GHS).

Not classified as a skin sensitiser or respiratory sensitiser according to the GHS.



A skin notation is not recommended based on the available evidence.



# APPENDIX

#### Primary sources with reports

Source	Year set	Standard			
SWA	1991	Peak limitation: 2 mg/m <sup>3</sup>			
ACGIH	2001	TLV-Ceiling: 2 mg/m³			
TLV-Ceiling workers. TL upper respi Summary o Human dat	g recommended _V is based on a ratory tract irrita of data: a:	to minimise irritation to eyes, skin and mucous membranes in exposed a concentration that produces a noticeable, but not excessive, ocular and ition.			
• Nu	merous cases o	f accidental and suicidal poisonings			
<ul> <li>Ing hyp reto</li> </ul>	estion causes n persalivation, em ching and sever	ecrosis, principally of the mouth, oesophagus and gastric mucosa, nesis, cardiovascular collapse, tracheal obstruction and dyspnoea, e pain			
• Dea	ath occurs due t	o shock, infection of corroded tissue, pulmonary necrosis or asphyxia			
• Re	port of one oesc	phageal carcinoma due to localised stricture following ingestion			
• 2 m	• 2 mg/m <sup>3</sup> considered 'a concentration that is noticeably, but not excessively, irritant':				
0	irritant effects of times provided	of caustic mists encountered at 1–40 mg/m <sup>3,</sup> (no exposure or sampling )			
<ul> <li>Bur airt</li> </ul>	rning and redne	ss of nose, throat or eyes reported in workers undertaking cleaning with tions of 0.005–0.7 mg/m <sup>3</sup> :			
0	solvents also p	resent up to 780 mg/m <sup>3</sup> (no exposure or sampling times provided)			
<ul> <li>Sevent</li> <li>furt</li> </ul>	vere eye injury r her information	eported in workers exposed to high concentrations (as dust or liquid); no			
<ul> <li>App cell mir</li> </ul>	olication of 1 g e ls in outer layer ns.	equivalent to forearms of volunteers (15–180 min) caused dissolution of of epidermis, progressing to oedema and destruction of epidermis in 60			
Animal data	a:				
• Ing	estion in dogs c	auses haemorrhagic gastritis and oesophageal strictures			
<ul> <li>Ora</li> </ul>	al intubation in ra	abbits:			
0	4% solution ca	used mucosal and submucosal necrosis			
0	12% solution e	roded into muscle			
0	28% solution c	aused perforation			
0	similar results i	In Cats			
• 5%		on applied to rappit skill (4.11) caused severe necrosis			
• Ra	is innaming aeros	sols so minira suffered pulmonary damage (concentrations not provided).			
Insufficient	data to recomm	end skin, SEN or carcinogenicity notations.			



Source	Year set	Standard		
DFG	1999	Not assigned		
MAK withdraw from available Summary of a	n as cannot io data. dditional data	dentify exposure concentration where irritation will certainly not occur		
Corro	sive to skin ar	nd eyes		
Poorly	documented	reports of irritation to airways and eyes from exposure at 0.5 mg/m <sup>3</sup>		
<ul> <li>No increased mortality from malignant and non-malignant diseases amongst 265 workers at factory producing chlorine, employed up to 30 y</li> </ul>				
• Negative in mutagenicity tests; clastogenic effects <i>in</i> vitro following increased pH of the culture medium with metabolic activation only				
<ul> <li>'Few a</li> </ul>	available tests	indicate not genotoxic at physiological pH:		
o lo af	cal carcinogei ter dermal ap	nic effects observed in humans following severe poisoning and animals plication due to cell regeneration following tissue damage and scarring		
• 48 h p	atch test in vo	olunteers:		
0 <b>n</b> (	o irritation fron	n 1% aqueous solution but irritant skin reactions at ≥2%.		
SCOEL	NA	NA		
No report				
OARS/AIHA	NA	NA		
No report				
HCOTN	2000	Ceiling limit: 2 mg/m³		
Summary of a	dditional data			
Most	dominant effe	ct is local irritation and corrosion		
• 10% s	olution corros	sive in <i>in vitro</i> skin corrosion test		
<ul> <li>TWA exposure levels of 265 workers survey (cited in DFG, 1999) estimated to be 0.5–2.0 mg/m<sup>3</sup>:</li> </ul>				
<ul> <li>medical aid was sought most for skin contact and least for inhalation</li> </ul>				
<ul> <li>2 h exposure at 65 mg/m<sup>3</sup> (nose only) or 250–3,200 mg/m<sup>3</sup> (whole body) in young and adult rats:</li> </ul>				
o larynx target organ				
<ul> <li>no effects on nasal turbinates, lungs, oesophagus or stomach</li> </ul>				
<ul> <li>no effects at 65 mg/m<sup>3</sup></li> </ul>				
$\circ$ 6/11 died at 3,200 mg/m <sup>3</sup>				
<ul> <li>aerosols mainly consisted of Na<sub>2</sub>CO<sub>3</sub></li> <li>LD = 4.250 map (up (ap bits down of))</li> </ul>				
<ul> <li>LD<sub>50</sub>: 1,350 mg/kg (rabbits, dermal)</li> <li>Committee concluded insufficient information to comment on OEI</li> </ul>				

### Secondary source reports relied upon

Source		Year	Additional information
US NIOSH	✓	1994	<ul> <li>REL 2 mg/m<sup>3</sup> ceiling</li> <li>IDLH 10 mg/m<sup>3</sup>.</li> </ul>



### Carcinogenicity — non-threshold based genotoxic carcinogens

Is the chemical mutagenic?
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No

The chemical is not a non-threshold based genotoxic carcinogen.

### **Notations**

Source	Notations
SWA	-
HCIS	-
NICNAS	-
EU Annex	NA
ECHA	-
ACGIH	-
DFG	-
SCOEL	NA
HCOTN	-
IARC	NA
US NIOSH	—

NA = not applicable (a recommendation has not been made by this Agency); — = the Agency has assessed available data for this chemical but has not recommended any notations

#### Skin notation assessment

Calculation	
Adverse effects in human case study:	
Dermal LD <sub>50</sub> ≤1000 mg/kg:	no
Dermal repeat-dose NOAEL ≤200 mg/kg:	
Dermal $LD_{50}$ /Inhalation $LD_{50}$ < 10:	
In vivo dermal absorption rate >10%:	
Estimated dermal exposure at WES >10%:	
	a skin notation is not warranted

#### IDLH

Is there a suitable IDLH value available? Yes



## Additional information

Molecular weight:	39.99		
Conversion factors at 25°C and 101.3 kPa:	1 ppm = Number mg/m <sup>3</sup> ; 1 mg/m <sup>3</sup> = Number ppm		
This chemical is used as a pesticide:			
This chemical is a biological product:			
This chemical is a by-product of a process:			
A biological exposure index has been recommended by these agencies:			

## Workplace exposure standard history

Year	Standard	
Click here to enter year		

## References

American Conference of Industrial Hygienists (ACGIH<sup>®</sup>) (2018) TLVs<sup>®</sup> and BEIs<sup>®</sup> with 7<sup>th</sup> Edition Documentation, CD-ROM, Single User Version. Copyright 2018. Reprinted with permission. See the <u>TLVs<sup>®</sup> and BEIs<sup>®</sup> Guidelines section</u> on the ACGIH website.

Deutsche Forschungsgemeinschaft (DFG) (1999) Sodium hydroxide – MAK value documentation.

Health Council of the Netherlands (HCOTN) (2000) Sodium hydroxide. Health-based calculated occupational cancer risk values. The Hague: Health Council of the Netherlands; publication no. 2000/15OSH/015.

US National Institute for Occupational Safety and Health (NIOSH) (1994) Immediately dangerous to life or health concentrations – Sodium hydroxide.

US National Institute for Occupational Safety and Health (NIOSH) (2011) NIOSH Skin Notation Profiles: Sodium Hydroxide (NaOH).