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A REVIEW ON HYDROGELS

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ABSTRACT

The Purpose of this study is to understand that hydrogels are a unique class of cross-linked polymeric threedimensional networks that can accommodate a significant fraction of aqueous solvents and biological fluids. Hydrogels now have attracted a growing interest from most of the scientists in various research fields. Hydrogels have played a significant role in a wide range of applications including drug delivery systems. Different methods, general benefits, general limitations, technical features, classifications, types, characterization, advantages, disadvantages, applications were studied.

INTRODUCTION

The term hydrogel describes a three-dimensional crosslinked polymeric network obtained from synthetic or natural polymers which has capacity to hold water within its porous structure.^[1,2] The water holding capacity of the hydrogels arise mainly due to the presence of hydrophilic groups, viz. amino, carboxyl and hydroxyl groups, in the polymer chains. These polymeric materials do not dissolve in water at physiological temperature and pH but swell considerably in an aqueous medium Hydrogels can be made from virtually any water-soluble polymer, encompassing a wide range of chemical compositions and bulk physical properties. Further- more, hydrogels can be formulated in a variety of physical forms informs including microparticles, nanoparticles, coatings, and films.^[3] Hydrogels have been widely used as a drug carrier due to its ease in manufacturing and selfapplication in clinical and fundamental applications. Applications of hydrogels in the biomedical field include contact lenses, artificial corneas, wound dressing, coating for sutures, catheters, and electrode sensors.^[4]

GENERAL BENEFITS OF HYDROGELS^[5,6]

- Biocompatible
- Can be injected in vivo (in a whole, living organism) as a liquid that then gels at body temperature
- Protect cells
- Good transport properties (such as nutrients to cells or cell products from cells)
- Timed release of medicines or nutrients
- Easy to modify
- Can be biodegradable or bioabsorbable

GENERAL LIMITATIONS OF HYDROGELS^[7,10]

- High cost
- Low mechanical strength
- Can be hard to handle
- Difficult to load with drugs/nutrients
- May be difficult to sterilize
- Non-adherent.

HYDROGEL TECHNICAL FEACTURES^[11]

The functional features of an ideal hydrogel material can be listed as follows:

- The highest absorption capacity (maximum equilibrium swelling) in saline.
- The desired rate of absorption (preferred particle size and porosity) depending on the application requirement.
- The highest absorbency under load (AUL).
- The lowest soluble content and residual monomer.
- The lowest price.

CLASSIFICATION OF HYDROGEL^[12]

Hydrogels can be classified into two groups based on their natural or synthetic origins. Classification according to polymeric composition, the method of preparation leads to formations of some important classes of hydrogels:

- Homopolymeric hydrogels
- Copolymeric hydrogels
- Multipolymer interpenetrating polymeric hydrogel (IPN)



Fig. 1: Classification of Hydrogels.

TYPES OF HYDROGELS^[13]

Depending on the type of polymer, hydrogels are divided into two categories: natural and synthetic hydrogels. Hydrogenated by natural or synthetic polymers are considered as raw materials for medical applications. Natural and synthetic polymers used to make hydrogels must be biocompatible, biodegradable, and in some applications where the hydrogel is in contact with the blood must be blood compatible.

- Natural Hydrogels
- Synthetic Hydrogels

METHOD FOR PREPARATION OF HYDROGEL^[14]

In general, hydrogels can be prepared from either synthetic polymers or natural polymers. The synthetic polymers are hydrophobic in nature and chemically stronger compared to natural polymers. Their mechanical strength results in slow degradation rate, but on the other hand, mechanical strength provides the durability as well. These two opposite properties should be balanced through an optimal design. Also, it can be applied to the preparation of hydrogels based on natural polymers provided that these polymers have suitable functional groups or have been functionalized with radically polymerizable groups. In the most succinct sense, a hydrogel is simply a hydrophilic polymeric network cross-linked in some fashion to produce an elastic structure. Thus, any technique which can be used to create a cross-linked polymer can be used to produce a hydrogel. Copolymerization/cross-linking free-radical poly-merizations are commonly used to produce hydrogels by reacting hydrophilic monomers with multifunctional cross-linkers. Water-soluble linear polymers of both natural and synthetic origin are crosslinked to form hydrogels in a number of ways:

- 1. Linking polymer chains via chemical reaction.
- 2. Using ionizing radiation to generate main-chain free radicals which can recombine as cross-link junctions.
- 3. Physical interactions such as entanglements, electrostatics and crystallite formation.

CHARACTERIZATIZATION HYDROGELS^[15,16]

- Morphological characterization
- Fourier Transform Infrared Spectroscopy(FTIR)
- Swelling Studies of IPN hydrogel

APPLICATIONS^[17,24]

- Drug delivery in the oral cavity
- Drug delivery in the gi tract
- Rectal delivery
- Ocular delivery
- Wound healing
- Hydrogels for transdermal drug delivery

ADVANTAGES OF HYDROGELS^[25,27]

- Their degree of flexibility extremely similar to natural tissue which is due to their remarkable water content.
- The environmentally sensitive hydrogels can sense the changes in temperature, pH or the metabolite concentration and sensing these changes they release the load.
- They are biocompatible, biodegradable and are also injectable.
- Entrapment of microbial cells with in hydrogel beads has the advantage of low toxicity.
- Timed release of growth factors and other nutrients to ensure proper tissue growth.
- Hydrogels have good transport properties.

DISADVANTAGES OF HYDROGELS^[25,26]

- They are high in cost and possess low mechanical strength.
- They are non-adherent and may need to be secured by secondary dressing.
- They may cause a sensation similar to that which is felt by movement of maggots.
- They can be hard to handle.
- Difficulty to load with drugs/nutrients can be experienced.

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• Hydrogels used as contact lenses causes lens deposition, hypoxia, dehydration and red eye reactions.

CONCLUSION

Hydrogels are polymer crosslinked networks that absorb significant amounts of aqueous solutions. The hydrogel are more resemble natural living tissue than any other type of synthetic biomaterial because of their highwater content. Hydrogels used in drug delivery in the oral cavity, drug delivery in the gi tract, rectal delivery, ocular delivery, wound healing, hydrogels for transdermal drug delivery. Different methods, general benefits, general limitations, technical features, classifications, types, characterization, advantages, disadvantages, applications were studied.

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