

**CHITOSAN NANOPARTICLES: VERSATILE CARRIERS FOR DRUG DELIVERY AND BEYOND****Shaharban K.\*, Sijo Pattam, Muhammed Harshad P. and Muhammed Sajir K.**

Department of Pharmaceutics, National College of Pharmacy, KMCT Medical College Campus, Manassery PO, Kozhikode 673602, Kerala, India.

**\*Corresponding Author: Shaharban K.**

Department of Pharmaceutics, National College of Pharmacy, KMCT Medical College Campus, Manassery PO, Kozhikode 673602, Kerala, India.

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**ABSTRACT**

Chitosan nanoparticles have emerged as a promising platform for drug delivery due to their unique properties such as biocompatibility, biodegradability, and mucoadhesive nature. Derived from chitin, these nanoparticles offer significant advantages in improving drug solubility, permeability, and targeted delivery. By leveraging their ability to crosslink with anions and interact with cell membranes, CSNPs enable controlled and sustained drug release while enhancing therapeutic efficacy. Various synthesis methods, including ionic gelation and emulsification, have been explored to optimize their applications across diverse biomedical fields. This review delves into the preparation techniques, advantages over other systems, and their potential for site-specific drug delivery.

**KEYWORDS:** Chitosan nanoparticle, controlled release, drug delivery, biocompatibility, ionic gelation.**INTRODUCTION**

Nanomedicine and nano delivery systems are rapidly evolving fields that involve the use of nanoscale materials as diagnostic tools or as carriers for therapeutic agents. These systems enable precise and controlled delivery of drugs to specific targeted sites within the body, enhancing both diagnosis and treatment.<sup>[1]</sup> Various drug delivery systems have been developed and some of them under development with an aim to minimize drug degradation or loss, to prevent harmful side effects and to improve drug bioavailability and also to favour and facilitate the accumulation in required bio- zone (site). There are number of novel carries which have been established and documented to be useful for controlled and targeted drug delivery. It is important to critically evaluate different terms used under the different broad categories of novel drug delivery system.<sup>[2]</sup>

Different drug delivery and drug targeting systems are presently being developed in order to reduce drug degradation and loss, avoid negative side effects, boost drug bioavailability, and increase the percentage of the medication accumulating in the necessary zone. Soluble polymers, microparticles, nanoparticle composed of insoluble or biodegradable natural and synthetic polymers, microcapsules, cells, cell ghosts, lipoproteins, liposomes, and micelles are a few examples of drug carriers. The carriers can be made slowly degradable, stimuli-reactive, and even targeted.<sup>[3]</sup>

**NANOPARTICLE**

“Solid colloidal particles with diameters ranging from 10 to 1000 nm and are superior to larger particles in biological applications due to their improved magnetic properties and better surface area-to-volume ratio are referred to as nanoparticle”. Different nanosystems can be used to classify nanoparticles. They are receiving a lot of interest because of their potential in medicinal applications, particularly in medication delivery, together with nanomaterials. These tiny carriers can be designed to deliver medications to particular human areas or cells. Researchers can enhance their capacity to target diseased cells while avoiding healthy ones by altering the surface properties of nanoparticles. This could improve therapy efficacy and lessen adverse effects.<sup>[4,5]</sup>

**CHITOSAN NANOPARTICLE**

Chitosan, a non-toxic and biocompatible polymer, is widely used in drug delivery systems, especially for the absorption of hydrophilic macromolecular drugs. Derived from the natural polymer chitin, chitosan is composed of  $\beta$ -[1-4]-linked D-glucosamine and N-acetyl-D-glucosamine subunits. Its free amine groups allow it to be easily cross-linked with anions, improving its mucoadhesive and biocompatible properties for enhanced drug delivery. One of its key features is its ability to increase drug permeation across mucosal epithelia. When protonated at a pH of 6.5, chitosan interacts with tight junctions between epithelial cells, temporarily opening paracellular pathways, thus

facilitating drug transport across the epithelial barrier. Chitosan nanoparticles (CSNPs), which are more effective than chitosan solution alone, are often used for controlled and improved drug delivery. The positively charged CSNPs bind to negatively charged cell membranes, enhancing drug penetration and delivery by altering membrane elasticity, though the exact mechanism of cellular entry remains debated.<sup>[6,7]</sup>

In comparison to other polymeric carriers, chitosan has a number of benefits, including biodegradability, biocompatibility, and simpler, less toxic formulation methods. Since chitosan nanoparticles dissolve in acidic aqueous solutions, they eliminate the need for hazardous organic solvents. Chitosan's cationic properties enhance crosslinking with multivalent anions. For the treatment of inflammatory bowel disease (IBD), microparticulates of several medications and prodrugs, such as salazosulfapyridine, prednisolone, and sulfasalazine have been effectively developed.<sup>[8]</sup>

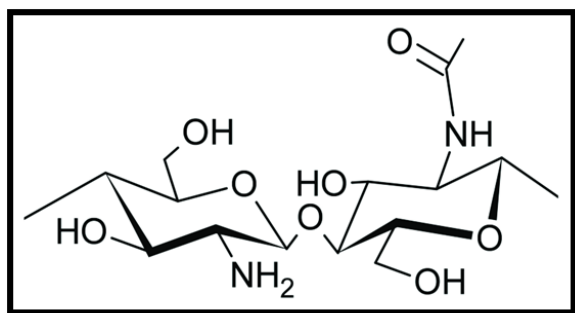


Figure 1: Structure of Chitosan.<sup>[9]</sup>

## PREPARATION METHODS OF CHITOSAN NANOPARTICLE

### Synthesis Techniques

Several techniques have been employed to prepare chitosan nanoparticles, each offering distinct advantages.

#### 1. Ionic gelation method

The process starts by dissolving positively charged chitosan in acetic acid at room temperature, followed by magnetic stirring for about an hour. In a separate solution, tripolyphosphate (TPP), a polyanion, is dissolved in deionized water (DI) to act as an ionic cross-linker.

When TPP is added to the chitosan solution, nanoparticles begin forming immediately due to electrostatic interactions between the positively charged chitosan and negatively charged TPP. The solution transitions through three stages: starting as clear (chitosan solution), becoming opalescent or milky (upon adding TPP), and then aggregating with more TPP, indicating the formation of CNPs. Mild stirring for 10 minutes stabilizes the nanoparticles.

To separate the CNPs from unreacted materials, the suspension is centrifuged. The collected nanoparticles are then thoroughly washed with water and dried for

further use. This method offers scalable and effective approach to prepare CNPs for various drug delivery applications.<sup>[10,11]</sup>

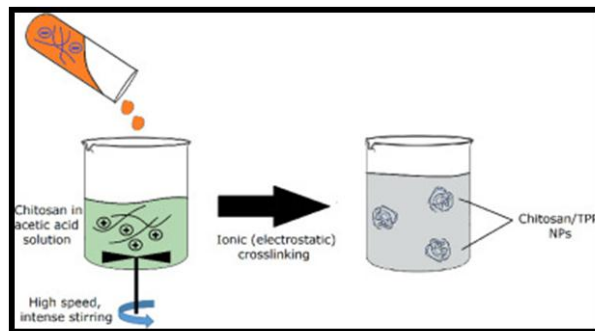


Figure 2: Ionic Gelation Method.<sup>[12]</sup>

#### 2. Emulsification Solvent Diffusion Method

In this method, an emulsion is formed by adding an organic phase to a chitosan solution containing a stabilizing agent (e.g., poloxamer). The mixture is mechanically stirred and homogenized under pressure. Upon dilution with water, polymer precipitation occurs, forming nanoparticles. The diffusion of the organic solvent into the aqueous phase is the key mechanism. This method has limitations due to high shear forces and the use of organic solvents.<sup>[13]</sup>

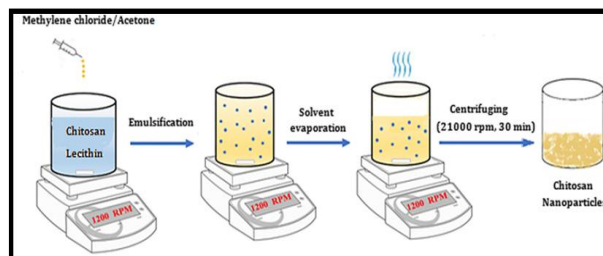


Figure 3: Emulsification Solvent Diffusion Method.<sup>[14]</sup>

#### 3. Emulsification and Cross-Linking Method

This method involves the formation of a water-in-oil (W/O) emulsion followed by the addition of a cross-linking agent, such as glutaraldehyde, which covalently bonds with the amino groups of chitosan. This process leads to the formation of hardened droplets, resulting in CNPs suitable for drug delivery applications.<sup>[15]</sup>

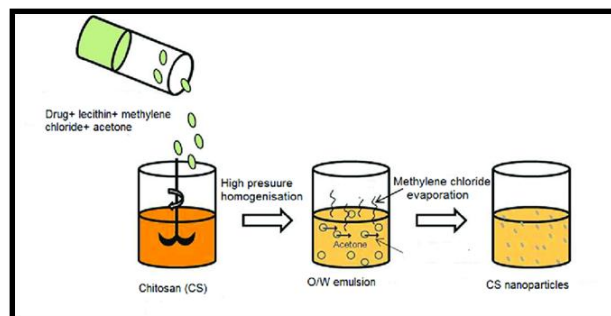


Figure 4: Emulsification and Cross-Linking Method.<sup>[16]</sup>

#### 4. Emulsion Droplet Coalescence

Chitosan and drug are dissolved together, and the mixture is added to liquid paraffin with Span 80 to form a W/O emulsion. A second emulsion is created by adding sodium hydroxide to liquid paraffin. Mixing both emulsions via high-speed homogenization causes droplet coalescence, where sodium hydroxide acts as a precipitating agent for chitosan. The particles are then washed and centrifuged.<sup>[17]</sup>

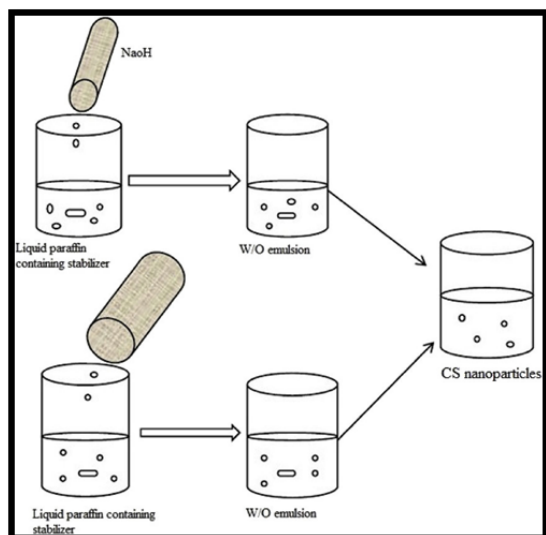


Figure 5: Emulsion Droplet Coalescence Method.<sup>[18]</sup>

#### 5. Reverse Micellization

In this method, reverse micelles form a W/O system by adding a lipophilic surfactant (e.g., cetyl triethyl ammonium bromide or sodium bis(ethylhexyl) sulfosuccinate) to an organic solvent like n-hexane. Chitosan, drug, and glutaraldehyde are dissolved in water and added to the organic phase. The nanoparticles are extracted after solvent evaporation. This method achieves uniform particle size and is suitable for hydrophobic drugs.<sup>[19]</sup>

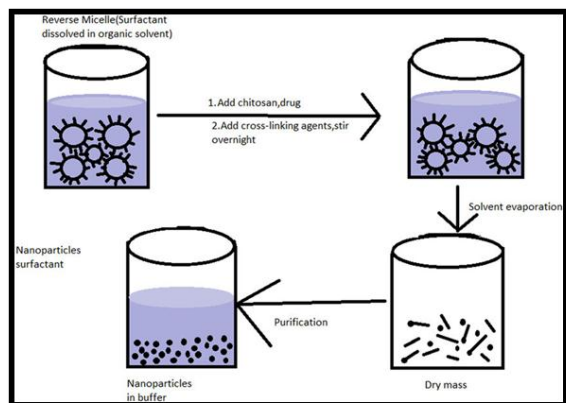


Figure 6: Reverse Micellization.<sup>[20]</sup>

#### 6. Precipitation Method

Chitosan nanoparticles are produced by blowing a chitosan solution into an alkaline solution of sodium hydroxide or methanol. The coacervate particles are

washed with hot air and cold water before filtration. The process is simple, but the resulting nanoparticles may have weak mechanical strength and irregular morphology, affected by factors like air pressure and chitosan concentration.<sup>[21]</sup>

#### 7. Microemulsion Method

In this method, reverse micelles are used to synthesize CNPs by mixing a surfactant into an organic solvent. The mixture is then added to an acidic chitosan solution. Glutaraldehyde is used as a cross-linker to react with the free amine groups of chitosan. The nanoparticles formed through this method range from 30 to 110 nm and exhibit size uniformity. It is effective for synthesizing nanoparticles with controlled size based on the degree of cross-linking.<sup>[10]</sup>

### COMPARISON OF CHITOSAN NANOPARTICLE WITH OTHER NANOPARTICLE SYSTEMS

Chitosan nanoparticles (CS-NPs) are often compared with other nanoparticle systems such as liposomes, dendrimers, and polymeric nanoparticles due to their unique properties and versatile applications. Each system has distinct advantages and limitations, making them suitable for specific purposes in drug delivery and biomedical applications.

#### 1. Liposomes vs. Chitosan Nanoparticles

Liposomes, phospholipid-based vesicles, are widely used for encapsulating both hydrophilic and hydrophobic drugs. They offer excellent biocompatibility and the ability to target specific tissues by modifying their surface with ligands. However, liposomes are prone to leakage and have limited stability in biological environments. In contrast, CS-NPs provide higher stability due to their rigid structure and better mucoadhesive properties, which enhance drug residence time at mucosal surfaces. Additionally, chitosan's intrinsic antimicrobial activity provides an advantage in applications requiring bacterial suppression.<sup>[22,23]</sup>

#### 2. Dendrimers vs. Chitosan Nanoparticles

Dendrimers are highly branched, synthetic macromolecules with a precise architecture and multiple functional groups for drug conjugation. While they offer high drug-loading capacity and excellent solubility, dendrimers can exhibit cytotoxicity due to their synthetic nature and strong cationic charges at high generations. CS-NPs, being naturally derived, are less cytotoxic and more biocompatible, making them a safer alternative for long-term therapeutic applications.<sup>[24]</sup>

#### 3. Polymeric Nanoparticles vs. Chitosan Nanoparticles

Polymeric nanoparticles, such as those made from PLGA (poly(lactic-co-glycolic acid)), provide controlled drug release and excellent biocompatibility. However, their biodegradation products, such as lactic acid, may cause localized tissue irritation. CS-NPs offer a similar controlled release profile but with added mucoadhesion

and pH responsiveness, making them particularly effective for site-specific drug delivery, such as in gastrointestinal or respiratory systems. Furthermore, CS-NPs are easier to functionalize for targeted delivery than most polymeric systems.<sup>[25,26]</sup>

#### 4. Metallic Nanoparticles vs. Chitosan Nanoparticles

Metallic nanoparticles, including gold and silver nanoparticles, have gained popularity for their plasmonic properties and applications in imaging, diagnostics, and

antimicrobial therapies. However, their long-term safety and biodegradability remain concerns. CS-NPs, being fully biodegradable, avoid accumulation in tissues and offer a safer alternative for drug delivery purposes. Moreover, combining chitosan with metallic nanoparticles can synergistically enhance their antimicrobial and imaging properties.<sup>[27,28]</sup>

Chitosan nanoparticle synthesized by different researchers along with their composition and preparation technique were depicted in table 1.

**Table 1: Chitosan nanoparticle synthesized by different researchers.**

Drug	Composition other than chitosan	Method of Preparation	Researcher	Year
Risedronate	Tripoly phosphate, acetic acid, NaOH	Ionic gelation Method	Sandhya Pathak et al <sup>[29]</sup>	2024
Lidocaine	Glyceryl oleate, Isopropyl palmitate, Propylene carbonate, Gluconic acid, Caproic acid	Micro emulsion Method	Amnon C Sintov et al <sup>[30]</sup>	2024
Metronidazole	Borax, Tannic acid	Cross linking Method	Reena Nayak et al <sup>[31]</sup>	2024
Spironolactone	Palmitic acid, Oleic acid, Tween80, Acetic acid	Ultrasonication method	Majid saeedi et al <sup>[32]</sup>	2023
Montelukast	Tripoly phosphate, Tween 80, Hyaluronic acid, Leucine	Ionic gelation method	Faqir Ullah et al <sup>[33]</sup>	2022
Carbamazepine	TPP, Acetic acid	Ionotropic gelation method	Citra Ariani Edityaningrum et al <sup>[34]</sup>	2022
Measalamine	Pectin, Methanol	Nanoprecipitation method	Tejas Pachpute <sup>[35]</sup>	2019
Vancomycin	Tripolyphosphate, Acetic acid	Ionic gelation method	T Cerchiara et al <sup>[36]</sup>	2015
5 Fluro uracil	Poly Vinyl Alcohol, Acetic acid, Mannitol	Solvent emulsification evaporation technique	Shashank Tummala et al <sup>[37]</sup>	2014
Simvastatin	Sodium Tripolyphosphate	Ionotropic gelation method	A ahmed et al <sup>[38]</sup>	2014
5 Amino Salicylic Acid	Acetic acid, TPP, HCl	Ionotropic Gelation method	Pooja Mongia et al <sup>[39]</sup>	2014
Stavudine	Acetic acid, TPP	Ionotropic Gelation method	T Vyjayanthimala <sup>[40]</sup>	2013

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