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IN VITRO RADICAL SCAVENGING ACTIVITY OF CEPHALANDRA INDICA

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ABSTRACT

Cephalandra indica (Bimba, kanduri, Cucurbitaceae) is famous for its hypoglycemic and anti-diabetic Properties in Ayurvedic system of medicine. Other applications include the therapy of various conditions such as skin diseases and gonorrhoea. There are many patented formulations derived from *Cephalandra indica*. Which are now distributed increasingly all over the world this has given rise to a

concomitant increase in research on the phytochemical constituents and biological activity of *C. indica*. Plants produce a diverse range of bioactive molecules, making them a rich source of different types of medicines. In the present study, we examined the anti-oxidant effects of root of *Cephalandra indica*. Dried and powdered root of C.indica were extracted with Ethyl acetate, n-butanol and methanol. Antioxidant assays were carried out by using different *in vitro* models such as, DPPH radical scavenging activity. The EC₅₀ values of different extract for DPPH radical scavenging activity was found. In DPPH radical scavenging assay the IC₅₀ value of the ethyl acetate extract was found to be $330\mu g/mL$ while β - carotene had the IC₅₀ value 220 µg/ml. Moreover *C.indica* Extract showed strong reducing power and total antioxidant capacity. The anti-oxidant activities of other solvent extracts were poor when compared to the Ethyl acetate extract. These results suggest that, the active antioxidant compounds are better extracted in Ethyl acetate and there is a direct correlation between the flavonoid glycoside and its anti-oxidant activity.

KEYWORDS: Medicinal plant; Cephalandra indica, solvent extracts; anti-oxidant activity.

INTRODUCTION

Medicinal plants have occupied a vital place in the socio-cultural, development of rural people of India. *Cephalandra indica* (family- Cucurbitaceae) is perennial; scandent or

prostrate plant which is commonly known as bimbu in Bengali, kova in Malayalam, kundru in Punjabi, kundaru in Urdu and bhimb in Hindi.^[24] Leaves are 5- 10 cm, long and broad, bright green above, paler beneath, studded and sometimes rough with papillae. Flowers are peduncles and sub-fusiform. The fresh tab root is thick, tuberous, long tapering, more or less tortuous with a few fibrous rootlets attached to it.^[6] Roots are flexible, soft and break with a fibrous fracture.^[4,5] The cork is composed of rows of cells. Fruits are fusiform-ellipsoid, slightly beaked, 2.5-5 by 1.3-2.5 cm marked when immature with white streaks, bright scarlet when fully ripped.^[15,18] Seeds are obovoid, rounded at the apex, slightly papillose, much compressed and yellowish grey. This plant is traditionally used in various diseases like psoriasis, ringworm, itching, small pox, skin diseases, ulcer, scabies, diabetes, asthma, bronchitis, dysentery, vomiting, cough and cold. Pharmacological studies on *C.indica* demonstrated antimicrobial, antilitihic and antioxidant activities.^[1,2] The purpose of the present study is to in-vitro methods **DPPH radical scavenging activity** of different solvent extract of *Cephalandra indica*.^[4]

MATERIALS AND METHODS

Chemicals: 1, 1-Diphenyl-2-picryl-hydrazyl (DPPH) was obtained from Sigma chemicals, Co., Chennai. All other chemicals used were of analytical grade.^[16]

Phytochemical screening: The freshly prepared extract of *C.indica* was qualitatively tested for the presence of chemical constituents. Phyto-chemical screening of the extract was performed using the following reagents and chemicals: Alkaloids with Dragendorff reagent, flavonoids with the use of Mg and HCl; tannins with ferric chloride and potassium dichromate solutions and Saponins with ability to produce suds. Gum was tested using Molish reagents and concentrated sulfuric acid. These were identified by characteristic color changes using standard procedures.^[22, 23]

Preparation of crude plant extracts: Test plants were collected locally or obtained from the local market. Plant material consisting of mature plant *Cephalandra indica* root were collected and dried. The dried root materials were powdered using a grinder. The extraction was done at room temperature.^[7, 8] About 100 g of dried, Ground plant materials were soaked in methanol (1 L of 98%) for 5-7 days separately.^[9] The soaked material was stirred every 18 h using a sterilized glass rod. The final extracts were passed through Whatman filter paper No.1. The filtrates obtained were concentrated under vacuum on a rotary evaporator at 40°C and stored at 4^o C for further use. The stock solution takes different fraction n- butanol, Ethyl

acetate and methanol. The crude extracts (5 mg/ml) were prepared by dissolving a known amount of dry extract in different fraction.^[19, 20] The Working solutions (1, 2, 4, 6, 8, 10, 15, 25, 50, 75, 100, 250, 500 and 750 μ g /ml) of the extracts was Prepared from the stock solution using suitable dilution.

Assay of DPPH test in-vitro radicals scavenging and antioxidant

Sample preparation

- 1) Ethyl acetate fraction
- 2) Butanol fraction
- 3) Methanol fraction

The effects of extract on DPPH radical were estimated by with miner modification. In brief, 2MI of DPPH in methanol (3.6×10^{-5}) is added to 50µl of various concentrations of extract / fraction. The mixture is vortexes for 15sec and left to stand at 37°c for 30 min¹⁰. The decrease in the absorbance at 515nm was continuously recorded in a spectrophotometer for 15 min at room temp. All determination was performed in triplicate. The DPPH scavenging activity (decrease of absorbance at 515nm) of Extracts and compound were plotted against time and the (%) percentage of DPPH radicals scavenging ability of the sample were calculated from the absorbance value and 15min duration as follows (Table-2).

% Inhibition = (Abs Control – Abs Sample) x 100 Abs Control

Reducing power: The reducing power of *Cephalandra indica was* determined according to the method previously described.^[11,12] Different concentrations of *Cephalandra indica* extract (100 μ g – 1,000 μ g) in 1 ml of distilled water was mixed with phosphate buffer (2.5 ml, 0.2 M, pH 6.6) and potassium ferricyanide [K3Fe(CN)6] (2.5 ml, 1%). The mixture was incubated at 50°C for 20 min. A portion (2.5 ml) of trichloro acetic acid (10%) was added to the mixture, which was then centrifuged at 3,000 rpm for 10 min. The upper layer of the solution (2.5 ml) was mixed with distilled water (2.5 ml) and ferric chloride (0.5 ml. 0.1%) and the absorbance was measured at 700 nm. Increased absorbance of the reaction mixture indicated increased reducing power. β - Carotene was used as a reference standard. Phosphate buffer (pH 6.6) was used as blank solution.^[13] The absorbance of the final reaction mixture of two parallel experiments were taken was expressed as mean ± standard deviation.

RESULTS AND DISCUSSION

Preliminary phytochemical screening of the extract of C. indica revealed the presence of various bioactive components of which flavonoids and tannins were the most prominent and the result of phyto-chemical test has been summarized in the Table 1. Phenolic compounds and flavonoids have been reported to be associated with anti-oxidative action in biological systems, acting as scavengers of singlet oxygen and free radicals. Literature reviews about the plant also confirm the presence of terpinoids group. Many naturally occurring triterpinoids exhibited a good anti-inflammatory activity have been isolated from various plants. Triterpinoids have a wide spectrum of biological activities and some of them may be useful in medicine.^[14,21] There is growing interest in natural triterpinoids caused as much by the scientific aspects extraction and structural analysis of these compounds, as by the fact of their wide spectrum of biological activities, they are bactericidal, fungicidal, antiviral, cytotoxic, analgesic, anti-inflammatory, anti-cancer and anti-allergic.^[24] In this present study the antioxidant activity of the methanol extracts of the root of C. *indica* were investigated by using DPPH scavenging assay, reducing power of the extract and by determining total antioxidant capacity of the extract. The two methods have proven the effectiveness of the Ethyl acetate extract compared to the reference standard antioxidant β -Carotene.^[25] The DPPH antioxidant assay is based on the ability of DPPH, a stable free radical, to decolorize in the presence of antioxidants. The DPPH radical contains an odd electron, which is responsible for the absorbance at 517 nm and also for visible deep purple color. When DPPH accepts an electron donated by an antioxidant compound, the DPPH is decolorized which can be quantitatively measured from the changes in absorbance.^[30] Comparison of the antioxidant activity of the extract and β – carotene, the Ethyl acetate extract of C. *indica* exhibited a significant dose dependent inhibition of DPPH activity, with a 50% inhibition (IC₅₀) at a concentration of 330 μ g/ml(Table-3). The IC₅₀ value of the extract was found to be comparable to reference standard β - carotene (IC₅₀ 220 µg/ml). The reducing ability of a compound generally depends on the presence of reductants, which have been exhibited antioxidative potential by breaking the free radical chain, donating a hydrogen atom.^[26,27] The presence of reduction (i.e. antioxidants) in C. indica extract causes the reduction of the Fe3+/ ferric-cyanide complex to the ferrous form.^[3] Therefore, the Fe2+ can be monitored by measuring the formation of Perl's Prussian blue at 700 nm.^[28] The reductive capabilities of the plant extract compared to β - carotene.^[4] The reducing power of extract of *Cephalantra indica* was very potent and the power of the extract was increased with quantity of sample.

PHYTOCHEMICAL STUDIES OF ROOT

Table -	- 1
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Chemical constituents Cephalandra indica					
Alkaloids Flavonoids		Glycosides	Tannin	Carbohydrates	triterpinoids
+	+ + +	+ + +	+ + +	+ + +	+ +

Table: 2: DPPE	I radical scav	venging act	tivities of	Cephalandr	a indica
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Concentrat	Optical density			
ion	β-Carotene	Ethyl acetate	Butanol	methanolic
1011		fraction	fraction	fraction
50	0.751±0.022	0.844 ± 0.011	0.875 ± 0.022	0.890±0.010
100	0.591±0.004	0.715 ± 0.045	0.785 ± 0.010	0.835 ± 0.007
250	0.363±0.011	0.606 ± 0.048	0.630±0.021	0.687 ± 0.009
500	0.291±0.005	0.420 ± 0.038	0.462 ± 0.034	0.490 ± 0.005
1000	0.202 ± 0.007	0.341 ± 0.003	0.349 ± 0.007	0.365 ± 0.007

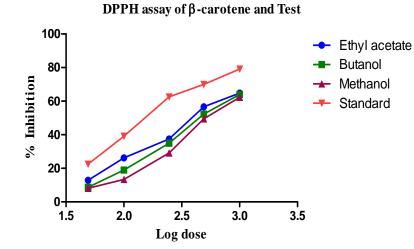


Figure: 1: DPPH free radical scavenging activity of standard β – Carotene

Concentration	% of Inhibition				
Concentration	β- Carotene	Ethyl acetate fraction	Butanol fraction	50% methanol fraction	
50	22.497 ± 0.2782	12.899 ± 0.3472	9.700 ± 0.6438	8.152 ± 0.3592	
100	39.112 ± 0.3852	26.212 ± 0.3485	18.988 ± 0.5644	13.400 ± 0.9266	
250	62.583 ± 0.1877	37.461 ± 0.3078	34.984 ± 0.5977	29.102 ± 0.5258	
500	69.969 ± 0.4381	56.656 ± 0.5891	52.321 ± 0.4936	49.432 ± 0.3874	
1000	79.153 ± 0.4223	64.809 ± 0.3241	63.983 ± 0.3464	62.332 ± 0.5858	
IC50 Value	220µg/ml	330µg/ml	510µm/ml	530µg/ml	

Table: 3: DPPH radical scavenging activities of Cephalandra indica in IC₅₀ value

CONCLUSION

C. indica is a famous plant for its safe anti-diabetic property. It proved the insulin stimulatory effect of *C indica* leaves from existing b-cells in diabetic rats. It possesses hypoglycemic, anti-diabetic, hypolipidemic, hepatoprotective, larvvicidal, anti-inflammatory, analgesic and

antipyretic activities (Figure-1). It is found to be devoid of anti-tuberculosis properties. Various phytoconstituents reported in *C. indica* are cephalandrol, tritriacontane, lupeol, b-sitosterol, cephalandrine A, cephalandrine B, stigma-7-en-3- one, taraxerone and taraxerol. Terpenoids are found to be responsible for anti-diabetic activity (Table-5). Despite the broad use of *C. indica* in traditional medicine, very few systematic pharmacological and phytochemical studies are reported till date assessing its therapeutic properties.

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