ejpmr, 2018,5(2), 421-427

EUROPEAN JOURNAL OF PHARMACEUTICAL AND MEDICAL RESEARCH

www.ejpmr.com

SJIF Impact Factor 4.161

<u>Research Article</u> ISSN 2394-3211

EJPMR

# ANALYSIS OF FUNCTIONAL GROUPS IN THE VERMICOMPOST OF FRUIT WASTE USING EUDRILUS EUGENIAE

# Sumathi S.<sup>1</sup> and Dr. Pawlin Vasanthi Joseph<sup>\*2</sup>

<sup>1</sup>Research Scholar Department of Zoology Nirmala College for Women (Autonomous) Coimbatore-641018 Tamil Nadu, India.

<sup>2</sup>Associate Professor and Head Department of Zoology Nirmala College for Women (Autonomous) Coimbatore-641018 Tamil Nadu, India.

\*Corresponding Author: Dr. Pawlin Vasanthi Joseph

Associate Professor and Head Department of Zoology Nirmala College for Women (Autonomous) Coimbatore-641018 Tamil Nadu, India.

Article Received on 11/12/2017

Article Revised on 02/01/2018

Article Accepted on 23/01/2018

# ABSTRACT

Environmental degradation is a major threat confronting the world and the rampant use of chemical fertilizers contributes largely to the deterioration of the environment through depletion of fossil fuels, generation of carbon dioxide and contamination of water resources. Vermicomposting, which involves the composting of organic wastes through earthworm activity, has proven to be successful in processing sewage sludge and solids. The objective of the research is to assess the presence of functional groups in the vermicompost through FTIR analysis and to estimate the degradability of fruit waste into useful manure through vermicomposting. Fruit waste used was banana and papaya in the vermicomposting process by *Eudrilus eugeniae*. Respective controls were maintained devoid of earthworms for a period of 60 days. FTIR Analysis was carried out on the 60<sup>th</sup> day. The vermicompost of banana waste showed strong peaks for the functional groups ketone, aldehyde and amides. The results confirm that there is more reduction of aromatic structures, aliphatic and polysaccharides in the vermicompost using *Eudrilus eugeniae*.

KEYWORDS: Vermicomposting, FTIR, Functional groups, Eudrilus eugeniae, fruit waste.

# INTRODUCTION

Environmental degradation is a major threat confronting the world and the rampant use of chemical fertilizers contributes largely to the deterioration of the environment through depletion of fossil fuels, generation of carbon dioxide and contamination of water resources. It leads to loss of soil fertility due to imbalanced use of fertilizers that has adversely impacted agricultural productivity and causes soil degradation. Now there is a growing realization that the adoption of ecological and sustainable farming practices can only reverse the declining trend in the global productivity and environment protection (Aveyard, 1988; Wani and Lee, 1992; Waniet *al.*, 1995).

Vermicomposting, which involves the composting of organic wastes through earthworm activity, has proven to be successful in processing sewage sludge and solids from wastewater (Neuhauser*et al.*, 1988; Dominguez *et al.*, 2000), materials from breweries (Butt, 1993), paper waste (Butt, 1993; Elvira *et al.*, 1995), urban residues, food and animal wastes (Allevi*et al.*, 1987; Edwards 1988; Elvira *et al.*, 1996) as well as horticultural residues from processed potatoes, dead plants and the mushroom industry (Edwards, 1988).

Benitez *et al.*, (1999) concluded that in the vermicomposting process, inoculated earthworm rich and complex organic substances into stabilized humus-like product maintains aerobic condition in the organic wastes, converts a portion of the organic material into worm biomass and respiration products and expels the remaining partially stabilized product. The potential of some epigeic earthworm to recycle organic waste materials into value-added products is well documented (Kale *et al.*, 1982; Elvira*et al.*, 1998; Garg and Kaushik, 2005; Garg*et al.*, 2006; Suthar, 2006).

Coimbatore is an important industrial city in Tamil Nadu. Urban population of the city has increased from 14.61 lakhs in 2001 to 21.51 lakhs in 2011. Hence, the demand for fresh fruits and vegetables has also increased over the years. Cultivation of fruits and vegetable is picking up in the peri-urban areas of the city due to better demand prediction, easy market access, reduced transportation cost and assured market. Hence, the demand of fruits and vegetable of Coimbatore city is mostly met from the peri-urban areas. Banana is an important fruit commonly present in the food basket of most of the consumers in the city. The present study primarily deals with the management of different types of fruit waste generated in vegetable markets, fruit shops and Juice outlets using the earthworm *Eudrilus eugeniae*. The performance of the composting process and the quality of the end product is assessed by the combined application of independent methodologies like physical, chemical, microbiological and statistical methods for the determination of its stability as manure. The objective of the research is to assess the presence of functional groups in the vermicompost through FTIR analysis and to estimate the degradability of fruit waste into useful manure through vermicomposting.

#### MATERIALS AND METHODS

The earthworm species, *Eudrilus eugeniae* was obtained from the vermicompost pit of Nirmala College for Women (Autonomous), Coimbatore.

Two different types of fruit waste generated from Pazhamudhir Nilayam, Vegetable market Coimbatore were collected and segregated as banana peel waste and papaya peel waste. The fruit wastes were shredded manually. Cow dung was obtained from a local cowshed and was sun dried and flaked.

The experiment was conducted in square plastic pots measuring 17 ×17 ×17 cm of length, breadth and height respectively. Holes were drilled at the bottom of the pots so as to drain excess water. The pots were filled from bottom up with successive layers of pebbles, coconut husk, cow dung flakes and shredded fruit peels respectively. The fruit waste was mixed with cow dung flakes in the ratio of 1:1. All pots were maintained in triplicates. Water was sprinkled daily on all pots to maintain the moisture content and turned at regular intervals for proper mixing and aeration. The experimental pots were kept under shade and covered with gunny bags to prevent moisture loss. This setup was maintained for 15 days for partial degradation and stabilization. After 15 days, 20 non-clitellated earthworms were introduced into each treatment pots containing banana peel waste and papaya peel waste. The control pots of banana peel waste and papaya peel waste were devoid of earthworms. This setup was also sprinkled with water daily and was monitored for a period of 60 days. On the 60th day of the experimental period, the samples of compost and vermicompost from all experimental units were collected and used for FTIR analysis.

#### FTIR Analysis

The Fourier transform Infrared (FTIR) analysis of the samples was carried out by FTIR equipment of mark spectrum incorporated with software (Perkin Elmer Instruments version 3.02.01) for the examination of the spectra. For sample analysis, 0.5 g of activated carbon was mixed with about the same amount of potassium bromide (Kbr). The mixture thus obtained was crushed in

a mortar to obtain a homogeneous powder which was introduced into a mould to obtain very fine plates. The plates were then introduced in to the spectrophotometer for analysis. The wave number was found to vary between 4000 and  $350 \text{ cm}^{-1}$ .

#### **RESULTS AND DISCUSSION**

The vermicompost of banana waste shows strong peaks for the functional groups amides and amines. Amine group shows N-H stretch while amide group shows N-H stretch and N-H bending in the vermicompost of banana waste (Fig 2 and Table 2).

The vermicompost of papaya waste shows strong peaks for the functional groups ketone, aldehyde and amides. Ketone group shows a, b – unsaturated and aryl ketone. Aldehyde group shows C=O stretch and N-H bending in the vermicompst of papaya waste (Fig 4 and Table 4).

The results confirm that there is more reduction of aromatic structures, aliphatic and polysaccharides in the vermicompost using *Eudrilus eugeniae*. This is associated with organic matter mineralization and it indicates the maturity and stability of the final product when compared to the control (Fig 1 and 3 and Table 1 and 3).

The FTIR analysis clearly confirms the higher degree of degradation, since the spectra of the final compound has clearly depicted the presence of amide groups in the vermicompost samples prepared from fruit wastes.

Carboxylic acid present in fruit peel has pharmaceutical properties. It is helpful in curing diseases such as ulcers, jaundice, headache, stomatitis, hemicranias, fever, pain in liver and wounds in cattle as well as in the treatment of edema and rheumatic joint pains (Nair et al., 2013). Besides, carboxylic acid also makes fruit peels suitable for metal adsorption. The major source of carboxylic acid in fruit peel is either pectin or cellulose or lignin (Schiewer and Iqbal, 2010; Schiewer and Patil, 2008) Amines, amides and amino acids are the main groups in fruit peels and owing to their presence fruit peels can be used for protein synthesis. The hydroxyl group in fruit peels plays a vital role in adsorption of anionic impurities such as dyes. Fruit peels does not show any peak between the regions 2220 and 2260 cm<sup>-1</sup>, which suggests the absence of cyanide groups, confirming that fruit peels studied do not contain any toxic substances (Zouet al., 2012). The amine, amide and amino groups in fruit peels are a good source of nitrogen.

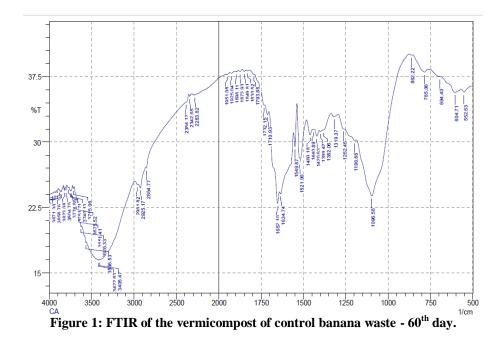


 Table 1: IR Absorption Frequencies of Functional Groups in the vermicompost of control banana waste - 60<sup>th</sup>

 day.

| S. NO. | PEAK    | INTENSITY | AREA   | FUNCTIONAL GROUPS  | TYPE OF VIBRATION     |
|--------|---------|-----------|--------|--|-----------------------|
| 1      | 552.63  | 35.72     | 25.63  | Alkyl Halide   | C-Br stretch          |
| 2      | 604.7   | 35.7      | 52.46  | Alkyl Halide   | C-cl stretch          |
| 3      | 785.06  | 38.03     | 40.54  | Alkyl Halide   | C-cl stretch          |
|        |         |           |        | Alcohol  | C-O stretch           |
| 4      | 1096.58 | 23.82     | 158.97 | Alkyl Halide   | C-F stretch           |
| 4      | 1090.38 | 25.82     | 136.97 | Amine  | C-N stretch           |
|        |         |           |        | Ester  | C-O stretch           |
|        |         |           |        | Alkyl Halide   | C-F stretch           |
| 5      | 1196.88 | 30.15     | 36.1   | Amine  | C-N stretch           |
|        |         |           |        | Ester  | C-O stretch           |
| 6      | 1319.37 | 32.93     | 19.47  | Alkyl Halide   | C-F stretch           |
| 0      | 1519.57 | 52.95     | 19.47  | Acid   | C-O stretch           |
| 7      | 1420.63 | 30.68     | 13.72  | Aromatic   | C=C stretch           |
| 8      | 1460.18 | 30.83     | 16.96  | Aromatic   | C=C stretch           |
| 9      | 1521.9  | 27.75     | 29.01  | Aromatic   | C=C stretch           |
| 9      | 1321.9  | 21.15     | 29.01  | Nitro  | N-O stretch           |
| 10     | 1549.87 | 29.49     | 9.77   | Aromatic   | C=C stretch           |
| 10     | 1349.07 | 29.49     | 9.77   | Alkyl HalideAlkyl HalideAlkyl HalideAlcoholAlkyl HalideAmineEsterAlkyl HalideAmineEsterAlkyl HalideAmineEsterAlkyl HalideAcidAromaticAromaticNitroAlkaneAmideAmideAlkaneAmideAlkaneAmideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAnideAlkaneAcidAlcohol  | N-O stretch           |
|        |         |           |        | Alkane   | C=C stretch           |
| 11     | 1634.74 | 24.05     | 47.35  | Amide  | C=O stretch           |
|        |         |           |        | Amide  | N-H Bending           |
| 12     | 1652.1  | 22.9      | 34.7   | Alkane   | C=C stretch           |
| 12     | 1052.1  | 22.9      | 54.7   | Amide  | C=O stretch           |
|        |         |           |        |  | C=O stretch           |
| 13     | 1710.93 | 33.15     | 11.43  |  | C=O stretch           |
|        |         |           |        | Ketone   | Acyclic stretch       |
|        |         |           |        |  | C=H stretch           |
| 14     | 2925.17 | 24.76     | 51.65  | Acid   | O-H stretch           |
|        |         |           |        | Ester<br>Alkyl Halide<br>Amine<br>Ester<br>Alkyl Halide<br>Acid<br>Aromatic<br>Aromatic<br>Aromatic<br>Nitro<br>Aromatic<br>Nitro<br>Aromatic<br>Nitro<br>Aromatic<br>Nitro<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Carbonyl<br>Acid<br>Ketone<br>Alkane<br>Anide<br>Acid<br>Acid<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Amide<br>Alkane<br>Anide<br>Alkane<br>Anide<br>Acid<br>Ketone<br>Alkane<br>Alkane<br>Alkane<br>Anide<br>Alkane<br>Anide<br>Acid<br>Ketone<br>Alkane<br>Alkane<br>Alkane<br>Anide<br>Acid<br>Ketone<br>Alkane<br>Alkane<br>Alkane<br>Alkane<br>Anide<br>Acid<br>Ketone<br>Alkane<br>Alkane<br>Alkane<br>Alkane<br>Alkane<br>Anide<br>Acid<br>Alcohol<br>Amine<br>Alcohol<br>Amine | =C-H stretch          |
|        |         |           |        |  | O-H stretch, H-bonded |
| 15     | 3405.47 | 16.51     | 292.71 |  | N-H stretch           |
|        |         |           |        |  | N-H stretch           |
|        |         |           |        |  | O-H stretch, H-bonded |
| 16     | 3422.83 | 16.46     | 113.8  |  | N-H stretch           |
|        |         |           |        | Amide  | N-H stretch           |

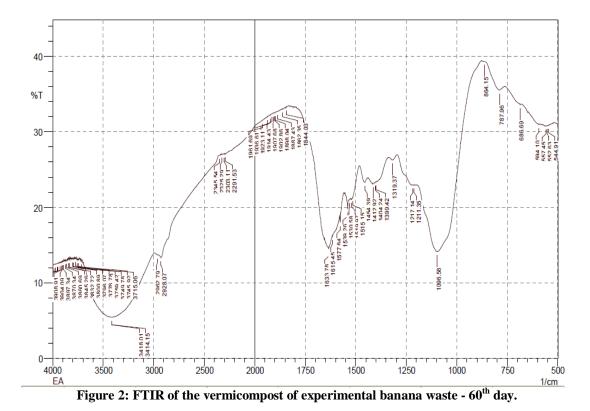


 Table 2: IR Absorption Frequencies of Functional Groups in the vermicompost of experimental banana waste - 60<sup>th</sup> day.

| ay.    | 7.                      |           |         |                   |                          |  |  |
|--------|-------------------------|-----------|---------|-------------------|--------------------------|--|--|
| S. NO. | PEAK                    | INTENSITY | AREA    | FUNCTIONAL GROUPS | <b>TYPE OF VIBRATION</b> |  |  |
| 1      | 787.96                  | 35.56     | 11.189  | Alkyl Halide      | C-cl stretch             |  |  |
|        |                         |           |         | Alcohol           | C-O stretch              |  |  |
| 2      | 1096.58                 | 14.136    | 208.863 | Alkyl Halide      | C-F stretch              |  |  |
| 2      | 1090.38                 |           |         | Amine             | C-N stretch              |  |  |
|        |                         |           |         | Ester             | C-O stretch              |  |  |
| 3      | 1319.37                 | 26.277    | 23.336  | Alkyl Halide      | C-F stretch              |  |  |
| 4      | 1454.39                 | 23.324    | 7.864   | Alkane            | -C-H bending             |  |  |
|        |                         |           |         | Nitro             | N-O stretch              |  |  |
| 5      | 1530.58                 | 21.029    | 2.605   | Aromatic          | C=C stretch              |  |  |
|        |                         |           |         | Amide             | N-H bending              |  |  |
|        |                         |           |         | Alkene            | C=C stretch              |  |  |
| 6      | 1633.78                 | 14.523    | 15.747  | Aromatic          | C=C stretch              |  |  |
| 0      | 1055.78                 |           |         | Carbonyl          | C=O stretch              |  |  |
|        |                         |           |         | Amide             | N-H bending              |  |  |
| 7      | 2928.07 13.345          | 12 245    | 414.834 | Alkane            | C-H stretch              |  |  |
| /      | 2928.07                 | 15.545    | 414.654 | Acid              | O-H stretch              |  |  |
| 8      | 2062 70                 | 12 604    | 25.715  | Alkane            | C-H stretch              |  |  |
| 0      | 8 2962.79 13.694 25.715 | 23.715    | Acid    | O-H stretch       |                          |  |  |
|        |                         |           |         | Alcohol           | O-H stretch, H-bonded    |  |  |
| 9      | 3414.15                 | 5.444     | 457.104 | Amine             | N-H stretch              |  |  |
|        |                         |           |         | Amide             | N-H stretch              |  |  |
|        | 3418.01                 | 5.445     | 6.095   | Alcohol           | O-H stretch, H-bonded    |  |  |
| 10     |                         |           |         | Amine             | N-H stretch              |  |  |
|        |                         |           |         | Amide             | N-H stretch              |  |  |
|        |                         |           |         |                   |                          |  |  |

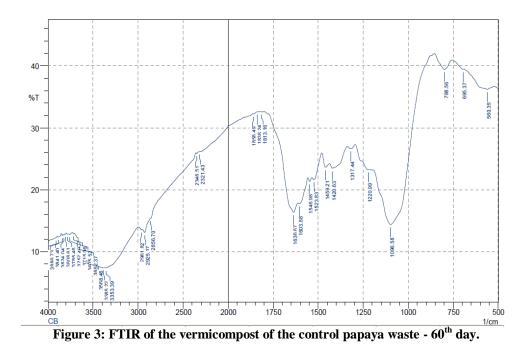


Table 3: IR Absorption Frequencies of Functional Groups in the vermicompost of control papaya waste - 60<sup>th</sup> day.

| S. NO. | PEAK    | INTENSITY | AREA   | FUNCTIONAL GROUPS | <b>TYPE OF VIBRATION</b> |
|--------|---------|-----------|--|-------------------|--------------------------|
| 1      | 798.56  | 39.437    | 27.00  | Alkene            | =C-H bending             |
| 1      | 790.30  | 39.437    | 57.99  | Alkyl Halide      | C-cl stretch             |
| 2      | 1096.58 | 14.394    | 222.2  | Alcohol           | C-O stretch              |
| 2      | 1090.38 | 14.394    | 222.3  | Ester             | C-O stretch              |
|        |         |           |  | Alkyl Halide      | C-F stretch              |
| 3      | 1317.44 | 26.631    | 24.231   | Amine             | C-N stretch              |
|        |         |           |  | Acid              | C-O stretch              |
| 4      | 1420.63 | 23.483    | 50.064   | Alkane            | C-H bending              |
| 4      | 1420.05 | 23.463    | 39.904   | Aromatic          | C=C stretch              |
| 5      | 1450 21 | 22 622    | 25 400   | Alkane            | C-H bending              |
| 3      | 1459.21 | 23.633    | 23.499   | Aromatic          | C=C stretch              |
| 6      | 1523.83 | 21.664    | 21 228   | Aromatic          | C=C stretch              |
| 0      | 1525.65 | 21.004    | 34.228   | Nitro             | N-O stretch              |
| 7      | 1546.98 | 21.343    | 37.99         222.3         24.231         59.964         25.499         34.228         14.106         115.773         341.083         76.961         34.943         374.604         213.363   | Aromatic          | C=C stretch              |
| /      | 1340.98 | 21.545    |  | Nitro             | N-O stretch              |
| 8      | 1636.67 | 16.364    | 115.773  | Alkene            | C=C stretch              |
| 9      | 2856.7  | 15.419    | 241.092  | C=C stretch       | C-H stretch              |
| 9      | 2030.7  | 13.419    | 341.065  | N-O stretch       | O-H stretch              |
| 10     | 2925.17 | 13.127    | 76.061   | C=C stretch       | C-H stretch              |
| 10     | 2923.17 | 15.127    | /0.901   | N-O stretch       | O-H stretch              |
| 11     | 2961.82 | 13.536    | 24.042   | C=C stretch       | C-H stretch              |
| 11     | 2901.62 | 15.550    | Acid59.964Alkane<br>Aromatic25.499Alkane<br>Aromatic34.228Aromatic<br>Nitro14.106Aromatic<br>Nitro115.773Alkene341.083C=C stretch<br>N-O stretch76.961C=C stretch<br>N-O stretch34.943C=C stretch<br>N-O stretch374.604Amine<br>AmideAlcohol | O-H stretch       |                          |
|        |         |           |  | Alcohol           | O-H, stretch H-bonded    |
| 12     | 3353.39 | 7.394     | 374.604  | Amine             | N-H stretch              |
|        |         |           |  | Amide             | N-H stretch              |
|        | 3385.22 | 7.407     | 213.363  | Alcohol           | O-H, stretch H-bonded    |
| 13     |         |           |  | Amine             | N-H stretch              |
|        |         |           |  | Amide             | N-H stretch              |

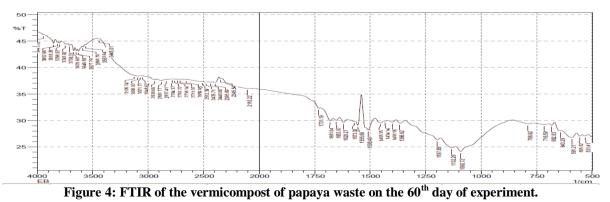


 Table 4: IR Absorption Frequencies of Functional Groups in the vermicompost of experimental papaya waste

|                         |    | ••  |  |
|-------------------------|----|-----|--|
| <b>60</b> <sup>th</sup> | da | ıy. |  |

| PEAK           531.41           555.52           591.21           582.83 | <b>INTENSITY</b><br>27.05<br>27.04  | <b>AREA</b> 14.73  | FUCTIONAL GROUPS<br>Alkyl Halide  | TYPE OF VIBRATION  |
|--|---|--|---|--|
| 555.52<br>591.21   |   | 14.73  | Alkyl Halide  | C Dr. stratal  |
| 591.21   | 27.04   |  | 7 mkyr i fullae   | C-Br stretch   |
|  |   | 14.17  | Alkyl Halide  | C-Br stretch   |
| 582.83   | 26.77   | 38.1   | Alkyl Halide  | C-Br stretch   |
| 02.05  | 29.1  | 13.37  | Alkene  | =C-H bending   |
|  |   |  | Alcohol   | C-O stretch  |
|  | 24.12   | 146.19   | Alkyl Halide  | C-F stretch  |
| 1092.72  |   |  | Amine   | C-N stretch  |
|  |   |  | Ester   | C-O stretch  |
|  |   |  | Ether   | C-O stretch  |
|  |   |  | Alcohol   | C-O stretch  |
|  |   |  | Alkyl Halide  | C-F stretch  |
| 197.85   | 26.32   | 87.58  | Amine   | C-N stretch  |
|  |   |  | Ester   | C-O stretch  |
|  |   |  | Ether   | C-O stretch  |
|  |   |  | Alkane  | C-H bending  |
| 366.62   | 29.46   | 21.85  | Alkyl Halide  | C-F stretch  |
|  |   |  | Nitro   | N-O stretch  |
|  |   |  | Alkane  | C-H bending  |
| 400.38   | 29.66   | 18.24  | Alkyl Halide  | C-F stretch  |
|  |   |  | Aromatic  | C=C stretch  |
|  |   |  | Alkane  | C-H bending  |
| 455.35   | 29.51   | 21.8   | Alkyl Halide  | C-F stretch  |
|  |   |  | Aromatic  | C=C stretch  |
|  |   |  | Aromatic  | C=C stretch  |
| 508.4  | 28.18   | 21.14  | Nitro   | N-O stretch  |
|  |   |  | Amide   | N-H bending  |
|  |   |  | Aromatic  | C=C stretch  |
| 555.66   | 29.09   | 13.82  | Nitro   | N-O stretch  |
|  |   |  | Amide   | N-H bending  |
| 620.27   | 29.62   | 17.65  | Alkene  | C=C stretch  |
|  |   |  | Garbanal  | C=O stretch  |
| CO1 04   | 20.0  | 29.41  |   | C=O stretch  |
| 681.04   | 29.9  | 28.41  |   | a, b unsaturated stretch;  |
|  |   |  | Ketone  | aryl ketone stretch  |
| 721 10   | 22.21   | 1 60 1 5   | Carbonyl  | C=O stretch  |
| /31.19   | 52.31   | 169.15   | Aldehyde  | C=O stretch  |
| 001 77   | 37.52   | 30.69  | Alkane  | C-H stretch  |
| 2881.77  |   |  | Acid  | O-H stretch  |
| 020 60   | 27.50   | 50.54  | Alkane  | C-H stretch  |
| 938.68   | 37.68   | 50.56  | Acid  | O-H stretch  |
| 646.58   | 43.21   | 7.34   | Alcohol   | O-H stretch free   |
| 670.69   |   |  | Alcohol   | O-H stretch free   |
|  | 366.62<br>400.38<br>455.35<br>508.4<br>555.66<br>520.27<br>581.04<br>731.19<br>381.77<br>938.68<br>546.58 | 366.62       29.46         400.38       29.66         455.35       29.51         508.4       28.18         555.66       29.09         520.27       29.62         581.04       29.9         731.19       32.31         381.77       37.52         938.68       37.68         546.58       43.21 | 366.62       29.46       21.85         400.38       29.66       18.24         455.35       29.51       21.8         508.4       28.18       21.14         555.66       29.09       13.82         502.27       29.62       17.65         581.04       29.9       28.41         731.19       32.31       169.15         381.77       37.52       30.69         938.68       37.68       50.56 | Ether197.85 $26.32$ $87.58$ Alcohol<br>Alkyl Halide<br>Amine<br>Ester<br>Ether366.62 $29.46$ $21.85$ Alkane<br>Alkane $400.38$ $29.66$ $18.24$ Alkyl Halide<br>Nitro $400.38$ $29.66$ $18.24$ Alkyl Halide<br>Aromatic $455.35$ $29.51$ $21.85$ Alkane $455.35$ $29.51$ $21.8$ Alkane $555.66$ $29.99$ $21.8$ Alkane $555.66$ $29.09$ $13.82$ Nitro<br>Amide $555.66$ $29.09$ $13.82$ Nitro<br>Amide $551.04$ $29.9$ $28.41$ Carbonyl<br>Alkene $581.04$ $29.9$ $28.41$ Carbonyl<br>Aldehyde $731.19$ $32.31$ $169.15$ Carbonyl<br>Aldehyde $38.68$ $37.68$ $50.56$ Alkane<br>Acid $328.68$ $37.68$ $50.56$ Alkane<br>Acid |

#### CONCLUSION

The present study concludes that the two types of fruit wastes utilized namely banana waste and papaya waste, can be degraded efficiently through vermicomposting using *Eudrilus eugeniae* than normal composting process. It is also interesting to note that papaya waste was degraded more efficiently than banana waste with higher macronutrient content. The most effective use of earthworms are organic waste management and supplement of readily available plant nutrients and demands credit as it maintains and improves soil health.

#### ACKNOWLEDGEMENT

With gratitude and respect we record our indebtedness to the Department of Zoology, Nirmala college for women, Coimbatore, for providing all the resources to carry out this research work and to Karpagam University for the FTIR analysis.

# REFERENCES

- Allevi, L, Citterio B and Ferrar. Vermicomposting of rabbit manure: modifications of microflora. Inde Bertoldi M, Ferranti MP, L' Hennite P, Zucconi F (eds.) Compost Production, Quality and Use. Elsevier, *Applied Science*, *Amsterdam*, 1987; 115-126.
- Aveyard Jim. Land degradation: Changing attitudes

   why? Journal of Soil Conservation New South Wales, 1988; 44: 46–51.
- Benitez E, Nogales R, Elvira C, Mas ciandaro G andCeccanti B. Enzyme activity as indicators of the stabilization of sewage sludge composting with *Eisenia foetida*. *Biores Technol*, 1999: 67: 297-303.
- 4. Butt K. R, Utilization of solid paper mill sludge and spent brewery yeast as a feed for soildwelling earthworms, *Bioresource Technol*, 1993: 44: 105-107.
- Dominguez J, Edwards C.A and Webster M.Vermicomposting of sewage sludge: effect of bulking materials on the growth and reproduction of the earthworm *Eisenia andrei*, *Pedobiologia*, 2000; 44: 24-32
- 6. Edwards CA. Breakdown of animal, vegetable and industrial organic wastes by earthworms In: Edwards, CA. Neuhauser E. F (eds) *Earthworms in Waste and Environmental Management SPB, The Hague, 1988: 21-31.*
- Elvira C, Dominguez J, Sampedro L and MatoS. Vermicomposting for the paper pulp industry, *Bio Cycle*, 1995: 62-63.
- 8. Elvira C, Goicoechea M, Sampdro L and Nogalas R. Bioconversion of solid paper pulp mill sludge by Earthworm, *Biores Technol*, 1996; 75: 173-177.
- Elvira C, Sampedro L, Benítez E and Nogales R. Vermicomposting of sludges from paper mill and dairy industries with *Eisena andrei*: a pilot scale study, *Bioresource Technology*, 1998; 63(3): 205–211.
- 10. Garg P, Gupta A and Satya, S. Vermicomposting of different types of waste using *Eisenia foetida*: A

comparative study, *Bioresource Tech*, 2006; 97: 391–395.

- 11. Garg V. K. and Kaushik, P. Vermi stabilization of textile mill sludge spiked with poultry droppings by epigeic earthworm *Eisenia foetida*. *Bioresource Tech*, 2005; 96: 1063–1071.
- 12. Kale R.D, Bano Kand Krishnamoorthy R.V. Potential of *Perionyx excavates* for utilization of organic wastes, *Pedibiologia*, 1982; 23: 419-425.
- 13. Nair L. D, Sar S. K, Arora A and Mahapatra, D. Fourier transfer infrared spectroscopy analysis of few medicinal plants of Chhattisgarh, India, *J. Adv. Pharm. Edu. Res.*, 2013; 3: 196–200.
- Neuhäuser M, Wirth S, Hellmann U and Bässler K. H. Utilization of *N*-acetyl-L-glutamine during longterm parenteral nutrition in growing rats: significance of glutamine for weight and nitrogen balance, *Clin. Nutr, 1988; 7: 145-150.*
- 15. Schiewer S and Iqbal M. The role of pectin in Cd binding by orange peel biosorbents: a comparison of peels, depectinated peels and pectic acid, *J. Hazard. Mater*, 2010; 177: 899–907.
- 16. Schiewer S and Patil S. B. Pectin-rich fruit wastes as biosorbents for heavy metal removal: equilibrium and kinetics, *Bioresour. Technol*, 2008: 99: 1896–1903.
- 17. Suthar S. Potential utilization of Guar gum industrial waste in vermicom post production, *Bioresour Technol*, 2006; 97: 2474–2477.
- 18. Wani S.P and Lee K.K. Biofertilizers role in upland crops production. In Fertilizers, organic manures, recyclable wastes and biofertilisers (Tandon HLS, edition). New Delhi, India: *Fertilizer Development and Consultation Organisation*, 1992; 91: 112.
- 19. Wani S.P, Rupela, O.P and Lee K.K. Sustainable agriculture in the semi-arid tropics through biological nitrogen fixation in grain legumes, *Plant Soil.* 1995; 174: 29–49.
- 20. Zou W. H, Zhao Land Zhu L. Efficient uranium (VI) biosorption on grape fruit peel: kinetic study and thermodynamic parameters, *J. Radioanal. Nucl. Chem*, 2012; 292: 1303–1315.