

**AN EFFICIENT ULTRASOUND ASSISTED ASPARTIC ACID CATALYZED
MULTICOMPONENT SYNTHESIS OF ISOXAZOLE-5(4H)-ONES**

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ABSTRACT

A green and efficient protocol was developed for the multicomponent synthesis of isoxazoles via the Ethanol:water mediated multicomponent reaction of aromatic aldehyde, ethyl acetoacetate, and hydroxylamine hydrochloride in ultrasonic conditions catalyzed by aspartic acid at ambient conditions. It is noteworthy that the aspartic acid promoted one-pot three-component process under ultrasonic conditions was observed to be very good and satisfactory.

KEYWORD:- Isoxazole, Aspartic acid, Ultrasound.

INTRODUCTION

Isoxazoles are aromatic heterocycles with five members that have nitrogen and oxygen atoms next to each other. Many naturally occurring substances and physiologically active molecules have the isoxazole ring structure.^[1] Numerous biological activities, including anti-inflammatory, antifungal, antiplatelet, anti-Alzheimer, anti-HIV and analgesic properties, are exhibited by isoxazole derivatives.^[2-5] Multicomponent reactions are processes that create a desired single product using readily available or commercially available starting components. There are far fewer processes involved in obtaining the target chemicals in a single pot. In order to produce heterocyclic compounds, multicomponent reactions have drawn a lot of attention in fields of combinatorial chemistry and medicinal chemistry.^[6-8] In both academia research and industry, the use of ultrasonic irradiation to accelerate organic processes has long been recognized. Thus ultrasound irradiation has been utilized in organic chemistry, medicinal chemistry and material sciences. Ultrasound's chemical and physical impacts are caused by acoustic cavitations, which include the creation, expansion, and implosive collapse of bubbles in a liquid.^[9-11] A number of catalysts have been used in one-pot MCRs of benzaldehyde, ethyl acetoacetate, and hydroxylamine for the synthesis of isoxazole scaffold including potassium hydrogen phthalate,^[12] N-bromosuccinimide,^[13] pyridinium p-toluenesulfonate,^[14] 1,4-aminobenzene-1-sulfonic acid,^[15] iodine,^[16] pyridine^[17] phosphotungstic acid^[18] etc.

However lower yield, utilization of harsh conditions, the use of hazardous chemicals and solvents, and the need for longer reaction times are some of the drawbacks of the reported synthetic procedures. Therefore, there is a great need to develop efficient and environmentally friendly processes for the synthesis of Isoxazole-5(4H)-ones. So in this protocol and in continuation of our previous works towards development of new convenient protocols,^[19-25] we present here aspartic acid catalyzed, one-pot, three-component method for the quick synthesis of substituted Isoxazole-5(4H)-one scaffolds using a aromatic aldehydes, ethyl acetoacetate, and hydroxylamine hydrochloride in EtOH: water as a solvent under ultrasonic conditions at room temperature.

RESULTS AND DISCUSSION

To produce the desired aryl-3-methylisoxazol-5(4H)-ones, a one-pot multicomponent reaction was utilized in an aqueous ethanol medium with suitable substrates catalyzed by aspartic acid in ultrasonic conditions (Figure 1).

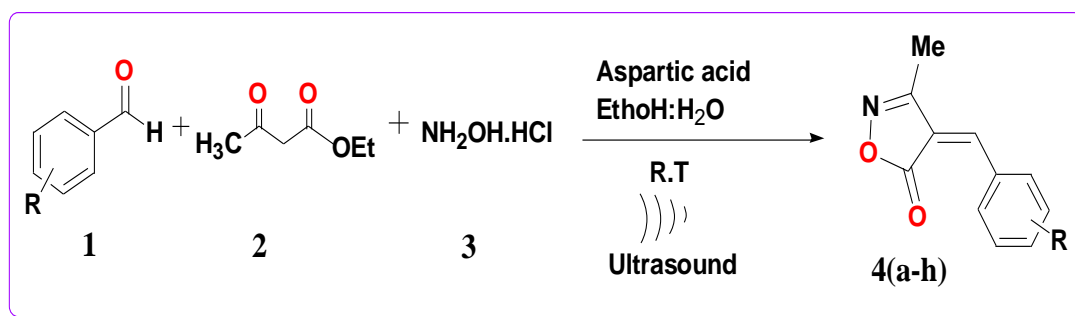


Figure 1: An Ultrasound assisted Aspartic acid catalyzed synthesis of arylmethylenedene-isoxazole-5(4H)-ones.

Initially, a model reaction was carried out with benzaldehyde, ethyl acetoacetate, and hydroxylamine hydrochloride. A variety of parameters were examined, including various solvents, catalyst loading, conventional and ultrasonic conditions and the reaction progress was monitored by TLC.

Our focus of was to incorporate an efficient, green and more convenient methodology for the multicomponent synthesis of these isoxazoles. When the reaction was carried out in ethanol alone using aspartic acid (20 mol%) as a catalyst at room temperature gave 85% of yield (Table 1, entry 1). When water alone was used as a solvent keeping other parameters constant poor yield of the corresponding product was obtained (Table 2, entry 2). Similar good results were obtained when EtOH: H₂O (1:1) used instead of ethanol alone (Table 1, entry 3). In order to improve the results, instead of conventional procedure, ultrasound assisted protocol was employed. This showed a noticeable impact on the yield and Improvement in the reaction profile in terms of reaction time (40 min) and yield (92%) was observed at ultrasound conditions (Table 1, entry 4). For

optimization of catalyst loading, performing the same reaction in the presence of 10 mol% aspartic acid catalysts in an aqueous ethanol medium at R.T. and under ultrasound irradiation led to the formation of less amount of 4a after 2 hr (Table 1, entry 5). Whereas increasing the aspartic acid concentration to 30 mol% did not offers noticeable change in the results as compared to using 20 mol% of the catalyst (Table 1, entry 6). Therefore it was conclude that 20 mol% of aspartic acid catalyst was sufficient to obtain the desired product 4a in 92 % yield after 40 min stirring at room temperature and under ultrasound conditions (Table 1, entry 4). No improvement was observed in the reaction profile when different solvents were utilized in the ultrasonic protocol such as acetonitrile and methanol (Table 2, entries 7, 8). Thus, the optimum reaction condition for the reaction was noted, when the benzaldehyde (1a), hydroxylamine hydrochloride (2) and ethyl acetoacetate (3) was sonicated in EtOH:H₂O (1:1) at room temperature in the presence of 20 mol% aspartic acid catalyst which afforded desired isoxazole derivative (4a) in 92 % yield in 40 min (Table 2, entry 4).

Table 1: Optimization of reaction conditions for the synthesis of arylmethylenedene-isoxazole-5(4H)-one.

| Entry | Condition | Solvents ^a | Aspartic Acid(mol%) | Temperature (°C) | Time (min) | Yield (%) ^b |
|-------|--------------|-----------------------------|---------------------|------------------|------------|------------------------|
| 1 | Conventional | EtOH | 20 | R.T | 75 | 85 |
| 2 | Conventional | H ₂ O | 20 | RT | 120 | 62 |
| 3 | Conventional | EtOH:H ₂ O (1:1) | 20 | R.T | 75 | 85 |
| 4 | Ultrasound | EtOH:H ₂ O (1:1) | 20 | RT | 40 | 92 |
| 5 | Ultrasound | EtOH:H ₂ O (1:1) | 10 | RT | 120 | 60 |
| 6 | Ultrasound | EtOH:H ₂ O (1:1) | 30 | RT | 40 | 92 |
| 7 | Ultrasound | Acetonitrile | 20 | RT | 120 | 60 |
| 8 | Ultrasound | MeOH | 20 | RT | 120 | 65 |

^a 5mL, ^b Isolated yields.

To test the generality of the developed protocol different derivatives are prepared, as shown in Table 2, and it was

found that this new procedure works with a variety of substrates present on benzene ring of aldehydes. The

reaction was carried out with six different types of substituted aromatic aldehydes and it was observed that

yields are consistently very good regardless of the substituent's type and position (Table 2, entries 1-6).

Table 2: An Ultrasound assisted Aspartic acid catalyzed synthesis of arylmethyldene-isoxazole-5(4H)-ones.

| Entry | R | P | Yield (%) ^a | Time (min) | M. p (°C) | |
|-------|---------------------|----|------------------------|------------|-----------|-------------------------------|
| | | | | | Observed | Literature ^[12-18] |
| 1 | H | 4a | 92 | 40 | 140-142 | 141-143 |
| 2 | 4-CH ₃ O | 4b | 91 | 50 | 176-178 | 175-177 |
| 3 | 4-CH ₃ | 4c | 90 | 50 | 132-134 | 132-134 |
| 4 | 4-NO ₂ | 4d | 88 | 60 | 181-183 | 183-184 |
| 5 | 4-Cl | 4e | 86 | 60 | 122-124 | 128-130 |
| 6 | 4-OH | 4f | 89 | 55 | 222-224 | 213-215 |
| 7 | 2-OH | 4g | 87 | 45 | 192-194 | 197-199 |
| 8 | 4-NO ₂ | 4h | 86 | 60 | 138-140 | ----- |

^aIsolated yields..

CONCLUSION

We have developed a new protocol for the synthesis of different substituted isoxazole derivatives (4a-4g) using a one-pot three-component reaction between substituted aromatic aldehydes, ethyl acetoacetate, and hydroxylamine hydrochloride under ultrasound irradiation in the presence of aspartic acid as an efficient organo-catalyst in aqueous ethanol medium under ultrasonic irradiation. The current methodology offers several advantages, including green ultrasonic conditions, the use of a safe and easily available aspartic acid organo-catalyst, ethanol: water as a green medium, rapid reaction completion, and maximum yields.

EXPERIMENTAL SECTION

MATERIAL AND METHODS

Chemicals were purchased from SD Fine Chemical Companies. The physical properties of the synthesized derivatives were confirmed by comparing the physical data with those described in the literature. A Bruker Advance NEO spectrometer was used to record NMR spectra, and mass spectra were recorded on the Waters GC-MS spectrophotometer. Silica gel TLC plates were used to monitor the reactions and the purity of the products."

Typical Procedure for the synthesis of arylmethyldene-isoxazol-5(4H)-ones (4a-h)

In a 25mL RBF an aromatic aldehyde (1 mmol) was added to a well stirred mixture of ethyl acetoacetate (1 mmol), hydroxylamine hydrochloride (1 mmol), and aspartic acid (20 mol%) in EtOH:H₂O (5 mL, 1:1), and the mixture was agitated at room temperature in the Ultrasound irradiated bath for the required amount of time as indicated by thin layer chromatography (Ethyl acetate:Hexane, 2:8). After reaction completion, 5 mL of water was added, and the resulting solid crude precipitate was filtered. To obtain the pure product, the crude product was recrystallized from a mixture of EtOH-H₂O (4:1).

Spectral data

3-Methyl-4-(4-methoxyphenyl)methylene-isoxazole-5(4H)-one (4a): M.p. 175-178 °C; ¹H NMR (500 MHz,

CDCl₃) δ(ppm): 2.27 (3H, s, CH₃), 3.91 (3H, s, OCH₃), 7.01 (2H, d, J = 8.9 Hz, HAr), 7.34 (1H, s, =CH), 8.44 (2H, d, J = 8.9 Hz, HAr); ¹³C NMR(125 MHz, CDCl₃) δ(ppm): 11.63, 55.71, 114.66, 116.37, 125.83, 136.96, 149.32, 161.27, 164.61, 168.77, Mass, m/z: 218.08 [M+1].

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