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BIOCHEMICAL COMPARATIVE STUDIES FOR SOME TRACE ELEMENTS IN TYPE2 DIABETIC PATIENTS WITH AND WITHOUT METABOLIC SYNDROME

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ABSTRACT

The aim of this study was to investigate the role of trace elements (copper, iron and zinc) as oxidants and antioxidants in type 2 diabetic patients and metabolic syndrome. Blood samples were obtained from (110) patients with type 2 diabetes with and without metabolic syndrome, as well as (50) healthy subjects as a control group. They divided into three groups as the following: Group A(control):- Included fifty healthy subjects aged (35-65 years). Group B (Type 2 diabetics):- Included sixty four patients with type 2 diabetes mellitus aged (35-65). Group C (diabetics type 2 with metabolic syndrome):- Included forty six patients with metabolic syndrome aged (35-65). **Results:** The results show a presence of a significant increase in copper and iron in all groups of patients in type 2 diabetes and type 2 diabetes with metabolic syndrome in comparison with control group. But, zinc levels showed a significant decrease in all groups of patients in type 2 diabetics with metabolic syndrome in comparison with type 2 diabetics with metabolic syndrome in comparison with type 2 diabetics group. While, Fe showed a significant increase in all patients groups of patients in type 2 diabetes in comparison with type 2 diabetics with metabolic syndrome group.

KEYWORD: The aim of this study antioxidants in type 2 group.

INTRODUCTION

Diabetes mellitus type 2 is the predominant form of diabetes and accounts for 90% of all cases of diabetes mellitus.^[1] DM 2 is heterogenous disorders caused by a group of genetic factors related to insulin resistance, impaired insulin secretion, and environmental factors such as over eating obesity, lack of exercise, and stress as well as aging. [2] It is typically a multifactorial disease involving multiple genes and environmental factors to varying extents. Diabetes type 2 is the common form of idiopathic diabetes and is characterized by a lack of the need for insulin to prevent ketoacidosis. It is not an autoimmune disorder, in most patients the susceptible genes that predispose to DM 2 have not been identified .This could be due to the heterogeneity of the genes responsible for the susceptibility to NIDDM. The pathogenesis of diabetes mellitus type 2under, normal physiological conditions, plasma glucose concentrations are maintained within a narrow range, although wide fluctuations in supply and demand, through a tightly regulated and dynamic interaction between tissue sensitivity to insulin (especially in liver) and insulin secretion.^[4] Diabetes mellitus type 2 include some mechanisms which work on broken of regulation between tissue sensitivity to insulin which leads to two major pathological defects are impaired insulin secretion through a dysfunction of the pancreatic β -cell cells and impaired insulin action through insulin resistance. [3] The relation very close between obesity and DM2, both go hand in hand. Abundant of the complications in both diseases are due to underlying primary causes insulin resistance and metabolic syndrome. [5] Metabolic syndrome mean cluster of signs and symptoms occurring in a complex that can increase the risk of cardiovascular disease, Type 2 Diabetes and all-cause mortality. It containing of a cluster of risk factors including elevated Fasting Blood Glucose or Impaired Glucose Tolerance, abdominal obesity, elevated blood pressure, dyslipidemia to mention a few. [6]

Trace elements

Trace elements are primary nutrients, chemicals with functions of regulatory, immunologic, and antioxidantas they act as essential components or cofactors of some enzymes throughout metabolism.^[7]

Copper is the more ample element in the human body. [8] Cu is required for several biological functions. It is required for the catalytic activity of superoxide dismutase (SOD) that participates in the protection of cells from superoxide radicals. [9] It is well known that copper is effective on oxidative stress. Copper is a potent cytotoxic element in its free form due to its redox chemistry. It readily participates in Fenton and Heiber Weiss reactions to produce reactive oxygen species such as hydroxyl radical stress. [10]

Iron (Fe) is an primary transition metal required for thesynthesis of two important functional proteins such as hemoglobin and myoglobin, which are involved in the transport of molecular oxygen during respiration. [11] It is also required in the elastin production along with Zn and ascorbic acid and collagen synthesis. [12]

Zinc (Zn) is a primary mineral that is a component of more than 200 enzymes such as protein kinase C ,carboxy-peptidase A, B, superoxide dismutase and alkaline phosphatases. [13] zinc acts as an antioxidant by protecting the sulfhydryl groups of proteins and enzymes versus free radical damage in the body [14], the antoxidantsynthesis impairs when deficiency of zinc, leading to increased oxidative stress. [15]

Patients and Method

Design of Study

This study conducted at AL-Hussein Education Hospital and The Special Center of The Endocrine Glands and Diabetes inThi-Qar from Biochemistry Laboratory in College of Science, at the period between 4/11/2015 to 5/5/2016. It included (160) cases, (50) control and (110) patients.

Table 1: Data of Controls and Patients Groups

Groups	<i>N0</i> .
Controls	50
Patients	110

There were (160) male and female subjects, control and diabetic of type 2 with trace elements as oxidants or antioxidants aged (35-65) years were included in this study. They divided into three groups as the following:-

Group A (Control): Included fifty (50) healthy subjects aged (35-65).

Group B (*Diabetes type 2*): Included sixty four (64) patients with diabetes mellitus type 2 aged (35-65).

Group C (Diabetes Type 2 with Metabolic syndrome) Included forty six (46) patients with metabolic syndrome aged (35-65).

Collection of Blood Samples

About (10mL) of blood samples from out treated hypertension patients and controls were taken and

allowed to clot at room temperature in empty disposable tubes centrifuge to separate it in the centrifuge at 3000 rotor per minute (rpm)for 10 min, the serum samples were separated and stored at (-20°C) for later measurement of biochemical parameters, unless used immediately.

BIOCHEMICAL PARAMETERS

Serum Copper

The cupric ions react with the chromogen Di-BrPAESA forming a blue compound, which intensity is proportional to the concentration present in the sample. [16][17][18]

Serum Iron

After dissociation of iron-transferrin bound in acid medium, ascorbic acid reduces Fe^{3+} iron into Fe^{2+} iron . Fe²⁺ iron then form a coloured complex with 3-(2-Pyridyl) - 5, -6-difuryl-1, -2, -4-triazine-disulfonate (Ferene). The absorbance thus measured at 600 nm (580-620) is directly proportional to the amount of iron in the specimen. [19]

Serum Zinc

Zinc reacts with the chromogen present in the reagent forming a coloured compound which colour intensity is proportional to the zinc concentration present in the sample. [16][20]

Statistical Analysis

Statistical analysis was done using the software SPSS version 17.0; the results were expressed as mean \pm standard deviation (mean \pm SD). One way ANOVA-test was used to compare parameters in different studied groups. P-values (P \leq 0.05) were considered statistically significant.

RESULT AND DISCUSSION

Clinical and Characteristic Features of the Studied Groups

There are 160 study cases included in the present study, with difference between clinical characteristic diabetics type 2 patients Groups and healthy control group in each (Hemoglobin A1c (HbA1c), Trace Elements Copper (Cu), Iron (Fe), Zinc (Zn)) and without difference in each (Age, Sex and BMI). as shown in table 3-1.

Table: 2 Characteristic data for all studied groups.

		Characteristic						
Groups	N	Age years Mean±SD	Sex (M/F)	SBP (mmHg) Mean ±SD	DBP (mmHg) Mean ±SD	TG mg/dl Mean±SD	FBG mg/dl Mean±SD	BMI Kg/m² Mean±SD
Control	50	45.7 ± 7.53	25/25	99.9±0.76	75.04 ± 0.28	106.32±19.65	87.88±5.96	25.46±1.74
Diabetics Type 2	64	51.56± 8.11	37/27	110.06±0.3	77.75 3±0.53	126.68±28.05	201.95±67.83	28.93±4.18
Diabetics Type2 With metabolic syndrome	46	51.18± 8.44	17/29	180.93±8.12	87± 1.66	180.54± 7.83	180.75±61.12	33.78±4.43

TRACE ELEMENTS Serum Copper Concentration

Table (2) shows a significant increase in concentrations of serum copper in MS group in comparison with DM 2 group ($P \le 0.05$). Also there is a significant increase in concentrations of serum copper in all patients groups in comparison with control group ($P \le 0.05$).

The present study are matched with the results of study Supriya^[10] and Sarkar.^[21] The reason for the increase of concentration of copper is due to hyperglycemia that may stimulate glycation^[22] which will stimulate release of copper from copper rich compounds such as ceruloplasmin^[23] and the impaired synthesis of this transport protein^[24] and this accelerate the oxidative stress and can result in the formation of AGEs.^[22]

Serum Iron Concentration

Table (3) shows a significant increase in concentrations of serum iron in DM2 group in comparison with MS group (P≤0.05) and there is a significant increase in concentrations of serum iron in all patients groups in comparison with control group ($P \le 0.05$). Our results are matched with the results of study of Swaminthan. [25] Evidence shows that the accumulation of body iron is related to the Type 2 diabetes and metabolic syndrome in general populations. [26] Increase iron concentration due to iron is strong pro-oxidant that catalyzed the formation of hydroxyl radicals. [27], this radicals are powerful prooxidants that attack cellular membrane lipids DNA and proteins. It has been hypothesized that the formation of hydroxyl radicals catalyzed by iron initially contributes to insulin resistance and subsequently to decreased insulin secretion and then to the development diabetes mellitus type 2.^[28]

Serum Zinc Concentration

Table (4) showed a significant increase in concentrations of serum zinc in MS group in comparison with control group (P≤0.05). However, there was a significant reduction in the serum concentration of zinc in DM 2 group as compared to control group ($P \le 0.05$). The present study are matched with the results of study Pujar^[29] that showed significant decrease in serum zinc levels in type 2 diabetic patients. The present study showed that serum zinc was associated with the components of metabolic syndrome, where association between zinc and metabolic syndrome in our study are in agreement with the results of other study Arnaud. [30] Zn has a biphasic effect in that it is required for insulin storage and cellular binding, although high concentrations can lead to a reduction in insulin release.[31]

Table: 3: Serum copper concentrations for studied groups

Groups	Cu μmol/L Mean±SD
Control	22.98 ± 9.32^{c}
DM2	25.89 ± 2.01^{b}
DM2 with MS	26.36 ± 1.95^{a}
LSD	1.26

Table 4: Serum iron concentrations for studied groups

Groups	Fe μmol/L Mean±SD
Control	21.79 ± 7.02^{b}
DM2	23.91± 15.49 ^a
DM2 with MS	22.89 ± 15.50^{b}
LSD	3.10

Table: 5 Serum zinc concentrations for studied groups

Groups	Zn µmol/L Mean±SD
Control	14.44 ± 1.48^{b}
DM2	9.03 ± 1.32^{c}
DM2 with MS	30.60 ± 6.98^{a}
LSD	0.90

CONCLUSION

- 1. There are a disturbances in trace elements that are associated with metabolic syndrome.
- 2. There is a disorder in trace elements metabolism in patients with diabetes mellitus type 2. The correction for this disorder might have significant value for control of diabetics and prevention of complications.

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