

O₂:CO₂ exchange ratios observed in a cool temperate deciduous forest ecosystem of central Japan

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ABSTRACT

Detailed observations of O₂:CO₂ exchange ratios were conducted in a cool temperate deciduous forest located in central Japan. The exchange ratios of soil respiration and net assimilation were found to be 1.11 ± 0.01 and 1.02 ± 0.03 from soil chamber and branch bag measurements, respectively. Continuous measurements of the atmospheric O₂/N₂ ratio and the CO₂ concentration, made inside the canopy during a summer season, indicated that the average exchange ratio was lower in the daytime (0.87 ± 0.02) than in the nighttime (1.03 ± 0.02) with a daily mean value of 0.94 ± 0.01 . The observed average daytime and nighttime exchange ratios were nearly consistent with the corresponding values obtained from a one-box canopy O₂/CO₂ budget model simulation of net turbulent O₂ and CO₂ fluxes between the atmosphere and the forest ecosystem. Our results suggest that the daily mean exchange ratios of the net turbulent O₂ and CO₂ fluxes depend sensitively on the forest ecosystem processes.

Keywords: O₂:CO₂ exchange ratio, forest ecosystem, atmospheric O₂/N₂ ratio, continuous measurements

1. Introduction

Estimations of oceanic and terrestrial biospheric CO₂ uptake based on the observations of the atmospheric O₂/N₂ ratio have been conducted by many research groups (e.g. Battle et al., 1996, 2000; Keeling et al., 1996; Langenfelds et al., 1999; Bender et al., 2005; Manning and Keeling, 2006; Tohjima et al., 2008; van der Laan-Luijkx et al., 2010; Ishidoya et al., 2012a, b) since the first study by Keeling and Shertz (1992). To apply this method, the global average terrestrial biospheric O₂:CO₂ molar exchange ratio ($\Delta O_2 / \Delta CO_2^{-1}$) is needed. Keeling (1988) estimated the $-O_2$:CO₂ exchange ratio (hereafter referred to as ER) of 1.05 for wood by surveying the results from various elemental abundance studies. Severinghaus (1995) obtained ER of around 1.2 from measurements of forest soils and about 1.1 for net ecosystem exchange.

The value of 1.10 ± 0.05 reported by Severinghaus (1995) has been used for the global average terrestrial biospheric ER in recent years (e.g. Bender et al., 2005; Manning and Keeling, 2006; Tohjima et al., 2008; Ishidoya et al., 2012a, b). However, to test the validity of this value, observations of the atmospheric O₂/N₂ ratio have been conducted in and over various forest canopies (e.g. Seibt et al., 2004; Kozlova et al., 2005, 2008; Sturm et al., 2005; Stephens et al., 2007). By analysing flask air samples collected at a forest site in Hainich National Park (51°N, 10°E) in central Germany for the atmospheric O₂/N₂ ratio and CO₂ concentration, Kozlova et al. (2005) obtained an average ER value of 0.99, with no significant difference between daytime and nighttime values (hereafter the ER value of forest canopy air is referred to as 'ER_{atm}'). From continuous observations of the atmospheric O₂/N₂ ratio and CO₂ concentration at the WLEF tall-tower research site (46°N, 90°W) in a forest in northern Wisconsin, USA, Stephens et al. (2007) also reported average ER_{atm} values ranging from 1.01 to 1.06, depending on the height. The ER_{atm} values obtained by

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Kozlova et al. (2005) and Stephens et al. (2007) are smaller than the value of 1.10. However, to examine the ER value of the net turbulent O_2 and CO_2 fluxes between the forest ecosystem and the atmosphere above the canopy (hereafter referred to as ' ER_F '), Seibt et al. (2004) analysed their observational results in the Griffin forest ($57^\circ N$, $4^\circ W$), UK using a one-box canopy O_2/CO_2 budget model. Seibt et al. (2004) showed that ER_F is different from ER_{atm} inside the canopy; their ER_F and ER_{atm} values estimated for the daytime are 1.26–1.38 and 1.01–1.12, respectively, the former being significantly larger than the latter. They suggested that such a difference is attributable to combined effects of the fluxes of turbulent exchange, assimilation, and plant and soil respiration, each with distinct exchange ratios, on the abundances of O_2 and CO_2 in canopy air. ER_F is an important parameter to evaluate the net exchanges of O_2 and CO_2 between the atmosphere and the terrestrial biosphere, and the value of 1.10 is usually accepted as a global average. If ER_F values of 1.26–1.38 of Seibt et al. (2004) are applicable to many forest ecosystems, it is necessary to re-examine terrestrial biospheric/oceanic CO_2 uptake and Atmospheric Potential Oxygen ($APO = O_2 + 1.1 \times CO_2$) (Stephens et al., 1998) calculated employing the 1.10 value.

To contribute to a better understanding of the terrestrial biospheric ER, we conducted soil chamber and branch bag measurements of the O_2/N_2 ratio and CO_2 concentration at Takayama deciduous broadleaf forest site in central Japan ($36^\circ 09' N$, $137^\circ 25' E$, 1420 m a.s.l.; designated as TKY in the AsiaFlux site code database). We also made continuous measurements of the atmospheric O_2/N_2 ratio and CO_2 concentration at the site during a summer season. Using

the data from these measurements, we calculated ER values of net assimilation and soil respiration and the ER_{atm} values, and estimated ER_F using a one-box canopy O_2/CO_2 budget model to compare with the obtained ER_{atm} .

2. Methods

The TKY site is situated about 15 km east of a provincial city, Takayama, as shown in Fig. 1. Major tree species around the site are deciduous broad-leaved trees such as birch and oak, with a canopy height of about 15–20 m, and the ground is covered with bamboo grass. Budding and leaf shedding occur in May and October, respectively, and the ground is usually covered with snow from December to April. The forest has been protected from deforestation for more than 50 yr. The annual mean temperature and precipitation are about $6.4^\circ C$ and 2100 mm, respectively. The rainy season takes place in the early summer when the site is strongly affected by the Asian monsoon. Possible influence of nearby anthropogenic sources on the atmospheric CO_2 concentration at the site is estimated to be relatively small from previous numerical model simulations (Kondo et al., 2001). More detailed descriptions of the TKY site have already been given in our previous papers (e.g. Murayama et al., 2003, 2010). Continuous measurements of the net CO_2 flux between the forest and the atmosphere, as well as of meteorological parameters, using a 27-m tall tower were first initiated in 1993 by employing an aero-dynamic method that was replaced in 1998 by an eddy covariance method (Yamamoto et al., 1999; Saigusa et al., 2005). The CO_2 flux data taken at TKY are available from the AsiaFlux database (<http://www.asiaflux.net/>).

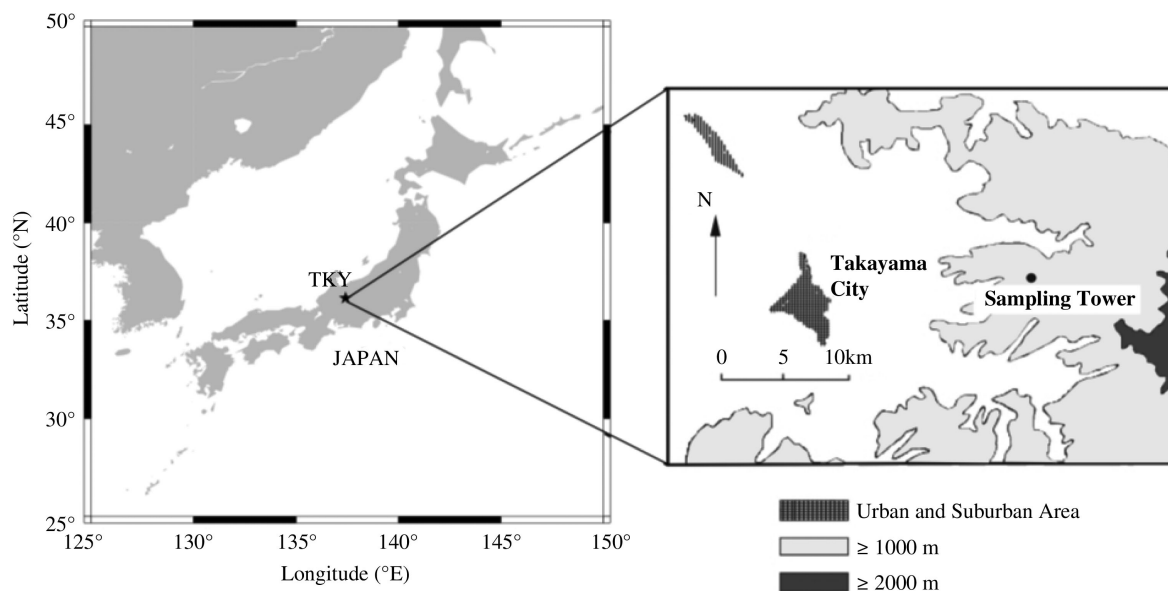


Fig. 1. Location of Takayama site ($36^\circ 09' N$, $137^\circ 25' E$, 1420 m a.s.l., TKY) in central Japan.

To measure the O₂/N₂ ratio and CO₂ concentration of soil respiration at the site, air samples were collected at two different places using a stainless-steel closed chamber with a volume of 100 l; one was placed on a small ridge and the other one in a small valley, separated by a distance of 60 m. Each chamber has a cover at its upper part, and it is connected to ambient atmosphere through a 1/16-inch O.D. stainless-steel tube to minimise the pressure imbalance between the soil air and the chamber air. About five air samples were taken from the chamber over a period of 30–60 minutes after closing the cover, using 250-ml Pyrex glass flasks with Viton O-ring seal stopcocks at both ends. During air sampling, we prevented exposing the chamber to direct sunlight using a parasol. As the volume of the chamber is sufficiently larger than the sampled air, it was assumed that an intrusion of ambient atmosphere into the chamber during the collection of air samples from inside the chamber had a negligible impact on the chamber air. Air samples were also collected to measure O₂/N₂ and CO₂ of the net plant assimilation. The air sampling was conducted during the daytime by using a branch bag method on the leaves of Mongolian oak that is dominant around the site. Its branches were first inserted into a 30 l bag with two outlets, and a 100 l bag filled with ambient air was connected to one outlet of the 30 l bag. Both of the bags were made of transparent polyvinylidene difluoride. Each set of five air samples were collected over a period of 30 minutes, by introducing the air from the 30 l bag at a flow rate of 3 l min⁻¹ using a diaphragm pump into 150-ml Pyrex glass flasks with Viton O-ring seal stopcocks at both ends. Water vapour contained in the sample air was removed using Mg(ClO₄)₂. The soil chamber measurements were carried out about once per month from August 2004 to October 2006, and from July 2011 to October 2012, while the branch bag measurements were performed eight times during the summer of 2011.

The collected flask samples were brought back to our laboratory and were analysed for the O₂/N₂ ratio and the CO₂ concentration. The O₂/N₂ ratio is usually reported in per meg unit:

$$\delta(\text{O}_2/\text{N}_2) = \left[\frac{(\text{O}_2/\text{N}_2)_{\text{sample}}}{(\text{O}_2/\text{N}_2)_{\text{standard}}} - 1 \right] \times 10^6, \quad (1)$$

where subscripts ‘sample’ and ‘standard’ indicate the sample air and the standard gas, respectively. Because O₂ is 20.946% of air by volume, 4.8 per meg (=1/0.20946) corresponds to a change of 1 ppm. In this study, the ratio of 4.8 per meg ppm⁻¹ was used to calculate the ER values using the observed $\delta(\text{O}_2/\text{N}_2)$. The $\delta(\text{O}_2/\text{N}_2)$ value of each air sample was determined against our working standard gas using a mass spectrometer (Finnigan MAT-252 or Thermo Scientific Delta-V). The measurement precision

was estimated to be ± 5.4 per meg ($\pm 1\sigma$) (Ishidoya et al., 2003). The CO₂ concentration was determined using the Delta-V for the leaf air samples with a precision of ± 0.3 ppm (Ishidoya and Murayama, 2013), and using a gas chromatograph (Shimadzu GC-9A) equipped with a flame ionization detector and a methanizer for soil respiration air samples with a precision of ± 0.2 ppm, against our air-based CO₂ standard gas system maintained at Tohoku University (Tanaka et al., 1983; Nakazawa et al., 1991). By analysing air samples with a wide range of (O₂/N₂) ratios of 0.1–0.3 using the MAT-252, we have confirmed that $\delta(\text{O}_2/\text{N}_2)$ is linearly related to the (O₂/N₂) ratio, with uncertainties of less than $\pm 2\%$ (Ishidoya et al., 2003). From the estimated uncertainties, it is expected that the ER value can be determined with a precision of ± 0.004 even for the soil respiration air with widely varying $\delta(\text{O}_2/\text{N}_2)$. Regarding the CO₂ concentration, the non-linear output effects of the gas chromatograph and the Delta-V are negligibly small, since our standard gases fully cover the values measured in this study.

We also carried out continuous measurements of the atmospheric O₂/N₂ ratio at the TKY site using a fuel cell analyser (Sable Systems International, Oxzilla II) during the period 4 August–4 September 2012. The continuous O₂ measurement system used is similar to that developed by Goto et al. (2013). Sample air was taken from 8.8 m (inside the canopy) and 27 m (above the canopy) using diaphragm pumps, and the air intake set at each height was equipped with an aspirator to avoid different thermal diffusions of O₂ and N₂ due to radiative heating/cooling (Blaine et al., 2006). The sample air taken from each height was introduced into the O₂ analyser at a flow rate of 80 ml min⁻¹ and measured for 36 minutes to obtain eight data values of $\delta(\text{O}_2/\text{N}_2)$, followed by measurements of the standard gases for 14 minutes. After this, the sample air from the other height was analysed by the same procedure. The sample air pressure was stabilised to an order of 10^{-3} Pa using a flow regulation valve (HORIBA STEC, PV-1000) and a precise differential pressure sensor (Setra, 239). The temperature of the fuel cell was stabilised to $32 \pm 0.1^\circ\text{C}$ using a Peltier control system. Our O₂ measurement system is also equipped with a non-dispersive infrared analyser (LiCOR, LI-6262) to simultaneously measure the CO₂ concentration of sample air with a precision of ± 0.05 ppm.

The removal of water vapour from the sample air and the preparation of the standard gases adopted for the present O₂ measurement system were different from those described in Goto et al. (2013). Our sample dryer had two air flow paths, and each path was equipped with a water trap set in a Stirling cycle refrigerator (Twinbird, SC-UE15R). The flow paths were alternately switched, so that one trap was cooled at -85°C to remove water vapour from the sample air and the other trap was kept at room

temperatures to discharge the melting ice by flowing ambient air dried by a heatless air dryer (CKD, HD-0.5) through it. By employing the sample dryer, the maintenance of the water trap was automated.

The standard gases were prepared by adding appropriate amounts of pure O₂ or N₂ to CO₂ standard gases, which were mixtures of industrially purified air and CO₂. The amount of O₂ or N₂ to be added to each standard gas was calculated based on the $\delta(\text{O}_2/\text{N}_2)$ of the ingredient air that was measured using the Delta-V. The output of the fuel cell O₂ analyser depends on the amount of O₂ available to diffuse across the membrane of the cell, and the O₂ amount is closely related to the total pressure and O₂ mole fraction of air in the cell. Therefore, to calculate $\delta(\text{O}_2/\text{N}_2)$ from the analyser output, it is necessary to evaluate the dilution effects caused by changes in gases other than O₂ (Keeling et al., 1998; Manning et al., 1999). In this study, we took account of the dilution effects of CO₂ and Ar for both the sample air and standard gas. The dilution effects of CO₂ contained in the sample air and standard gases were calculated using their measured CO₂ concentrations. However, we did not measure the Ar/N₂ ratio (hereafter defined by $\delta(\text{Ar}/\text{N}_2)$ in the same way as $\delta(\text{O}_2/\text{N}_2)$ for the O₂/N₂ ratio) of sample air at TKY. However, Keeling et al. (2004) and Cassar et al. (2008) reported that changes in the atmospheric $\delta(\text{Ar}/\text{N}_2)$ are within several tens per meg on a global scale. Considering their observational results, we did not correct our measured $\delta(\text{O}_2/\text{N}_2)$ of the sample air for the dilution effect by changes in atmospheric Ar. The dilution effect of Ar on $\delta(\text{O}_2/\text{N}_2)$ of the sample air, arising from the standard gases with different $\delta(\text{Ar}/\text{N}_2)$ values, was corrected by the following method. We first determined $\delta(\text{Ar}/\text{N}_2)$ of the ingredient air of our standard gases using the Delta-V, and then compared the results with continuously measured values of natural air near the surface at Tsukuba (36°N, 140°E), Japan using the same mass spectrometer (Ishidoya and Murayama, 2013). From this comparison, we found differences of sever per mil between our standards and the ambient air. Therefore, to calculate $\delta(\text{O}_2/\text{N}_2)$ of the sample air from the analyser output, we used the formula;

$$\delta(\text{O}_2/\text{N}_2) = aV + \frac{\delta X_{\text{CO}_2}}{(1 - X_{\text{O}_2})} + \frac{\delta X_{\text{Ar}}}{(1 - X_{\text{O}_2})} \text{ (per meg)}, \quad (2)$$

where V is the voltage output from the analyser, which represents the difference in O₂ mole fraction between the sample air (or standard gas) and an arbitrary reference air, a is the span factor in units of per meg V⁻¹, calculated by analysing high- and low-span standard gases with known O₂ mole fractions, δX_{CO_2} (δX_{Ar}) is the difference in CO₂ (Ar) mole fraction of the sample air from the reference air, in ppm, and X_{O_2} is the standard mole fraction of O₂ in dry

air to be described below. δX_{CO_2} and δX_{Ar} account for the respective dilution effects of changes in CO₂ and Ar on the O₂ mole fraction in the fuel cell. Equation (2) is the same as that used in Manning et al. (1999) except for inclusion of δX_{Ar} . By assuming the standard mole fractions of N₂, O₂ and Ar in dry air to be 0.78084, 0.20946 and 0.00934, respectively (Nicolet, 1960; Machta and Hughes, 1970), the change of 1 per mil in $\delta(\text{Ar}/\text{N}_2)$ corresponds to the change of about 9 ppm in δX_{Ar} , leading to the change of 12 per meg in $\delta(\text{O}_2/\text{N}_2)$. Therefore, the correction of $\delta(\text{O}_2/\text{N}_2)$ of the sample air due to our standard gases amounted to several tens per meg.

The precision of our continuous measurements of atmospheric $\delta(\text{O}_2/\text{N}_2)$ was estimated to be about ± 7 per meg, which is worse than that in Goto et al. (2013) (± 1.4 – 1.9 per meg). The cause may be attributable to unstable room temperature inside the observation hut at TKY.

3. Results

3.1. Exchange ratios of O₂:CO₂ obtained from soil chamber and branch bag measurements

In this study, soil chamber and branch bag measurements were made 42 and eight times, respectively. The $\delta(\text{O}_2/\text{N}_2)$ and CO₂ concentration data obtained from the respective measurements are shown in Fig. 2a and b. As mentioned above, five air samples were usually collected in sequence during each sampling. In these figures, the $\delta(\text{O}_2/\text{N}_2)$ and CO₂ concentration values are expressed as deviations from the corresponding values obtained for the first air sample in each sampling group. By applying a linear regression analysis to the $\delta(\text{O}_2/\text{N}_2)$ and CO₂ concentration values, the ER values of soil respiration and net assimilation were estimated to be 1.11 ± 0.01 and 1.02 ± 0.03 ($\pm 1\sigma$), respectively. These values are significantly different at 99% confidence level. We did not observe any statistically significant difference between the ER values of soil respiration obtained at the two different places. As soil respiration amounts to about 90% of the ecosystem respiration (RE) at TKY (Saigusa et al., 2005), we have assumed that the ER value for RE (hereafter referred to as 'ER_R') is equal to the soil respiration ER. The ER_R value of 1.11 ± 0.01 obtained in this study falls in the range of 1.06–1.22 found by Severinghaus (1995) using a flow-through chamber under laboratory conditions. The value is, however, larger than 0.94 ± 0.04 obtained from the soil chamber measurements in the Griffin forest by Seibt et al. (2004).

The photosynthesis ER, that is, for gross primary production (GPP) (hereafter referred to as 'ER_A'), is expected to be 1.00 from photochemical reaction and the Calvin–Benson–Bassham cycle ($6\text{CO}_2 + 12\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{H}_2\text{O} + 6\text{O}_2$). It is consistent with the ER value of 1.02 ± 0.03

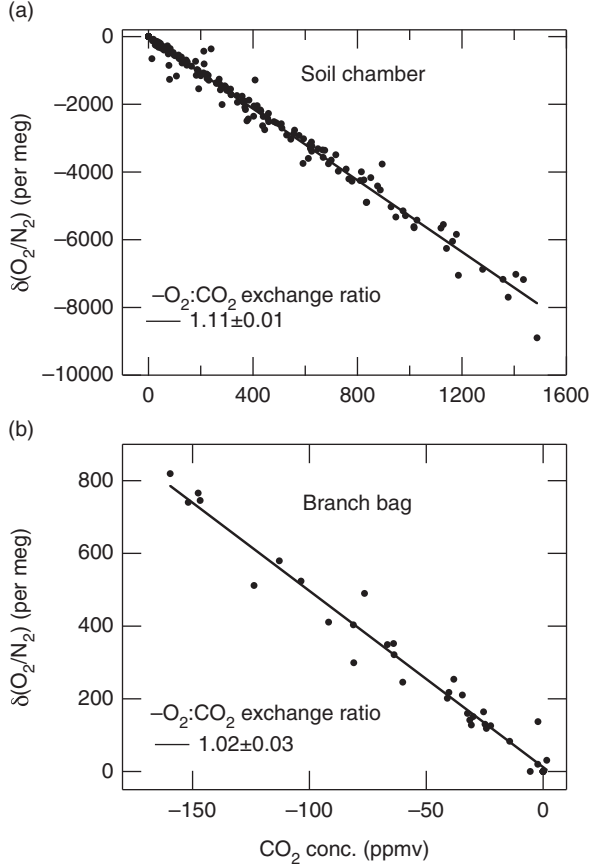


Fig. 2. Relationships between $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration obtained from (a) soil chamber and (b) branch bag measurements. Solid lines denote the regression lines fitted to the data.

obtained at TKY for net assimilation. While Seibt et al. (2004) reported ER values of 1.19 ± 0.12 for the Griffin forest site and 1.08 ± 0.16 for the Hainich National Park site based on branch bag measurements, their uncertainties are substantially larger than ours.

3.2. Continuous measurements of the atmospheric $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration

Figure 3 shows $\delta(\text{O}_2/\text{N}_2)$ and the CO_2 concentration observed continuously at the 8.8 and 27 m tower heights during 4 August–4 September 2012. Spline smoothing was applied to the data and the 24-hour running means are also shown. As seen in the figure, the $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration values vary diurnally almost in opposite phase with each other. The daily maximum (minimum) of $\delta(\text{O}_2/\text{N}_2)$ (CO_2 concentration) occurs during the daytime, due to O_2 production (CO_2 consumption) caused by GPP that is larger than RE (Murayama et al., 2003; Saigusa et al., 2005). Small diurnal amplitudes of $\delta(\text{O}_2/\text{N}_2)$ and CO_2

were observed on the days with small photosynthetically active radiation (PAR). This also suggests that GPP is a main contributor to the diurnal cycles. It is also seen from Fig. 3 that the $\delta(\text{O}_2/\text{N}_2)$ (CO_2 concentration) values observed at 8.8 m are generally lower (higher) than the values at 27 m in the nighttime, due to the loss of O_2 (accumulation of CO_2) by soil respiration in the stable atmosphere near the ground. However, the daytime $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration values observed at 8.8 m are close to the values at 27 m, probably due to strong convective mixing.

The 24-hour running means of the $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration values show day-to-day variations, their phases being opposite. It is also seen from Fig. 3 that the day-to-day variations in the CO_2 concentration are generally in phase with those of air temperature at 25 m, with a delay of about 1 d. If the phase of the CO_2 variations is delayed by 1 d, then the correlation coefficient of the day-to-day variations between the air temperature and the CO_2 concentration is calculated to be 0.7. This suggests that the day-to-day variations in $\delta(\text{O}_2/\text{N}_2)$ and CO_2 are closely related with changes in RE, since RE at TKY is mostly a function of air temperature (Saigusa et al., 2005). To confirm our suggestion, we examined the relationship between the atmospheric $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration. From plots of the daily mean values of these variables shown in Fig. 4, an ER_{atm} value of 1.10 ± 0.05 ($\pm 1\sigma$) is obtained for both heights of 8.8 and 27 m. This value of ER_{atm} agrees well with the ER value derived from our soil chamber measurements (1.11 ± 0.01). As described above, ER_R can be approximated by the soil respiration ER at TKY. Therefore, the inter-diurnal variation in the daily mean $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration at TKY could be caused mainly by the variation in RE, at least for the observation period under discussion.

4. Discussion and conclusions

By treating the forest canopy air as a well-mixed reservoir unaffected by air–sea exchange and fossil fuel combustion, the respective budgets of CO_2 and O_2 in a forest ecosystem can be represented by

$$M \frac{d(\text{CO}_2)}{dt} = -A + R - F_c, \quad (3)$$

and

$$M \frac{d(\text{O}_2)}{dt} = A \cdot \text{ER}_A - R \cdot \text{ER}_R + F_c \cdot \text{ER}_F. \quad (4)$$

Here, A ($\mu\text{mol m}^{-2} \text{s}^{-1}$), R ($\mu\text{mol m}^{-2} \text{s}^{-1}$) and F_c ($\mu\text{mol m}^{-2} \text{s}^{-1}$) represent GPP, RE and net turbulent CO_2 flux from the forest to the overlying atmosphere, respectively. M (mol m^{-2}) represents the number of moles per

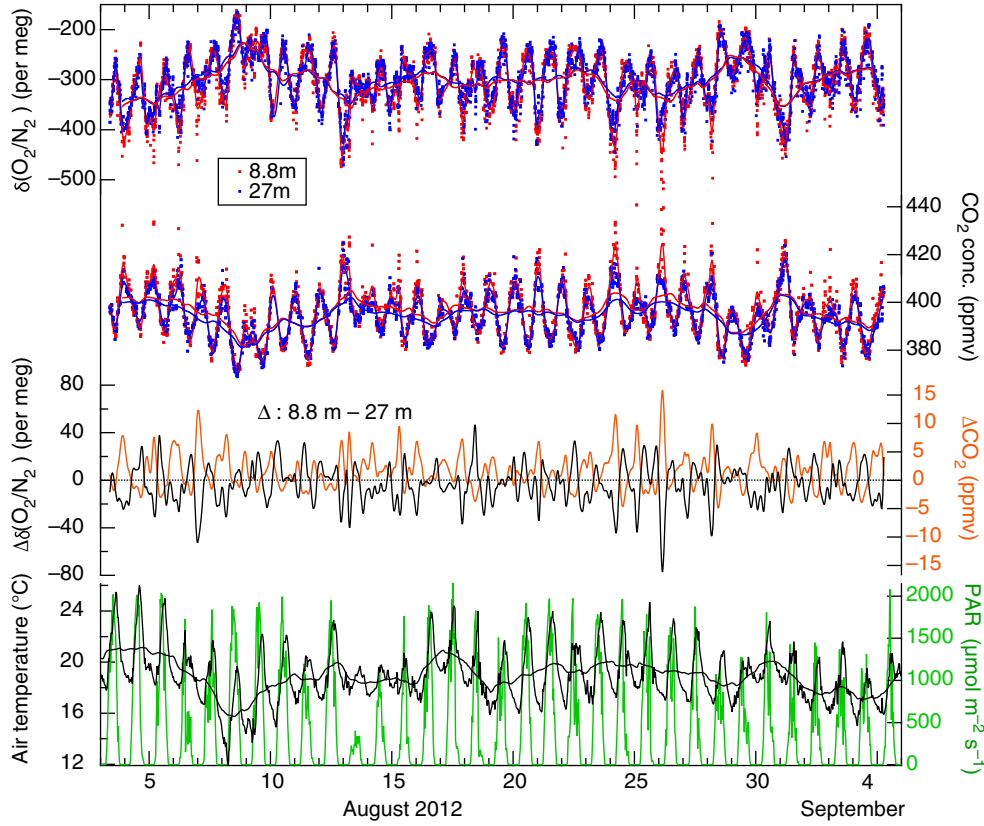


Fig. 3. $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration observed at heights of 8.8 (red dots) and 27 m (blue dots) for the period 4 August–4 September 2012. Smoothing spline curves fitted to the observed data and their 24-hour running mean values are also shown. $\Delta\delta(\text{O}_2/\text{N}_2)$ (ΔCO_2), representing the difference of the smoothing spline curve of the $\delta(\text{O}_2/\text{N}_2)$ (CO_2 concentration) at 8.8 m from that at 27 m, is also plotted by black (orange) line, together with air temperature (black line) and PAR (green line) at 25 and 19.5 m heights, and 24-hour running mean values of air temperature are also plotted.

unit area contained in a column of air extending from the forest floor to the top of the canopy, and CO_2 ($\mu\text{mol mol}^{-1}$, or ppm) and O_2 ($\mu\text{mol mol}^{-1}$, or ppm) are the respective concentrations of CO_2 and O_2 of air in the same column. ER_A and ER_R have the same meanings as defined above, and ER_F denotes the exchange ratio of $-\text{O}_2:\text{CO}_2$ for the net turbulent CO_2 flux. In this study, the data obtained from the eddy covariance flux measurements at TKY were used for F_c , and RE was calculated as a function of air temperature inside the canopy (Saigusa et al., 2005). GPP was obtained as the sum of net ecosystem production (NEP) and RE ($\text{GPP} = \text{NEP} + \text{RE}$), in which NEP was calculated from F_c and $Md(\text{CO}_2)dt^{-1}$ ($\text{NEP} = -(F_c + Md(\text{CO}_2)dt^{-1})$). The canopy height was assumed to be 25 m, and ER_R was taken to be 1.11 in accordance with the result shown in the preceding section. With respect to ER_A , we preferred the value of 1.00 to 1.02 obtained from the branch bag measurements for net assimilation, since we regarded ER_A as ER for photosynthesis assuming that leaf respiration, included in net assimilation, is a part of RE. The budgets of O_2 and CO_2 in a forest ecosystem,

as expressed by eqs. (3) and (4), are shown schematically as a vector diagram in Fig. 5. The data in Fig. 8 (discussed below) show that the net changes in O_2 and CO_2 ($Md(\text{O}_2)dt^{-1}$ and $Md(\text{CO}_2)dt^{-1}$) are generally very small (in magnitude) compared to changes in O_2 and CO_2 associated with NEP. Thus, the vector diagram makes it clear that if NEP is positive (negative), ER_F must be smaller than 1.00 (larger than 1.11).

To examine the relationship between ER_{atm} and ER_F on a diurnal time scale, we first extracted intradiurnal components of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration from their measured values, to calculate ER_{atm} . The calculation involved subtracting 24-hour running mean values of each variable from the corresponding measured values. The diurnal cycles of the atmospheric $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration on 23 August 2012, obtained by detrending the data, are plotted in Fig. 6a, as a typical example of their diurnal cycles. Best-fit curves to the data, represented by the fundamental and its first harmonics (periods of 24 and 12 hours) terms, are also shown in the figure. As seen from Fig. 6a, $\delta(\text{O}_2/\text{N}_2)$ and CO_2 vary diurnally almost in

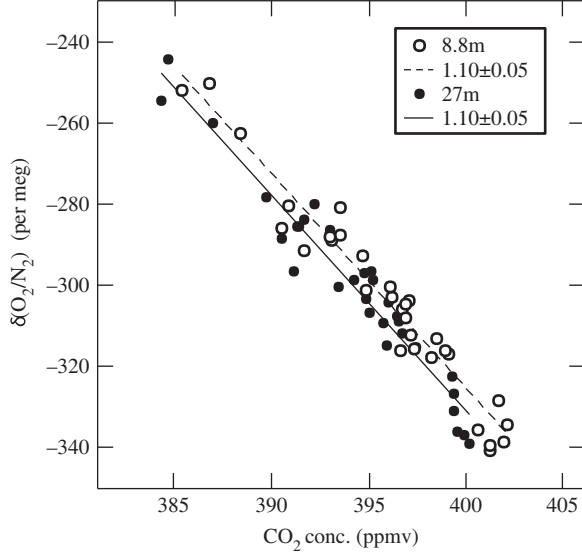


Fig. 4. Relationships between the daily mean values of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration observed at 8.8 (open circles) and 27 m (closed circles).

opposite phase. The relationships between the detrended values of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 are shown in Fig. 6b. By applying a linear regression to all the data over the whole day, ER_{atm} was found to be 0.98 ± 0.04 ($\pm 1\sigma$) for 8.8 m and 1.06 ± 0.05 ($\pm 1\sigma$) for 27 m. These values are in a range of the ER_{atm} values reported in previous studies (Seibt et al., 2004; Kozlova et al., 2005; Stephens et al., 2007). The ER_{atm} values calculated for the time intervals of 6:00–18:00 and 18:00–6:00 at 8.8 (27) m are 1.01 ± 0.05 (1.12 ± 0.07) and 0.90 ± 0.10 (1.04 ± 0.11), respectively. Considering the uncertainties, the daytime and nighttime ER_{atm} values are not significantly different from each other at both heights.

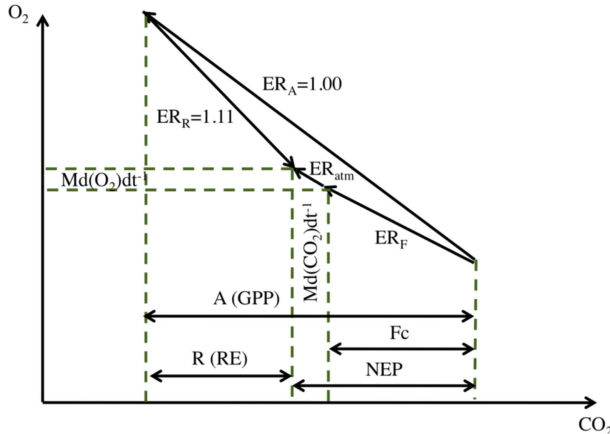


Fig. 5. Schematic vector diagram representing the contributions to O_2 and CO_2 budgets in a forest ecosystem.

After detrending, the data for all the days in this study were combined in a ‘climatology’ of the diurnal cycle. The cycles for both O_2 and CO_2 are shown in Fig. 7a along with the two-harmonic fits. The average diurnal cycles of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 at 8.8 (27) m height show peak-to-peak amplitudes of 111 ± 6 (94 ± 6) per meg and 24.2 ± 1.2 (20.0 ± 1.0) ppm, respectively. By approximating the relationship between the $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration shown in Fig. 7a by a linear line, we found the average value of ER_{atm} over the observation period to be 0.94 ± 0.01 at 8.8 m and 0.96 ± 0.01 at 27 m. Figure 7b shows the relationships between the best-fit curves of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration values shown in Fig. 7a. It is clearly seen in Fig. 7b that the ER_{atm} value obtained from the average diurnal cycles of both variables is smaller in the daytime (6:00–13:00) than in the nighttime (18:00–24:00). It was also found that the daytime ER_{atm} values at both heights are clearly lower than ER_A (1.00). The ER_{atm} values at 8.8 m (27 m) for the time intervals of 6:00–13:00 and 18:00–24:00 were calculated to be 0.87 ± 0.02 and 1.03 ± 0.02 (0.86 ± 0.02 and 1.05 ± 0.02), respectively, by applying a linear regression analysis to the $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration data shown in Fig. 7a.

As mentioned above, ER_F is used to estimate regional (and global) CO_2 fluxes from changes in O_2 and CO_2 in the well-mixed troposphere, and it is not directly measurable at present. We calculated the average diurnal variation of ER_F by performing a one-box budget model analysis, expressed by eqs. (3) and (4) and illustrated in Fig. 5, using values of GPP, RE, $Md(\text{O}_2)dt^{-1}$ and $Md(\text{CO}_2)dt^{-1}$ obtained for the observation period; we then compared ER_F and ER_{atm} . To obtain average diurnal variations of GPP and RE, the same procedure as employed above for $\delta(\text{O}_2/\text{N}_2)$ and CO_2 was applied to these variables. The values of $Md(\text{O}_2)dt^{-1}$ and $Md(\text{CO}_2)dt^{-1}$ were derived respectively from the average diurnal cycles of the atmospheric $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration observed at 8.8 m. The average diurnal variations of ER_F , GPP, RE, NEP, $Md(\text{O}_2)dt^{-1}$ and $Md(\text{CO}_2)dt^{-1}$ are shown in Fig. 8. Also shown in this figure is the ER_F for the daily-integrated net turbulent flux (hereafter referred to as ‘daily mean ER_F ’), along with the ER_{atm} values at 8.8 m shown in Fig. 7b for the time intervals of 6:00–13:00 and 18:00–24:00 averaged over the observational period. The daily mean ER_F is obtained by dividing the daily-integrated values of the net turbulent O_2 flux [$F_o = F_c \times \text{ER}_F$, calculated from eq. (4)] by that of net turbulent CO_2 flux [F_c from eq. (3)] as follows:

$$\text{daily mean ER}_F = \frac{\int (F_o) dt}{\int (F_c) dt} = \frac{\int (-A \cdot \text{ER}_A + R \cdot \text{ER}_R + M \cdot d(\text{O}_2)/dt) dt}{\int (-A + R - M \cdot d(\text{CO}_2)/dt) dt}. \quad (5)$$

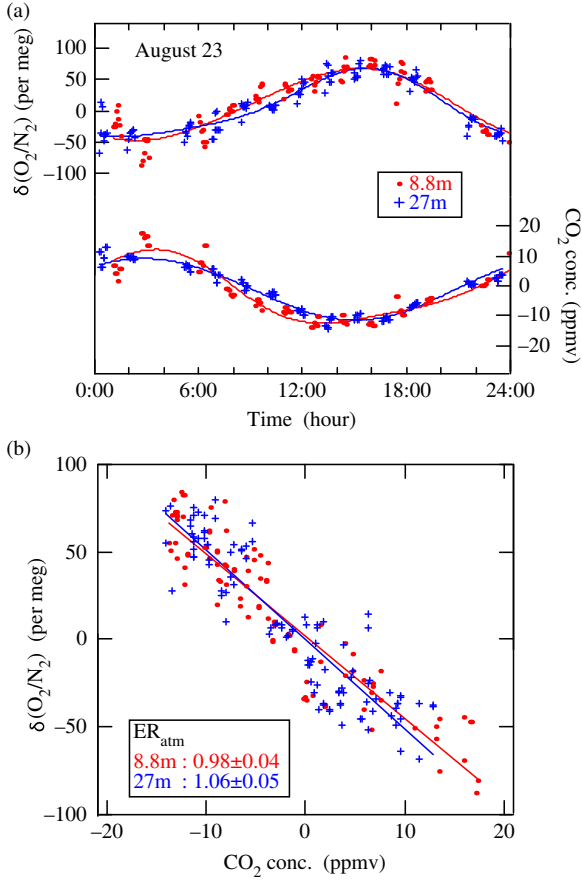


Fig. 6. (a) Diurnal cycles of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration observed at 8.8 and 27 m in the forest canopy on 23 August 2012 (circles), together with their best-fit curves (solid lines). (b) Relationships between $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration shown in (a). Solid lines denote the regression lines fitted to the data.

In eq. (5), the time integration range is from 0:00 to 24:00. Therefore, the daily mean ER_F is flux-weighted value and not consistent with a simple daily average of the temporally different ER_F values. The time periods of 6:00–13:00 and 18:00–24:00 correspond to the times when the atmospheric $\delta(\text{O}_2/\text{N}_2)$ monotonically increases and decreases (see Fig. 7), respectively, as well as when the absolute magnitude of A-R is greater than F_c (see Fig. 5). The ER_F value is found to be lower in the daytime than in the nighttime, similar to ER_{atm} , and the average daily mean ER_F value of 0.89 agrees with an average ER_{atm} value for 6:00–13:00 (0.87 ± 0.02). In contrast to our results, Seibt et al. (2004) found that there are large differences between ER_{atm} (~ 1.0) and ER_F (1.26 to 1.38) at the Griffin forest site. The difference between the ER_F values reported by Seibt et al. (2004) and our study is mainly ascribed to the fact that we employed 1.00 for ER_A and 1.11 for ER_R in our one-box model analysis, while

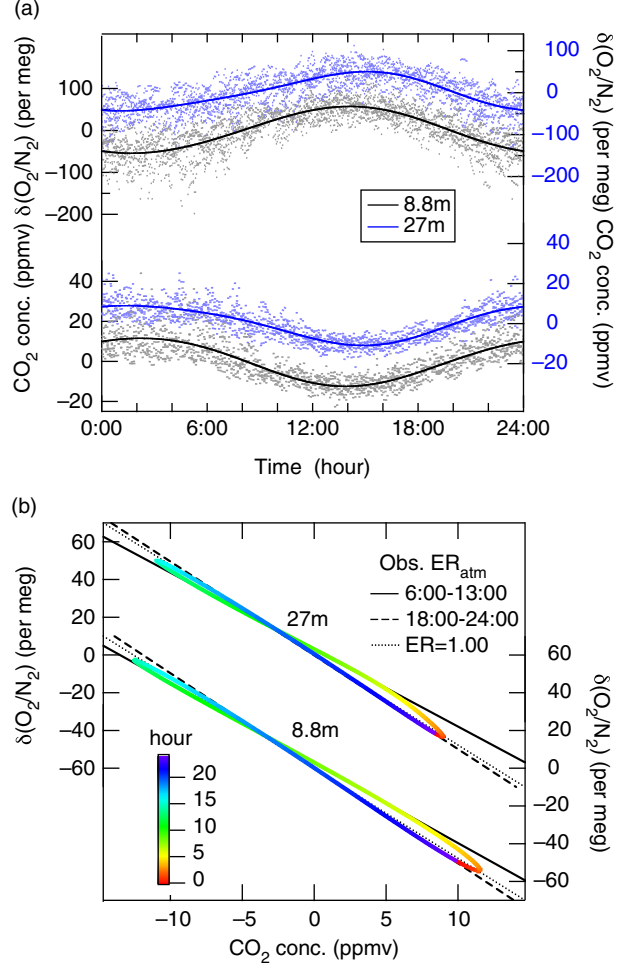


Fig. 7. (a) Plots of diurnal cycle component values of $\delta(\text{O}_2/\text{N}_2)$ and CO_2 concentration observed at 8.8 and 27 m heights for the period 4 August–4 September 2012, and their best-fit curves, and (b) relationships between the best-fit curves of the two variables. In (b), the colour scale denotes the time of the day, and black solid, dashed and dotted lines indicate the relationships derived from the data for the periods 6:00–13:00 and 18:00–24:00 and expected from the exchange ratio of 1.00, respectively.

Seibt et al. (2004) used the corresponding values of 1.19 and 0.94, based on their branch bag and soil chamber measurements.

In this study, we closely examined the conversion between O_2 and CO_2 in a Japanese cool temperate deciduous forest ecosystem. The ER value for soil respiration was found to be 1.11 ± 0.01 , which is larger than that for net assimilation (1.02 ± 0.03). The average ER_{atm} value, as well as the average ER_F calculated using a one-box canopy O_2/CO_2 budget model on the assumption that ER is 1.00 for GPP (ER_A) and 1.11 for RE (ER_R), was clearly lower in the daytime than in the nighttime. The averaged ER_{atm} over the time interval of 6:00–13:00 was also found

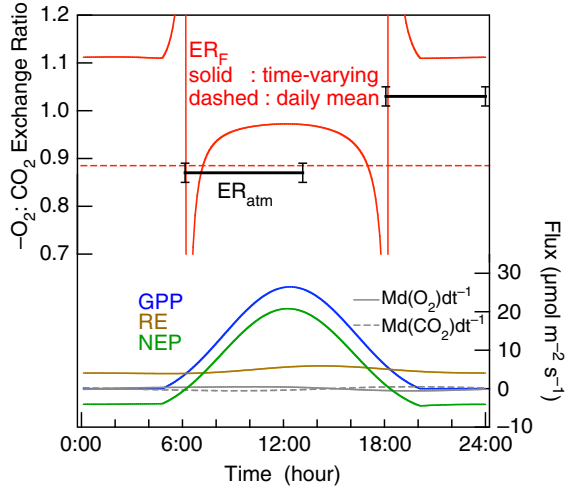


Fig. 8. Calculated average ER_F values (red lines) and average observed ER_{atm} values (black solid lines) for the time intervals of 6:00–13:00 and 18:00–24:00 at 8.8 m height for the period 4 August–4 September 2012. Average diurnal variations of GPP, RE, NEP (blue, brown, green solid lines), $Md(O_2)dt^{-1}$ and $Md(CO_2)dt^{-1}$ (grey solid and dashed lines) are also shown in lower part of the figure. The time-varying ER_F values (red solid line) are calculated using eqs. (3) and (4) and the average diurnal variations of GPP, RE, $Md(O_2)dt^{-1}$ and $Md(CO_2)dt^{-1}$. The daily mean ER_F value (red dashed line) is the ratio of the daily-integrated net turbulent O_2 and CO_2 fluxes calculated from eqs. (3) and (4) (see text).

to be in agreement with the averaged daily mean ER_F . Our results also suggest that APO (Stephens et al., 1998), which has been often used to estimate the global CO_2 budget and the air-sea O_2 flux (e.g. Manning and Keeling, 2006; Ishidoya et al., 2012b), should be re-examined. APO is defined by assuming that ER for net O_2 and CO_2 fluxes caused by terrestrial biospheric activities (ER_F in this study) is 1.1 on average. However, our results indicate that the daily mean ER_F could be significantly different depending on ER_R and daily-integrated values of GPP and RE, since ER_A is a constant value of 1.00 and the daily-integrated $Md(O_2)dt^{-1}$ and $Md(CO_2)dt^{-1}$ are zero. If we assume 0.9 for the global average ER_F to calculate APO, instead of the widely accepted value of 1.1, then the terrestrial biospheric CO_2 uptake of 1.0 ± 0.8 GtC yr⁻¹ and the oceanic CO_2 uptake of 2.5 ± 0.7 GtC yr⁻¹ reported by Ishidoya et al. (2012a) for the period 2000–2010 are increased and decreased by 0.22 GtC yr⁻¹, respectively. This change in ER_F reduces the discrepancy between the oceanic CO_2 uptake estimated from the APO method (Ishidoya et al., 2012a, b) and ocean models (Sarmiento et al., 2010). Therefore, it is important not only to perform further observations with high precision to determine ER_{atm} but also to directly estimate ER_F by conducting

simultaneous measurements of CO_2 and O_2 turbulent fluxes in various forests.

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References

- Battle, M., Bender, M., Sowers, T., Tans, P. P., Butler, J. H. and co-authors. 1996. Atmospheric gas concentrations over the past century measured in air from firn at the South Pole. *Nature*. **383**(6597), 231–235.
- Battle, M., Bender, M. L., Tans, P. P., White, J. W. C., Ellis, J. T. and co-authors. 2000. Global carbon sinks and their variability inferred from atmospheric O_2 and $\delta^{13}C$. *Science*. **287**, 2467–2470.
- Bender, M. L., Ho, D. T., Hendricks, M. B., Mika, R., Battle, M. O. and co-authors. 2005. Atmospheric O_2/N_2 changes, 1993–2002: implications for the partitioning of fossil fuel CO_2 sequestration. *Glob. Biogeochem. Cycles*. **19**, GB4017. DOI: 10.1029/2004GB002410.
- Blaine, T. W., Keeling, R. F. and Paplawsky, W. J. 2006. An improved inlet for precisely measuring the atmospheric Ar/N_2 ratio. *Atmos. Chem. Phys.* **6**, 1181–1184.
- Cassar, N., McKinley, G. A., Bender, M. L., Mika, R. and Battle, M. 2008. An improved comparison of atmospheric Ar/N_2 time series and paired ocean–atmosphere model predictions. *J. Geophys. Res.* **113**, D21122. DOI: 10.1029/2008JD009817.
- Goto, D., Morimoto, S., Ishidoya, S., Ogi, A., Aoki, S. and co-authors. 2013. Development of a high precision continuous measurement system for the atmospheric O_2/N_2 ratio and its application at Aobayama, Sendai, Japan. *J. Meteorol. Soc. Jpn.* **91**, 179–192.
- Ishidoya, S., Aoki, S., Goto, D., Nakazawa, T., Taguchi, S. and co-authors. 2012a. Time and space variations of the O_2/N_2 ratio in the troposphere over Japan and estimation of global CO_2 budget. *Tellus B*. **64**, 18964.
- Ishidoya, S., Aoki, S. and Nakazawa, T. 2003. High precision measurements of the atmospheric O_2/N_2 ratio on a mass spectrometer. *J. Meteorol. Soc. Jpn.* **81**, 127–140.
- Ishidoya, S., Morimoto, S., Aoki, S., Taguchi, S., Goto, D. and co-authors. 2012b. Oceanic and terrestrial biospheric CO_2 uptake estimated from atmospheric potential oxygen observed at Ny-Alesund, Svalbard, and Syowa, Antarctica. *Tellus B*. **64**, 18924.

- Ishidoya, S. and Murayama, S. 2013. Development of a new high precision continuous measuring system for atmospheric O₂/N₂ and Ar/N₂ and its application to the observation in Tsukuba, Japan. *submitted to Tellus B*.
- Keeling, R. F. 1988. Development of an Interferometric Oxygen Analyzer for Precise Measurement of the Atmospheric O₂ Mole Fraction. PhD Thesis. Harvard University, Cambridge.
- Keeling, R. F., Blaine, T., Paplawsky, B., Katz, L., Atwood, C. and co-authors. 2004. Measurement of changes in atmospheric Ar/N₂ ratio using a rapid-switching, single-capillary mass spectrometer system. *Tellus B*. **56**, 322–338.
- Keeling, R. F., Manning, A. C., McEvoy, E. M. and Shertz, S. R. 1998. Methods for measuring changes in atmospheric O₂ concentration and their application in southern hemisphere air. *J. Geophys. Res.* **103**, 3381–3397.
- Keeling, R. F., Piper, S. C. and Heimann, M. 1996. Global and hemispheric CO₂ sinks deduced from changes in atmospheric O₂ concentration. *Nature*. **381**(6579), 218–221.
- Keeling, R. F. and Shertz, S. R. 1992. Seasonal and interannual variations in atmospheric oxygen and implications for the global carbon cycle. *Nature*. **358**, 723–727.
- Kondo, H., Saigusa, N., Murayama, S., Yamamoto, S. and Kannari, A. 2001. A numerical simulation of the daily variation of CO₂ in the central part of Japan summer case. *J. Meteorol. Soc. Jpn.* **79**, 11–21.
- Kozlova, E., Manning, A. C., Jordan, A. and Brand, W. 2005. Investigations of the land biotic O₂:CO₂ exchange ratios in photosynthesis and respiration. In *Proceedings (CD) of 7th International Carbon Dioxide Conference*, Boulder, CO, September 25–30.
- Kozlova, E. A., Manning, A. C., Kisilyakhov, Y., Seifert, T. and Heimann, M. 2008. Seasonal, synoptic, and diurnal-scale variability of biogeochemical trace gases and O₂ from a 300-m tall tower in central Siberia. *Glob. Biogeochem. Cycles*. **22**, GB4020. DOI: 10.1029/2008GB003209.
- Langenfelds, R. L., Francey, R. J., Steele, L. P., Battle, M., Keeling, R. F. and co-authors. 1999. Partitioning of the global fossil CO₂ sink using a 19-year trend in atmospheric O₂. *Geophys. Res. Lett.* **26**(13), 1897–1900.
- Machta, L. and Hughes, E. 1970. Atmospheric oxygen in 1967 to 1970. *Science* **168**, 1582–1584.
- Manning, A. C. and Keeling, R. F. 2006. Global oceanic and terrestrial biospheric carbon sinks from the Scripps atmospheric oxygen flask sampling network. *Tellus B*. **58**, 95–116.
- Manning, A. C., Keeling, R. F. and Severinghaus, J. P. 1999. Precise atmospheric oxygen measurements with a paramagnetic oxygen analyzer. *Glob. Biogeochem. Cycles*. **13**, 1107–1115.
- Murayama, S., Saigusa, N., Chan, D., Yamamoto, S., Kondo, H. and co-authors. 2003. Temporal variations of atmospheric CO₂ concentration in a temperate deciduous forest in central Japan. *Tellus B*. **55**, 232–243.
- Murayama, S., Takamura, C., Yamamoto, S., Saigusa, N., Morimoto, S. and co-authors. 2010. Seasonal variations of atmospheric CO₂, δ¹³C, and δ¹⁸O at a cool temperate deciduous forest in Japan: influence of Asian monsoon. *J. Geophys. Res.* **115**, D17304. DOI: 10.1029/2009JD013626.
- Nakazawa, T., Aoki, S., Murayama, S., Fukabori, M., Yamanouchi, T. and co-authors. 1991. The concentration of atmospheric carbon dioxide at Japanese Antarctic station, Syowa. *Tellus B*. **43**, 126–135.
- Nicolet, M. 1960. The properties and constitution of the upper atmosphere. In: *Physics of the Upper Atmosphere* (ed. J. A. Ratcliffe). Elsevier, New York, pp. 17–71.
- Saigusa, N., Yamamoto, S., Murayama, S. and Kondo, H. 2005. Inter-annual variability of carbon budget components in and ASIAFLUX forest site estimated by long-term flux measurement. *Agr. Forest. Meteorol.* **134**, 4–16.
- Sarmiento, J. L., Gloor, M., Gruber, N., Beaulieu, C., Jacobson, A. R. and co-authors. 2010. Trends and regional distributions of land and ocean carbon sinks. *Biogeosciences*. **7**, 2351–2367.
- Seibt, U., Brand, W. A., Heimann, M., Lloyd, J., Severinghaus, J. P. and co-authors. 2004. Observations of O₂:CO₂ exchange ratios during ecosystem gas exchange. *Glob. Biogeochem. Cycles*. **18**, GB4024. DOI: 10.1029/2004GB002242.
- Severinghaus, J. 1995. Studies of the Terrestrial O₂ and Carbon Cycles in Sand Dune Gases and in Biosphere 2. PhD Thesis, Columbia University, New York.
- Stephens, B., Keeling, R., Heimann, M., Six, K., Murnane, R. and co-authors. 1998. Testing global ocean carbon cycle models using measurements of atmospheric O₂ and CO₂ concentration. *Glob. Biogeochem. Cycles*. **12**(2), 213–230.
- Stephens, B. B., Bakwin, P. S., Tans, P. P., TecLaw, R. M. and Baumann, D. 2007. Application of a differential fuel-cell analyzer for measuring atmospheric oxygen variations. *J. Atmos. Ocean. Tech.* **24**, 82–94.
- Sturm, P., Leuenberger, M. and Schmidt, M. 2005. Atmospheric O₂, CO₂ and δ¹³C observations from the remote sites Jungfraujoch, Switzerland, and Puy de Dôme, France. *Geophys. Res. Lett.* **32**, L17811. DOI: 10.1029/2005GL023304.
- Tanaka, M., Nakazawa, T. and Aoki, S. 1983. High quality measurements of the concentration of atmospheric carbon dioxide. *J. Meteorol. Soc. Jpn.* **61**, 678–685.
- Tohjima, Y., Mukai, H., Nojiri, Y., Yamagishi, H. and Machida, T. 2008. Atmospheric O₂/N₂ measurements at two Japanese sites: estimation of global oceanic and land biotic carbon sinks and analysis of the variations in atmospheric potential oxygen (APO). *Tellus B*. **60**, 213–225.
- van der Laan-Luijkx, I. T., Karstens, U., Steinbach, J., Gerbig, C., Sirignano, C. and co-authors. 2010. CO₂, δO₂/N₂ and APO: observations from the Lutjewad, Mace Head and F3 platform flask sampling network. *Atmos. Chem. Phys.* **10**, 10691–10704. DOI: 10.5194/acp-10-10691-2010.
- Yamamoto, S., Murayama, S., Saigusa, N. and Kondo, H. 1999. Seasonal and inter-annual variation of CO₂ flux between a temperate forest and the atmosphere in Japan. *Tellus B*. **51**, 402–413.