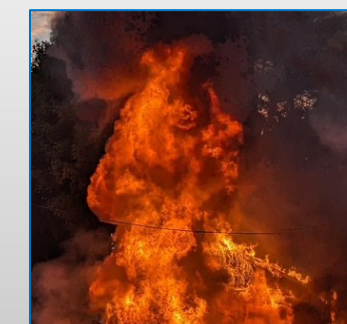


Pyrolysis and GC×GC-MS. A hot topic!

MDCW 2025, Liège



Robert (Chip) Cody



AccuTOF GC-Alpha in our lab

JEOL accessories

- EI/CI/PI combination ion source
- EI/FI/FD combination ion source
- Direct insertion probe
- Escrime quantitative analysis software
- msFineAnalysis AI qualitative analysis software

3rd-party

- Agilent 8890 GC
- Frontier Lab Multi-shot pyrolyzer
- HTA 2800T autosampler (liquid, headspace, SPME)
- SepSolve thermal modulator for GCxGC-MS





Software for GCxGC-MS data analysis in our lab

- **ChromSpace (SepSolve)**

- Primarily used for controlling our SepSolve Insight thermal modulator
- Looking forward to learning more about capabilities for data analysis

- **AnalyzerPro XD (SpectralWorks)**

- All-in-one software support for GC-MS, GCxGC-MS, and ambient ionization
- Open Access through RemoteAnalyzer software
- Peak detection incorporates deconvolution automatically
- Best way to compare multiple batches of samples to find differences

- **GC Image (GC Image)**

- Full-featured GCxGC software
- Supports accurate-mass data
- I like the graphic functions for making figures!

- **msFineAnalysis AI (JEOL)**

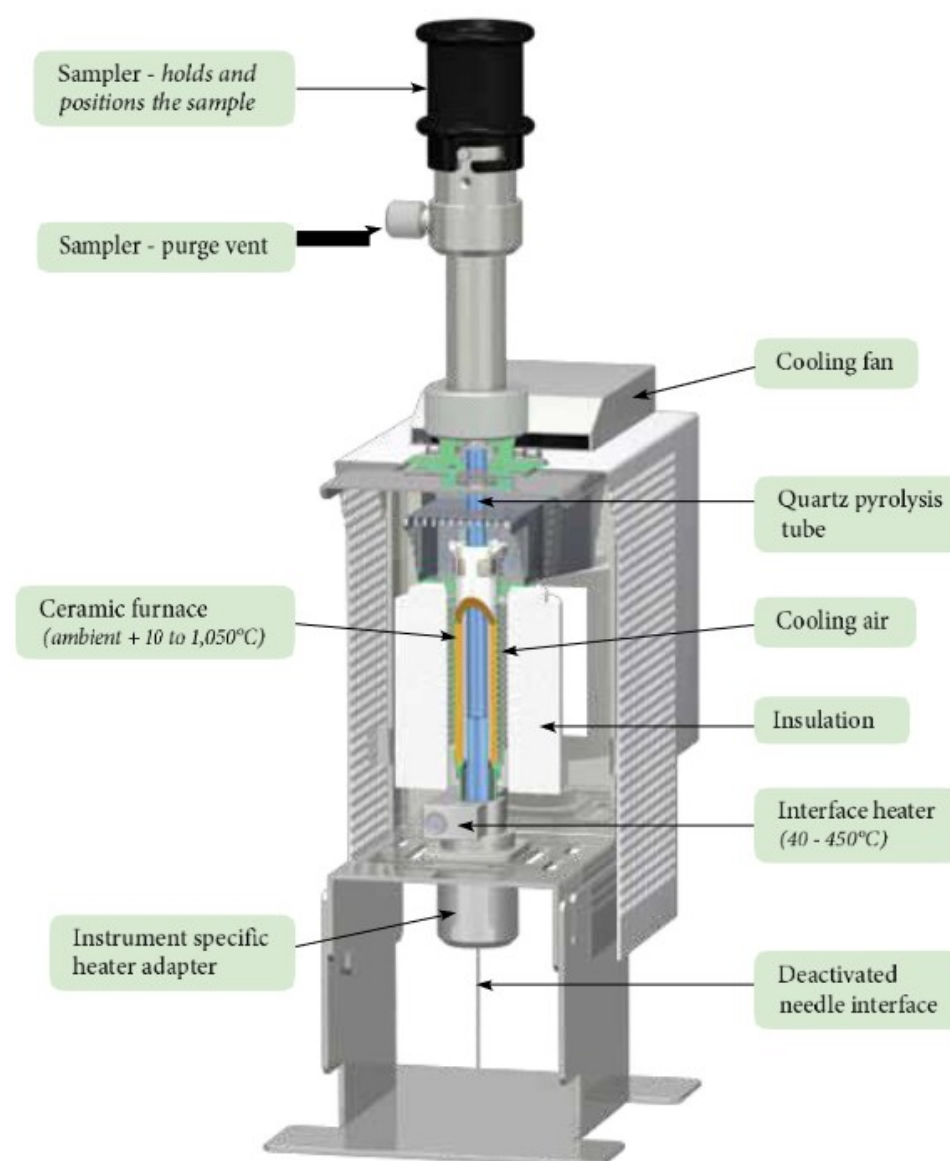
- JEOL software for qualitative GC-MS analysis
- Uses all available information from GC-MS measurements (EI, accurate mass, RI, and soft ionization)
- **Unknown analysis with AI support and searchable AI-generated 130-million compound database**
- **Just updated with GCxGC support.**

**Silver Sponsors
of this workshop**



Sample introduction: Frontier Lab Multi-Shot Pyrolyzer (EGA/PY-3030D)

Pyrolysis is the key tool for qualitative polymer analysis by GC-MS and GCxGC-MS



- Furnace-based system
- Thermal desorption
- Pyrolysis
- Evolved gas analysis

"Biodegradable" plastic bag purchased online

Inconsistent claims!

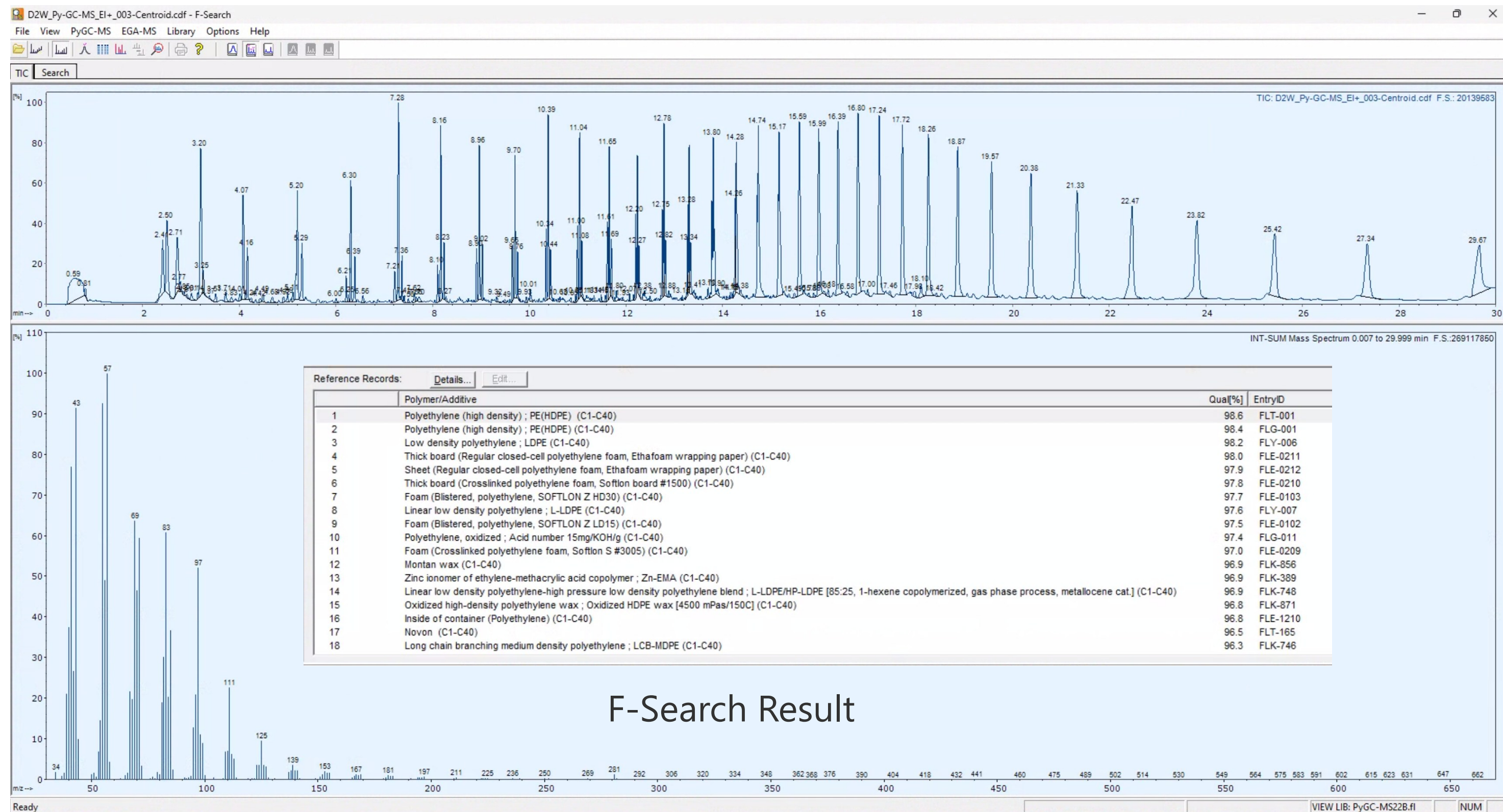


- Product description says that the bags are made from polylactic acid (PLA)
- Product description refers to ASTM standard D5511 "Biodegradation of Plastic Materials Under High-Solids Anaerobic Digestion Conditions"
- Label refers to a different ASTM standard: D6954-01 "Standard Guide for Exposing and Testing Plastics that Degrade in the Environment by a Combination of Oxidation and Biodegradation"

About this item

- Natural Materials: Our bags are constructed from PLA, which is made of 100% plant based resources like cornstarch. This also means our t shirt bag is BPA-free.
- Certified: Tested and proven to comply with the ASTM D5511 requirements for composting. Commercially compostable only*.

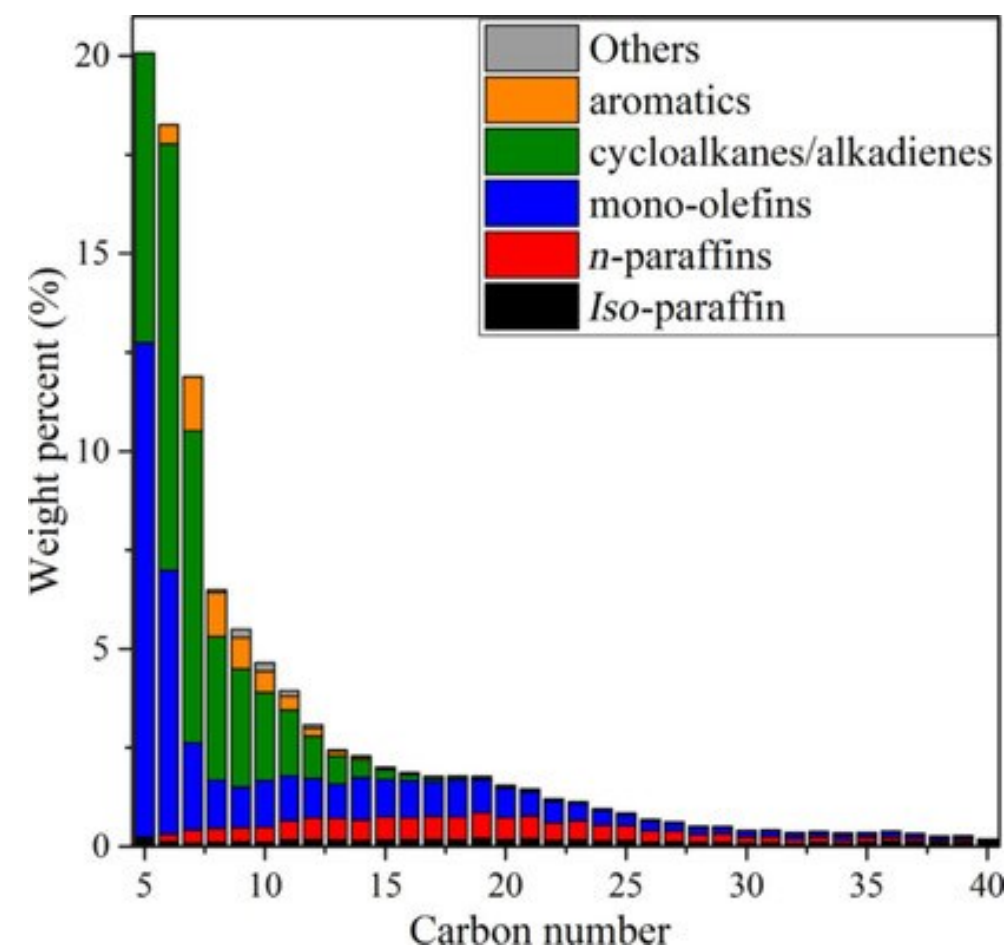
The bags are not PLA: they're polyethylene!





Oxidation and degradation of polyethylene?

Degradation and pyrolysis of polyethylene is a topic of interest for recycling plastics.



GC x GC-HRTOFMS can

- Monitor pyrolysis reactions on a small scale
- Characterize the products of pilot-plant pyrolysis

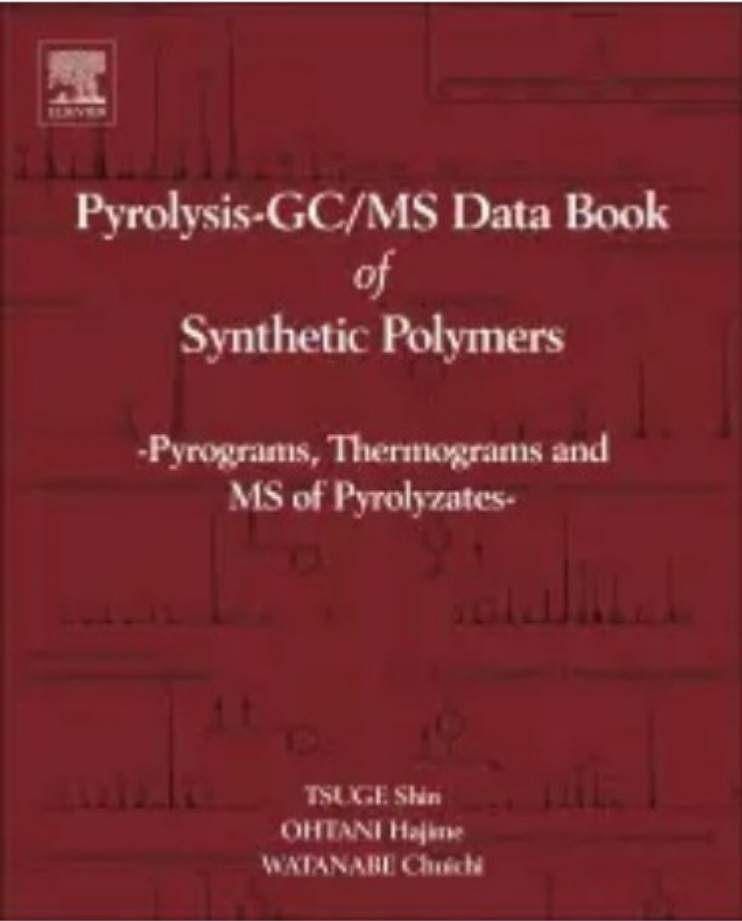
Important tools for this application:

- High-resolution MS
- Soft ionization (FI, PI, CI)

Zhao, D., Wang, X., Miller, J.B., Huber, G.W.: The Chemistry and Kinetics of Polyethylene Pyrolysis: A Process to Produce Fuels and Chemicals. *ChemSusChem*. **13**, 1764-1774 (2020)



Let's look deeper at pyrolysis GC-MS for polyethylene

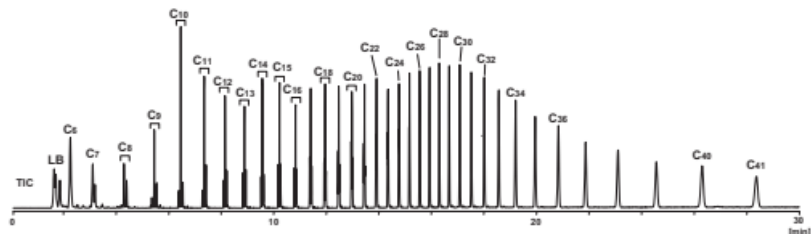
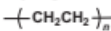


ISBN: 978-0-444-53892-5

2.2 DATA COMPILATION OF PYROGRAMS, THERMOGRAMS AND MS DATA OF MAJOR PYROLYZATES FOR 163 TYPICAL POLYMER SAMPLES

2.2.1 Polyolefins (homopolymers)

001 Polyethylene (high density); PE(HDPE)



Peak Notation	Assignment of Main Peaks	Molecular Weight	Retention Index	Relative Intensity
LB	propylene + propane	42; 44	300	43.7
C6	CH2=CH(CH2)3CH3	84	583	91.8
C7	CH2=CH(CH2)4CH3	98	689	42.4
	CH3(CH2)5CH3	100	700	19.3
C8	CH2=CH(CH2)4CH=CH2	110	782	2.1
	CH2=CH(CH2)5CH3	112	791	25.1
	CH3(CH2)6CH3	114	800	14.3
C9	CH2=CH(CH2)5CH=CH2	124	883	5.8
	CH2=CH(CH2)6CH3	126	892	30.4
	CH3(CH2)7CH3	128	900	10.3
C10	CH2=CH(CH2)6CH=CH2	138	983	6.6
	CH2=CH(CH2)7CH3	140	991	64.2
	CH3(CH2)8CH3	142	1000	10.4
C11	CH2=CH(CH2)7CH=CH2	152	1083	7.1
	CH2=CH(CH2)8CH3	154	1092	49.8
	CH3(CH2)9CH3	156	1100	16.1
C14	CH2=CH(CH2)10CH=CH2	194	1385	12.3
	CH2=CH(CH2)11CH3	196	1392	49.2
	CH3(CH2)12CH3	198	1400	13.5
C20	CH2=CH(CH2)16CH=CH2	278	1985	25.3
	CH2=CH(CH2)17CH3	280	1993	38.0
	CH3(CH2)18CH3	282	2000	16.2
C30	CH2=CH(CH2)27CH3	420	2993	100.0
C40	CH2=CH(CH2)37CH3	560	3997	94.1
C41	CH2=CH(CH2)38CH3	574	4096	82.8

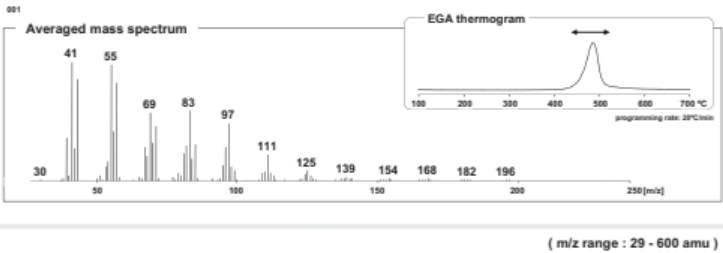
[Related References]

- 1) Michajlov, L.; Zugenmaier, P.; Cantow, H.-J. *Polymer* 1968, 9, 325.
- 2) Sugimura, Y.; Tsuge, S. *Anal. Chem.* 1978, 50, 1968.
- 3) Sugimura, Y.; Tsuge, S. *Macromolecules* 1979, 12, 512.
- 4) Ohtani, H.; Tsuge, S.; Usami, T. *Macromolecules* 1984, 17, 2557.
- 5) Duc, S.; Lopez, N. *Polymer* 1999, 40, 6723.

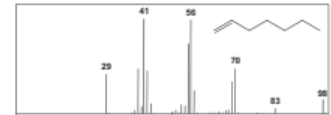
This book belongs to Robert Cody (cody@jeol.com)

Pyrograms and Thermograms of 163 High Polymers, and MS Data of the Major Pyrolyzates

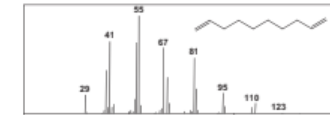
13



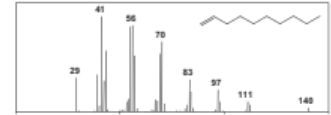
C7 : 1-heptene



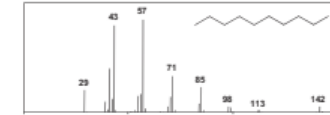
C10 : 1,9-decadiene



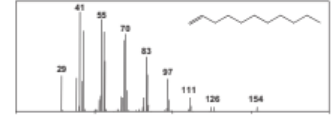
C10 : 1-decene



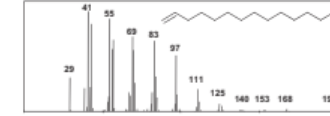
C10 : n-decane



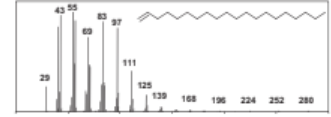
C11 : 1-undecene



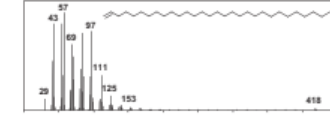
C14 : 1-tetradecene



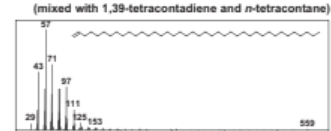
C20 : 1-eicosene



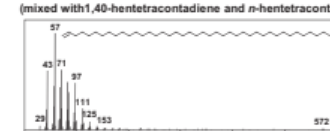
C30 : 1-triacontene



C40 : 1-tetracontene



C41 : 1-hentetracontene



Copyright Elsevier 2023

This book belongs to Robert Cody (cody@jeol.com)

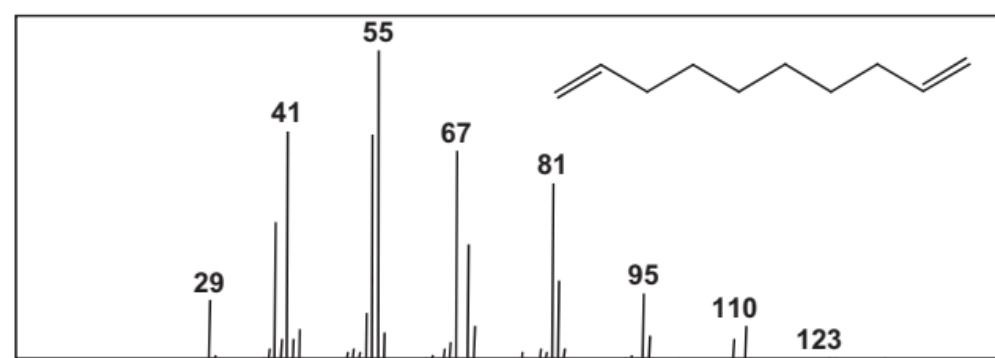
Copyright Elsevier 2023

Here's the reference book entry for pyrolysis 1D GC-MS of polyethylene

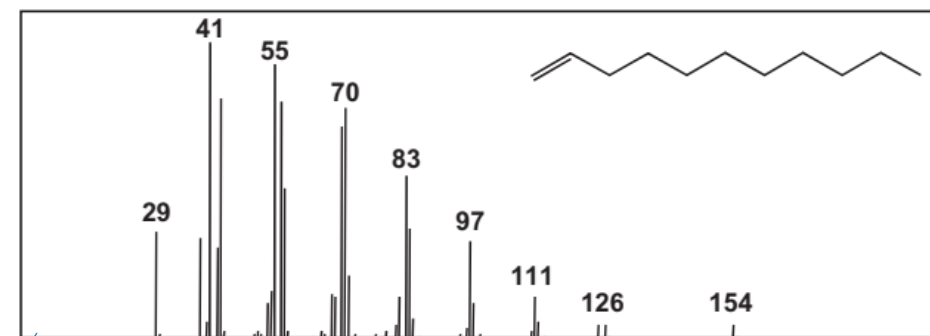


Py HDPE chromatogram from the Pyrolysis GC/MS Data Book

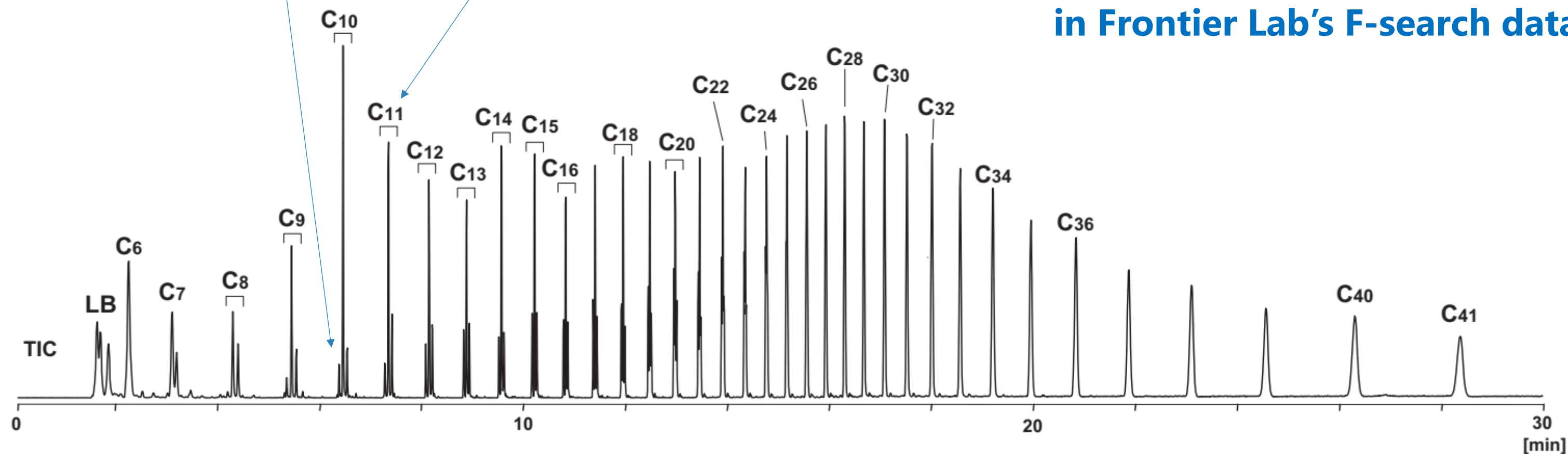
C10 : 1,9-decadiene



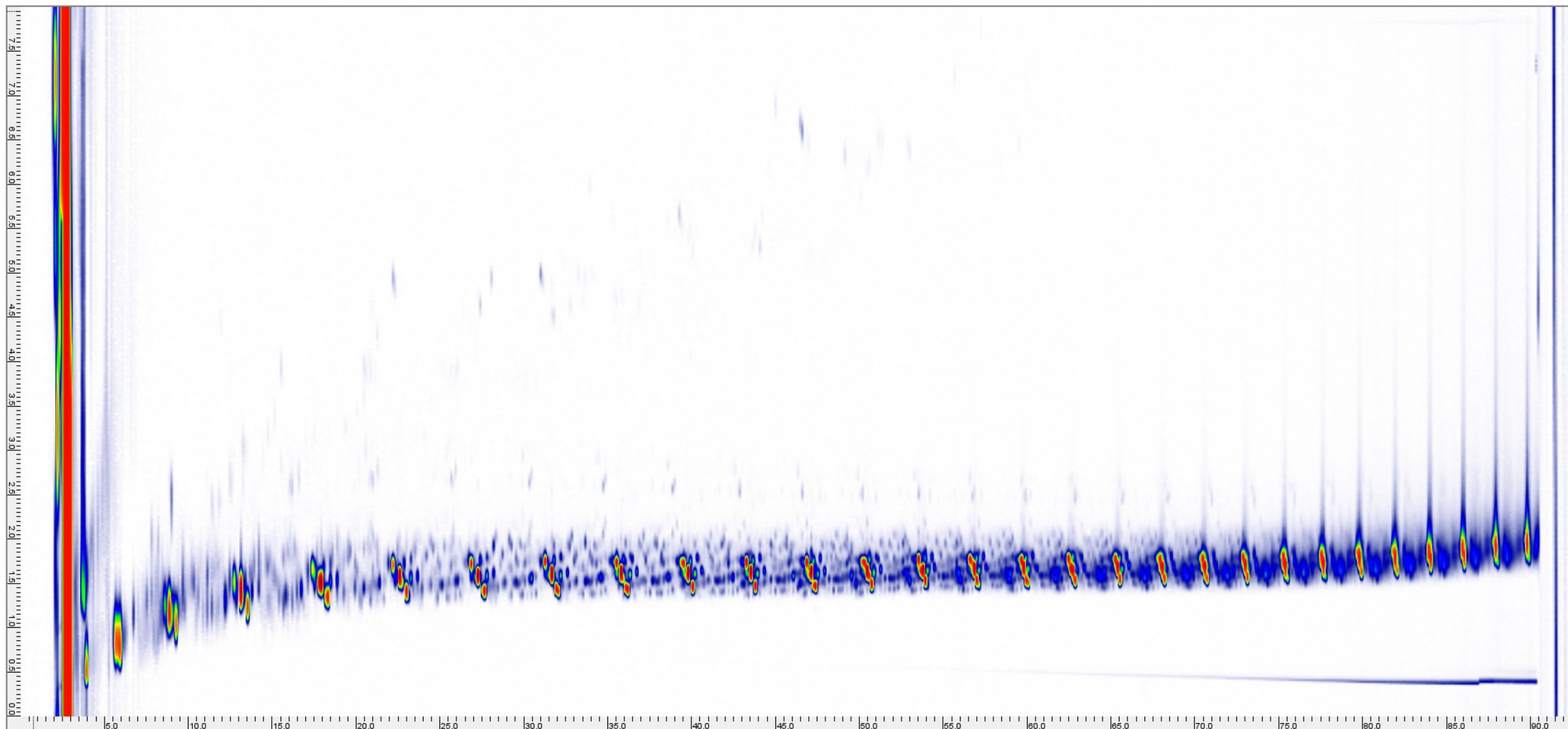
C11 : 1-undecene



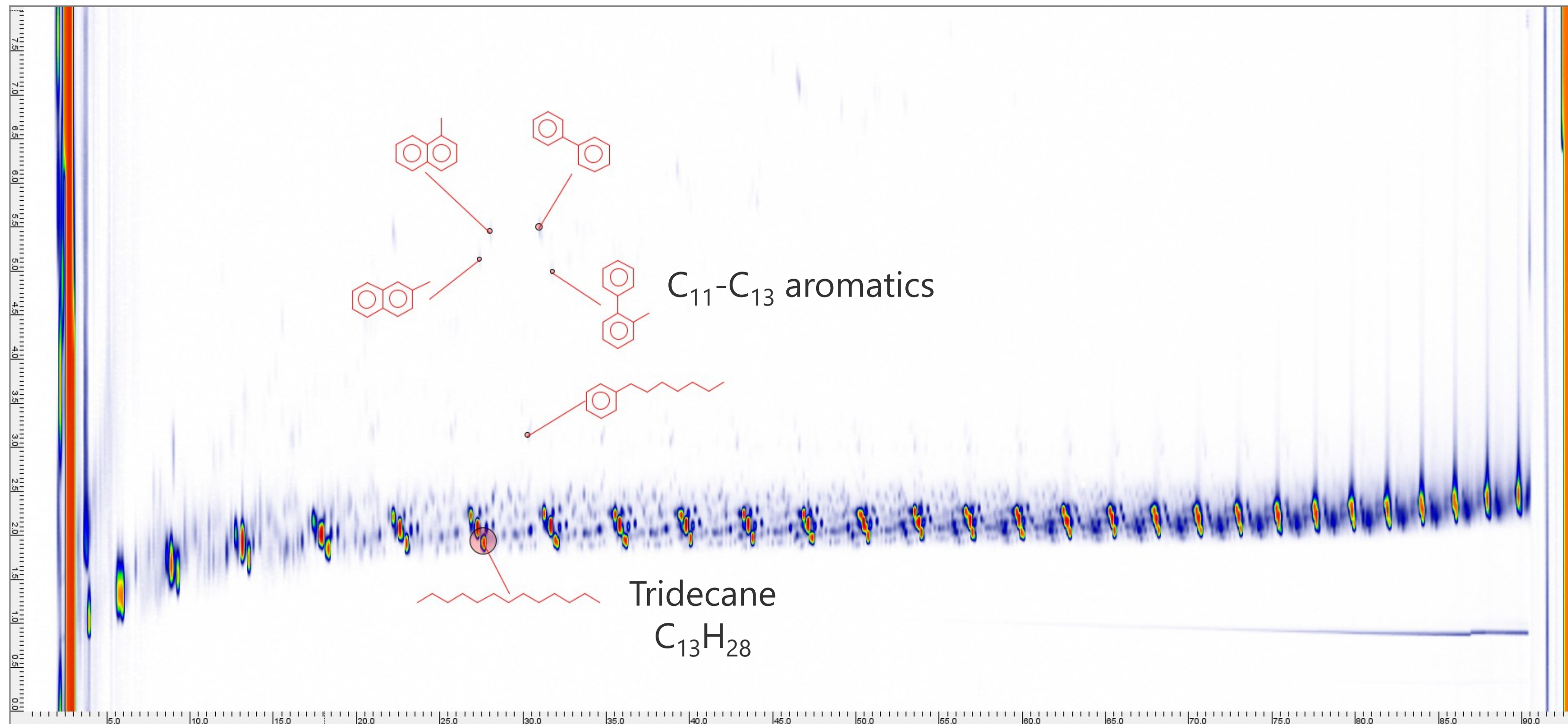
**Major compounds are identified
in Frontier Lab's F-search database**



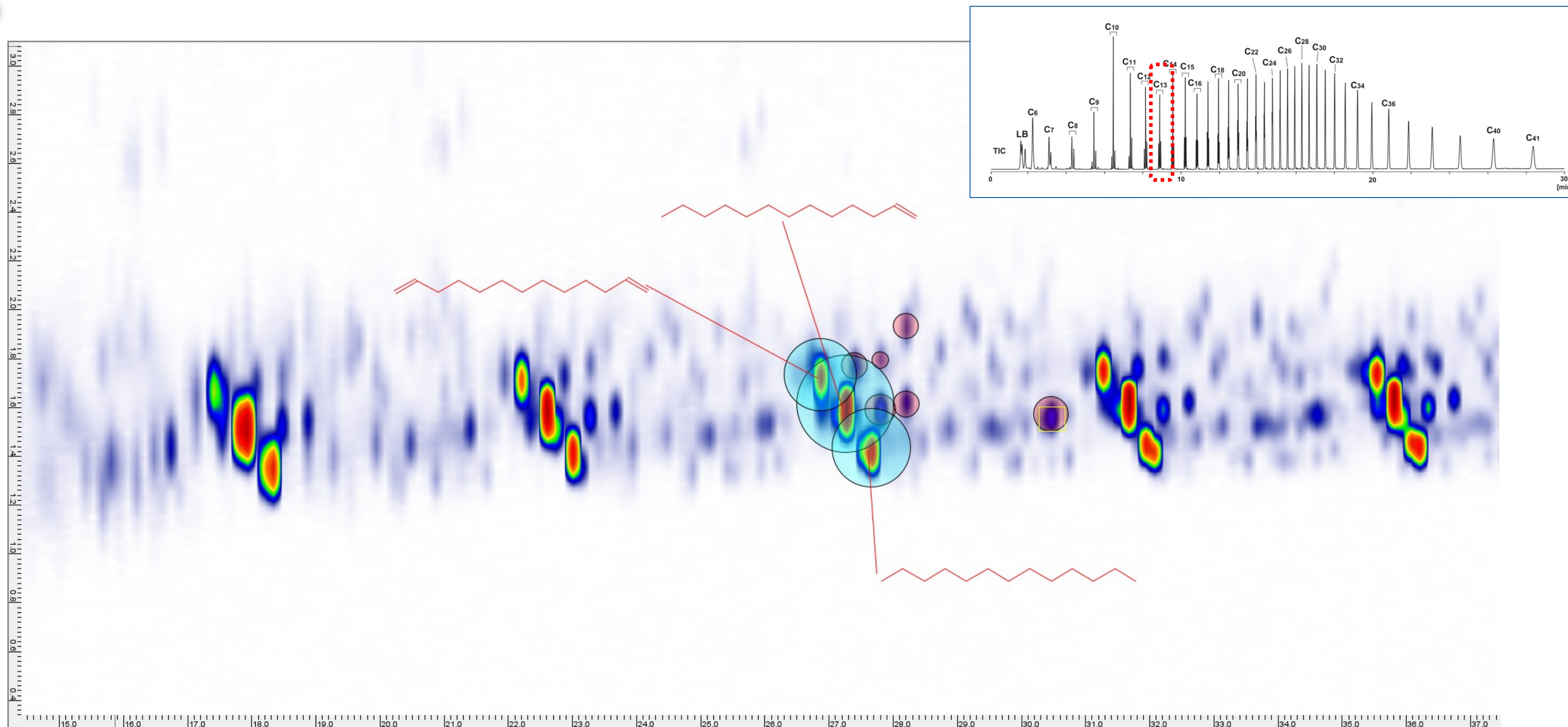
Pyrolysis GCxGC-TOFMS for LDPE shows many more compounds



Assignments for selected blobs

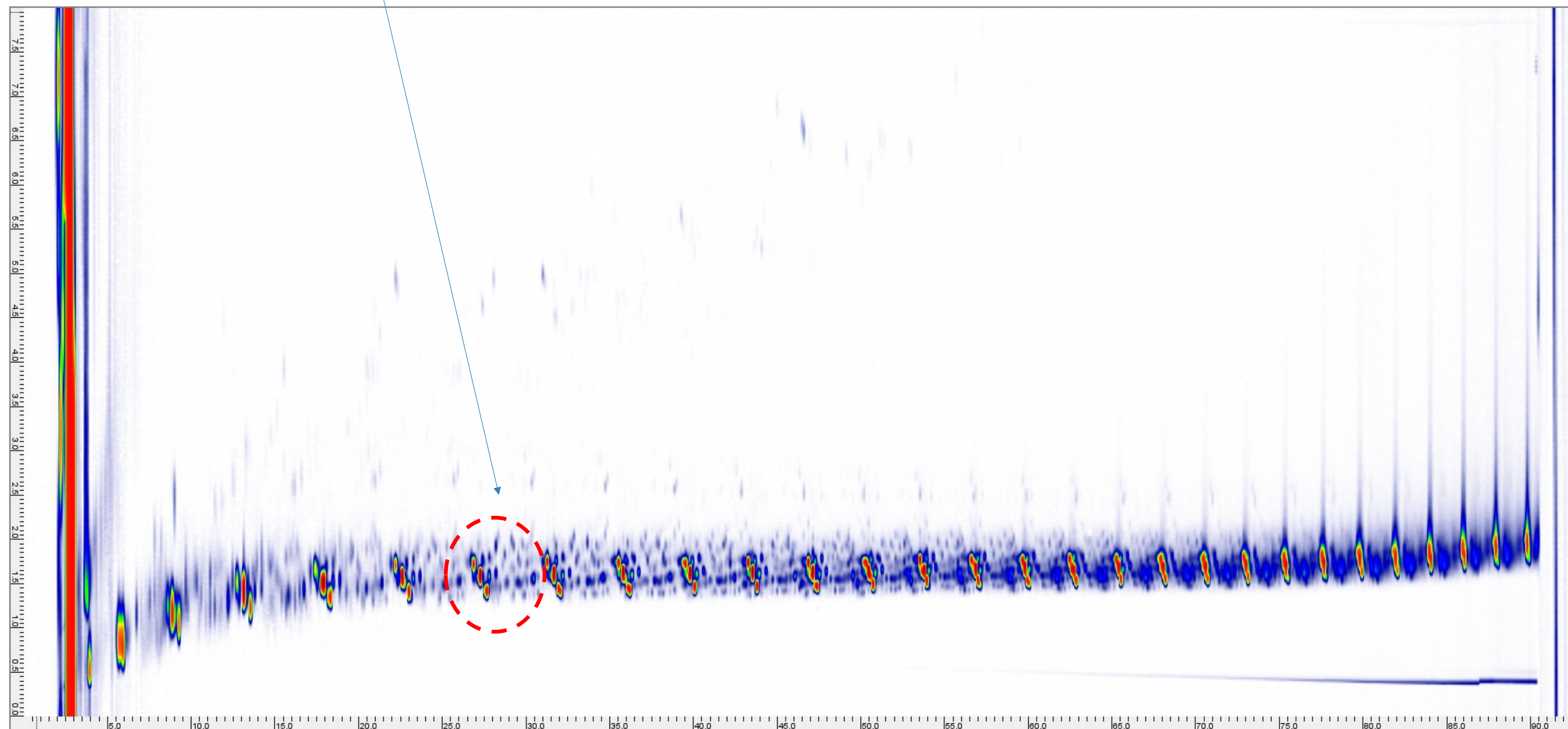


Tridecane, 1-tridecane, and 1,12-tridecadiene



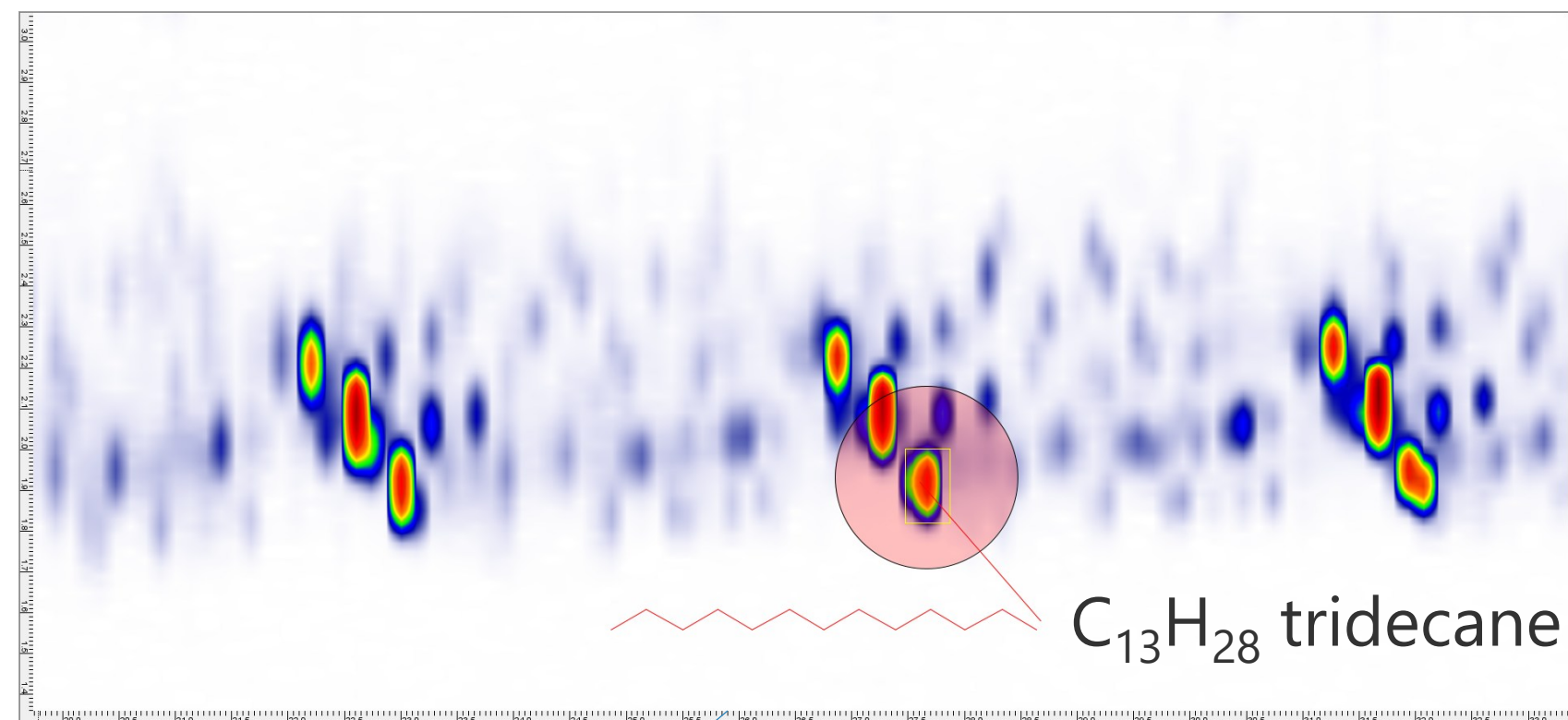


Zoom in on this region

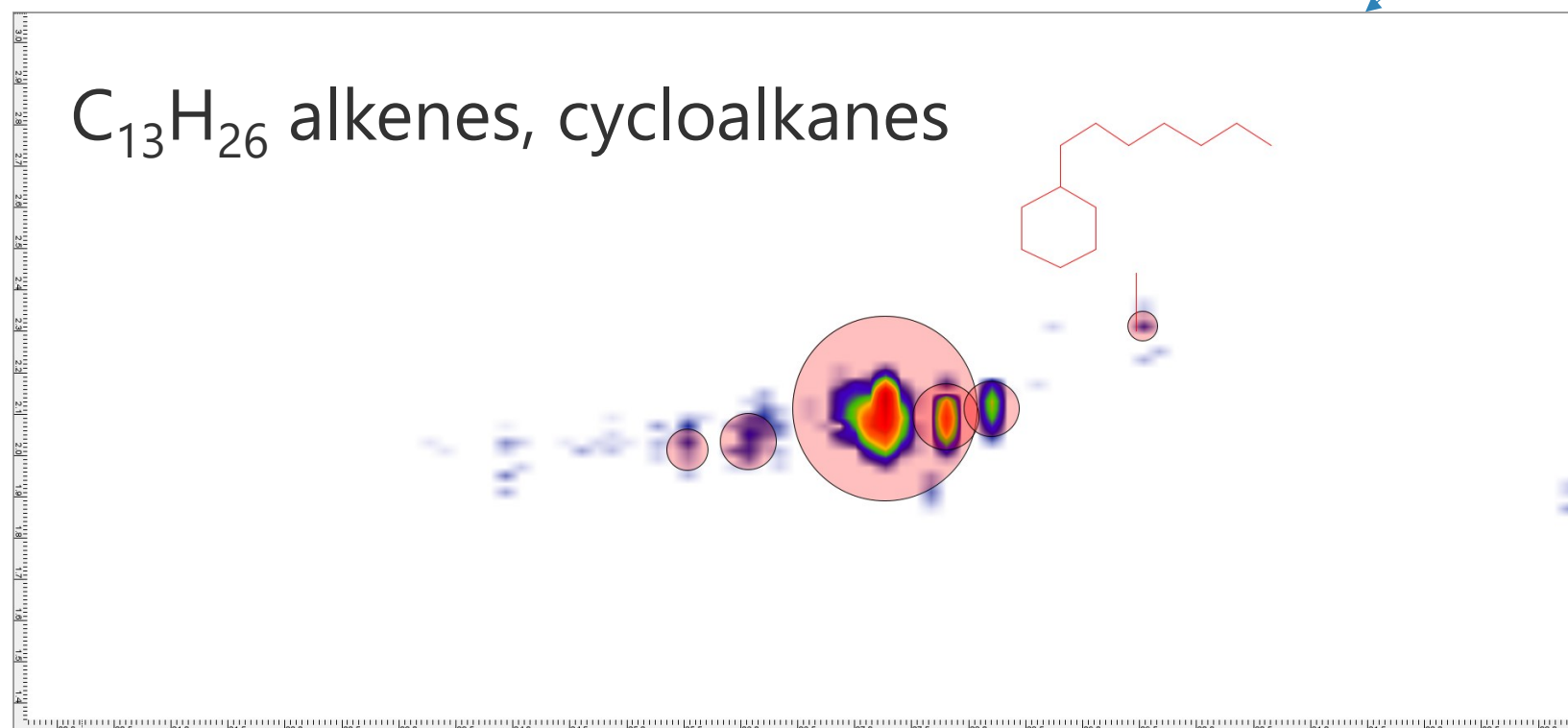




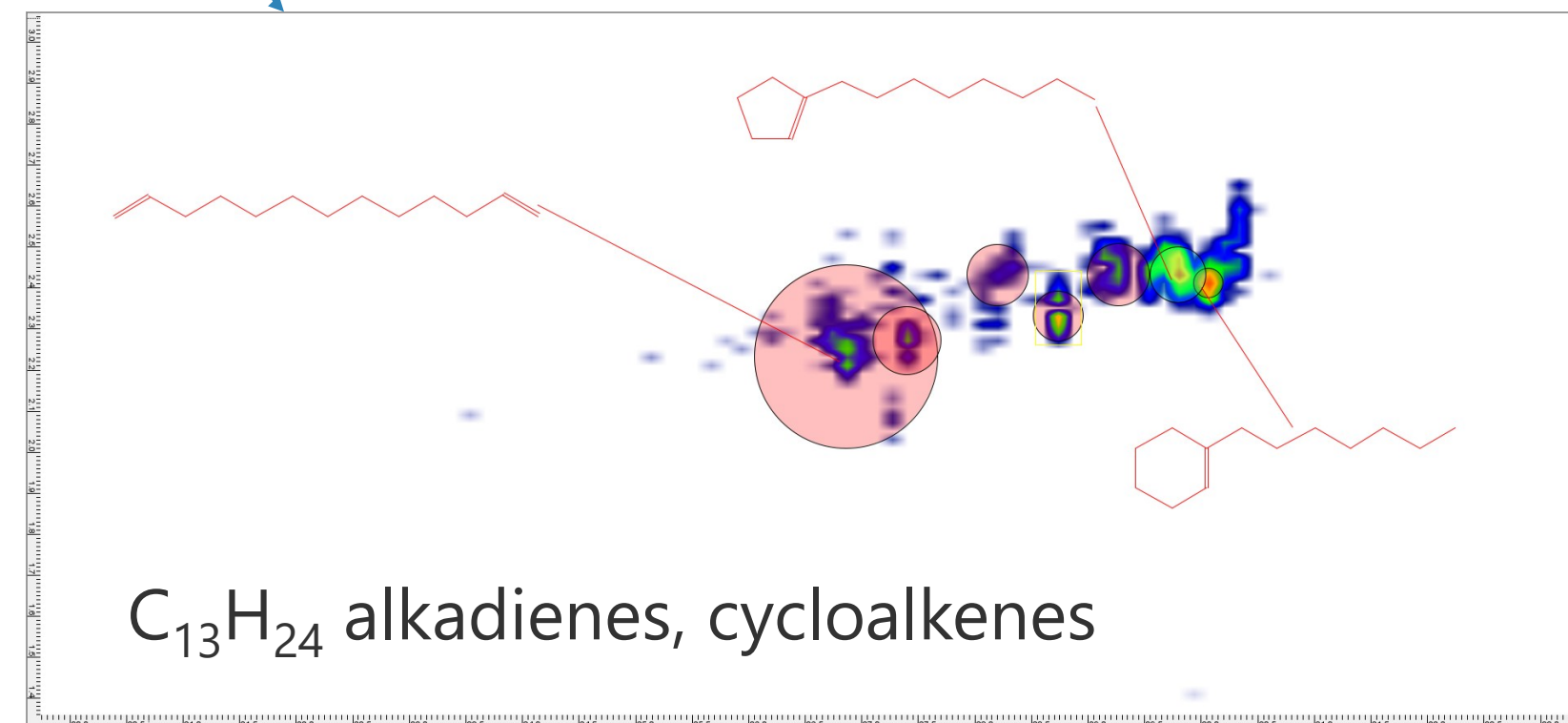
$C_{13}H_{2n-x}$ components in this region



$C_{13}H_{28}$ tridecane



$C_{13}H_{26}$ alkenes, cycloalkanes

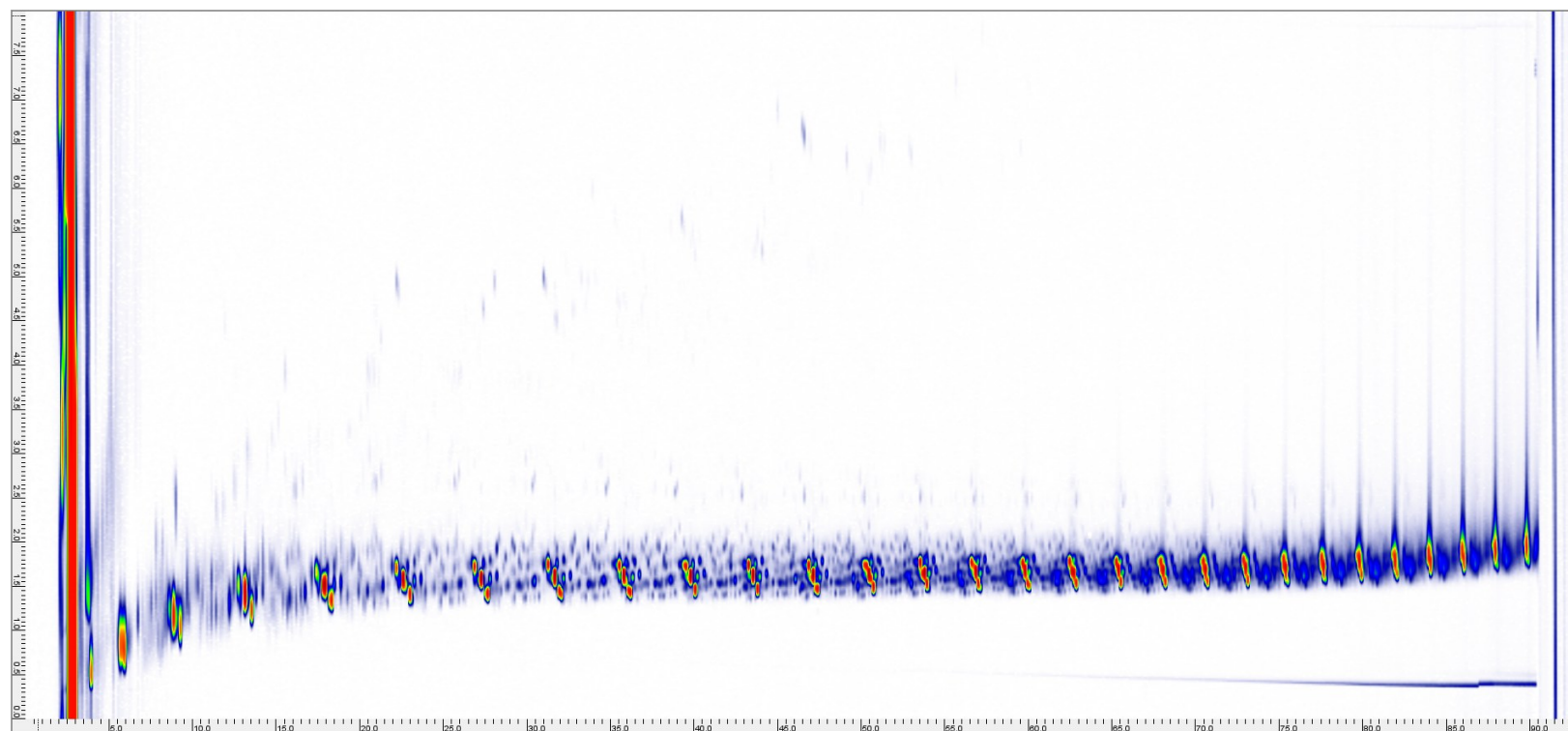


$C_{13}H_{24}$ alkadienes, cycloalkenes

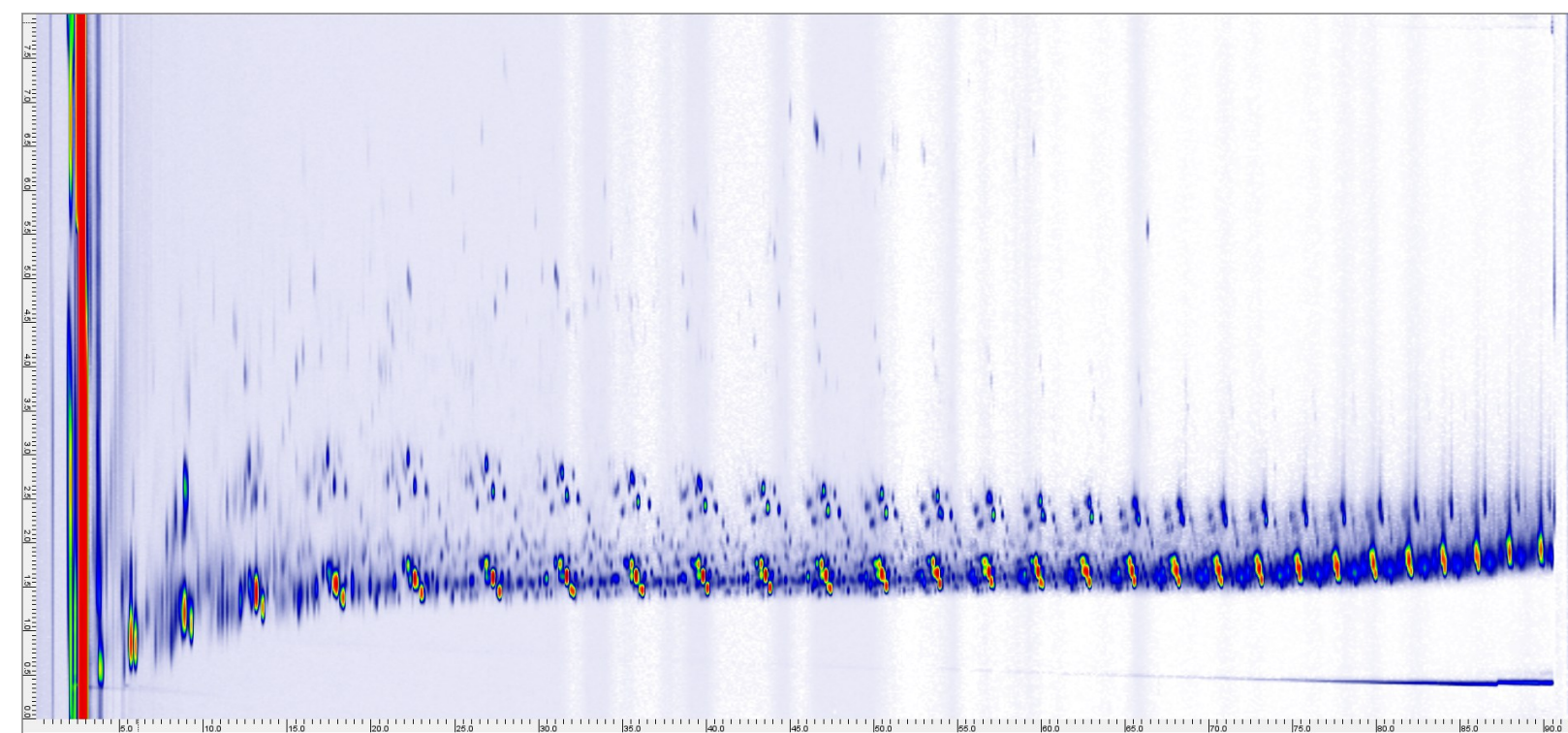


How does oxidation change LDPE?

Low-density polyethylene (LDPE)

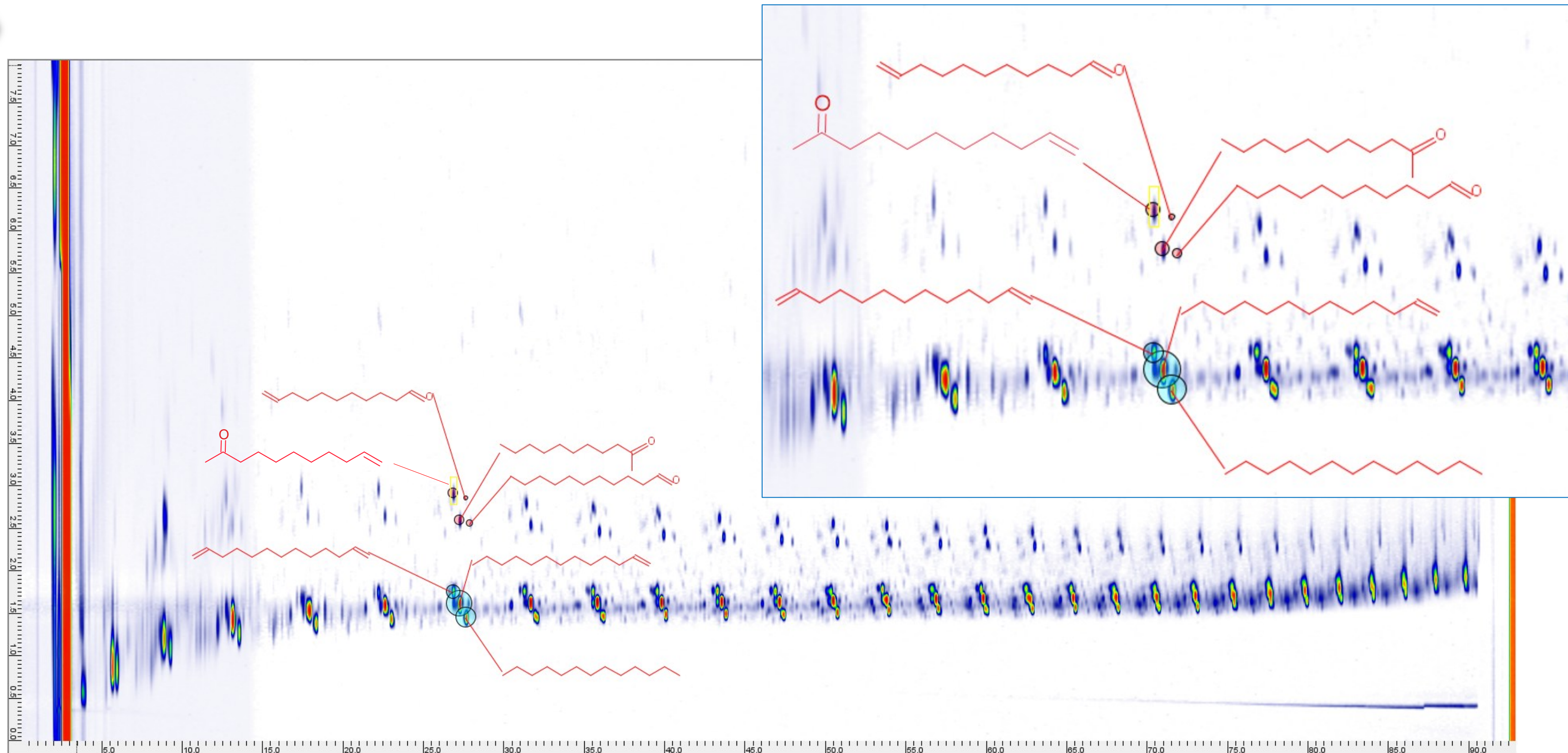


Polyethylene (oxidized)

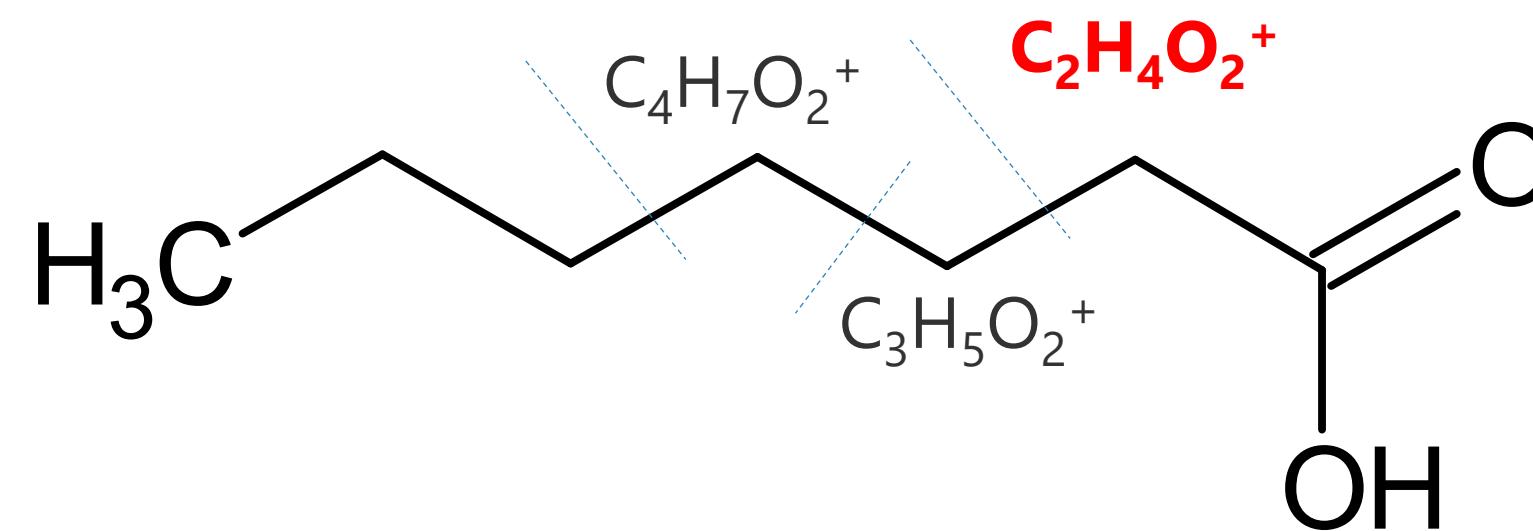
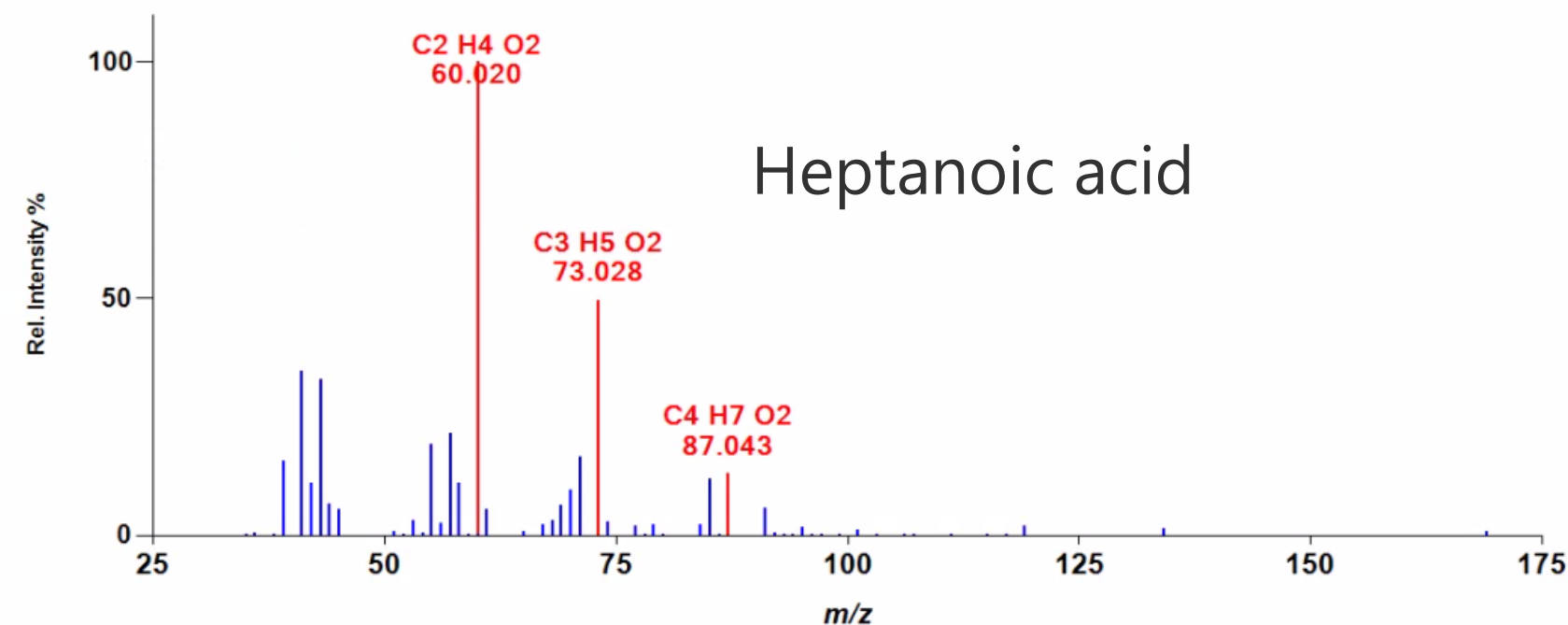


Scientific Polymer Products
Acid number 16 mg KOH/g

Oxidized PE: ketones and aldehydes

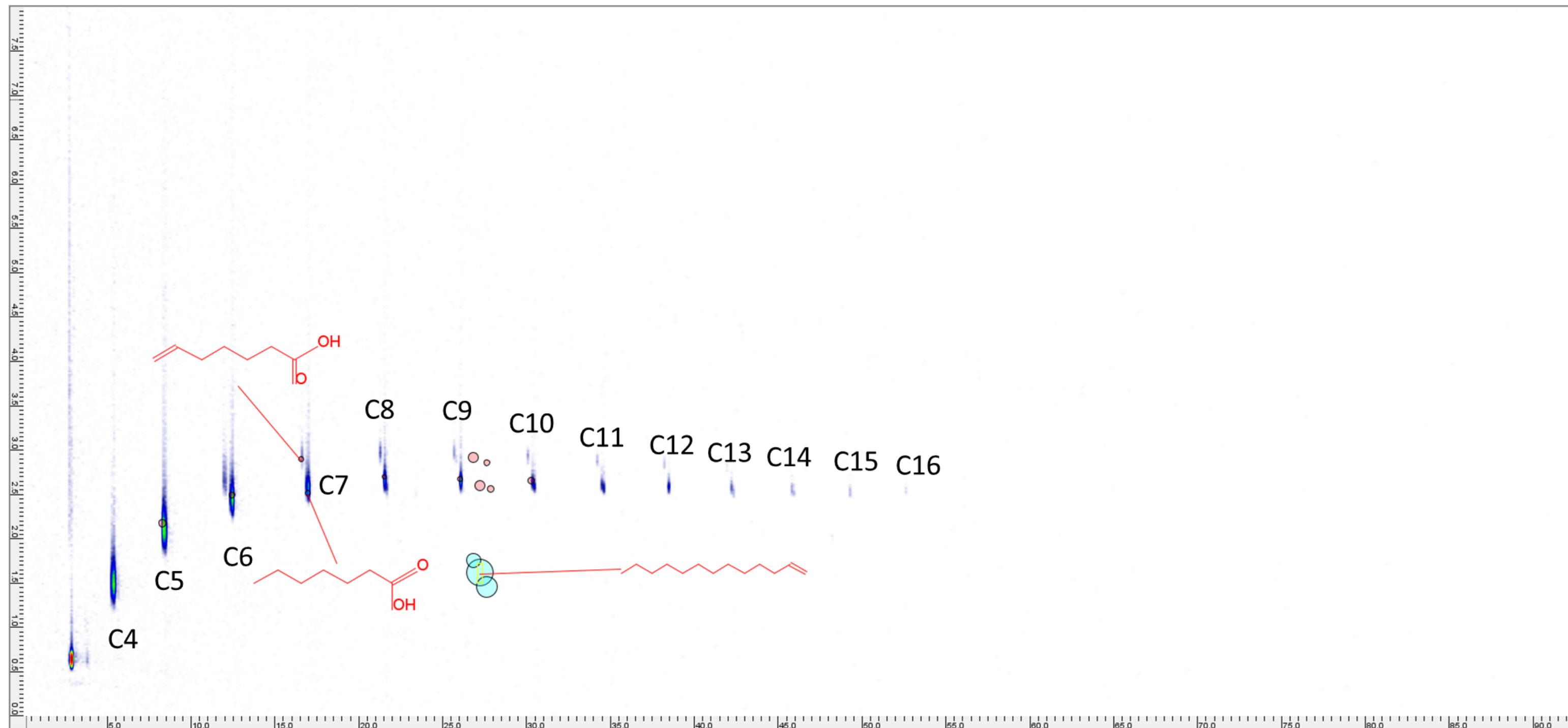


Oxidized PE waxes are characterized by their acid number



- The acid number relates to how oxidized PE waxes can form emulsions for coating agents and/or perform as lubricants.
- Determined by titration: This sample has an acid number of 16 mg KOH/g
- Carboxylic acids in the GCxGC-MS data are minor peaks in the TICC, but they can be identified by creating SICC's for the characteristic fragments in their EI mass spectra.
- Electron-capture negative-ion mass spectra would also identify the acids

Carboxylic acids in the oxidized PE: SICC for $C_2H_4O_2^+$ fragment





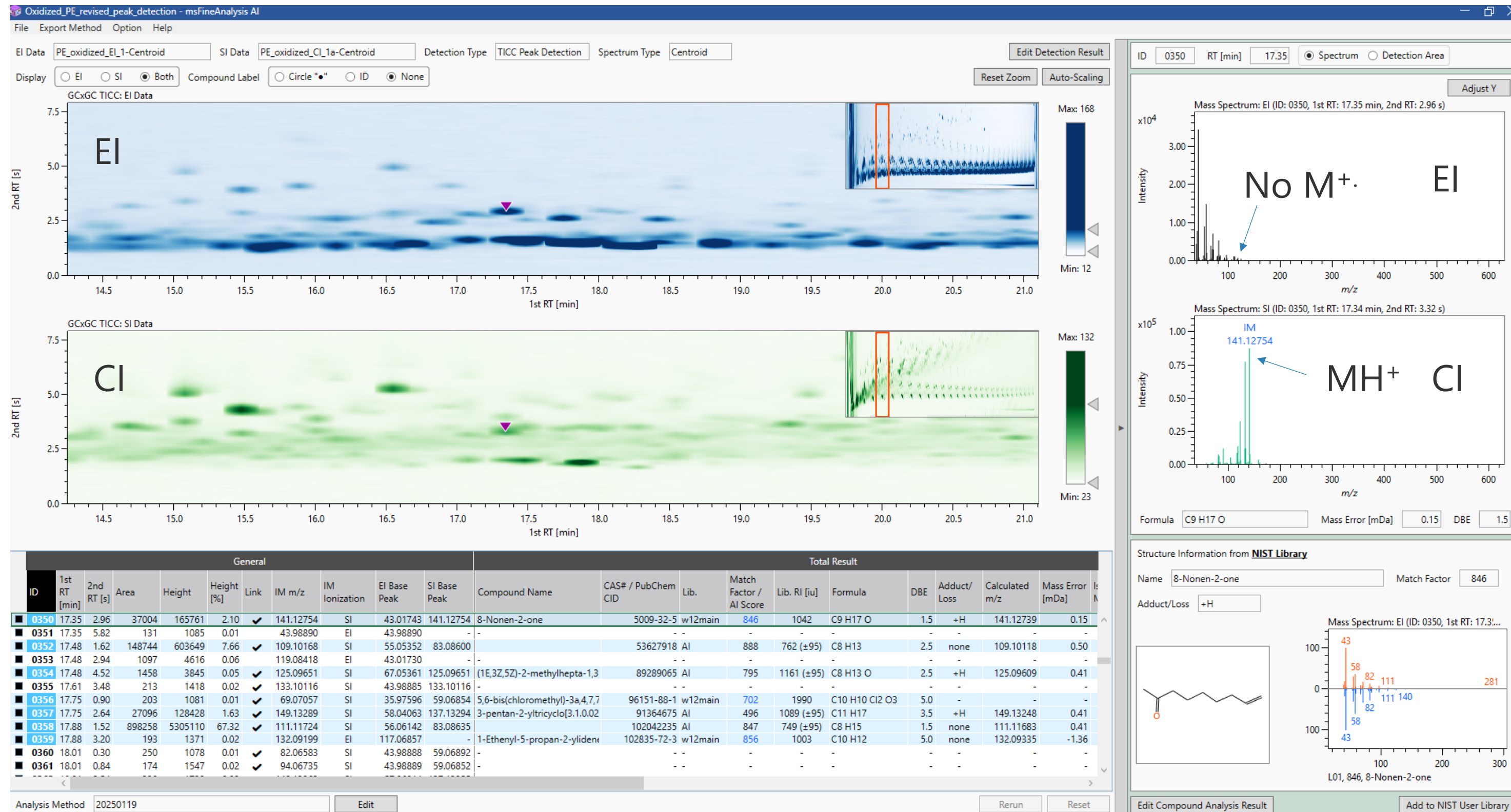
New: msFineAnalysis AI adds support for GCxGC data

Oxidized Polyethylene



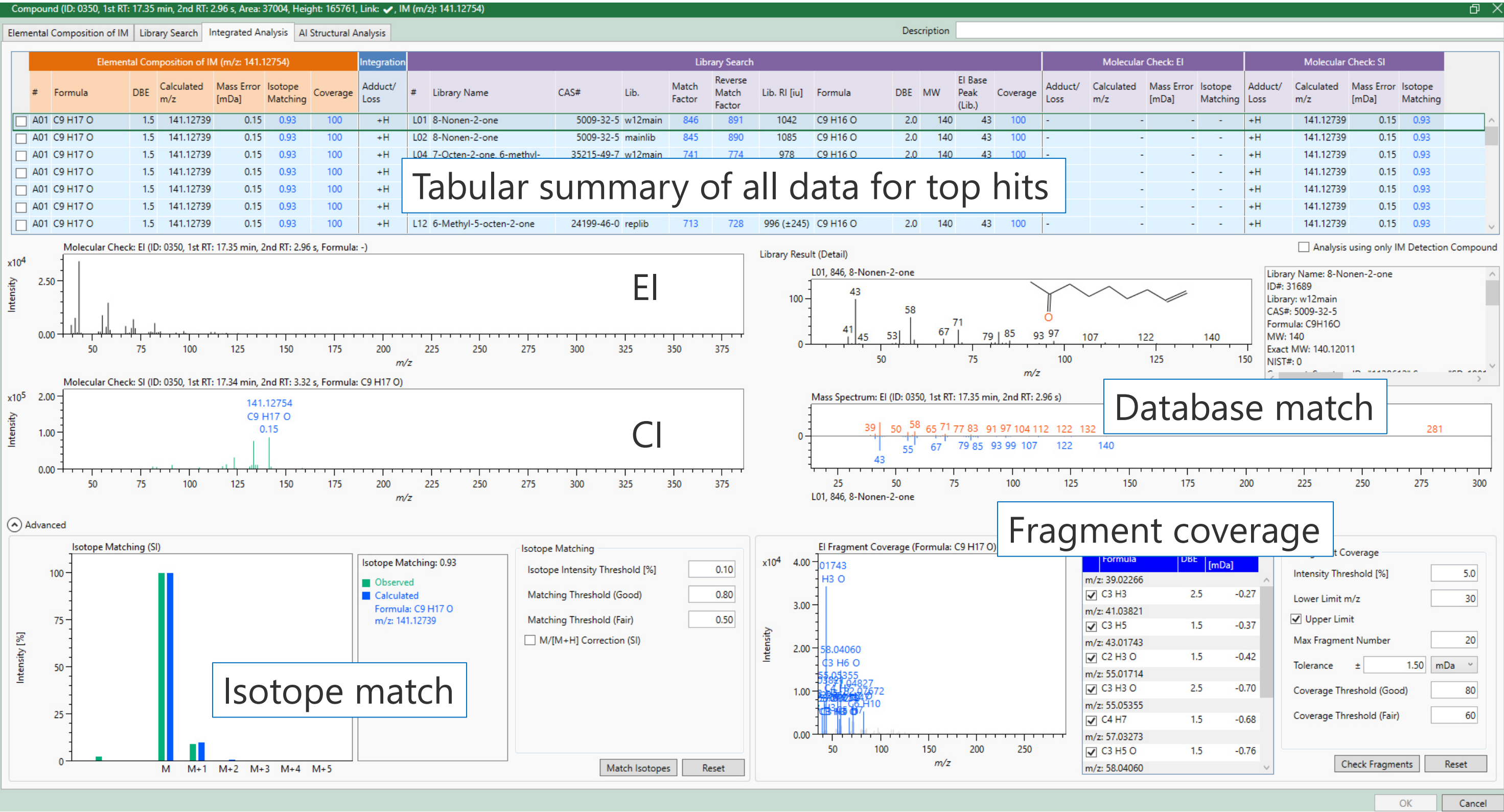
- Correlates EI and SI data
- Integrates all information
 - Elemental composition
 - Isotope match
 - Database search
 - Fragment coverage
 - RI matching
- AI database including pyrolysates
- Higher confidence in assignments

8-Nonene-2-one identification.





Integrated Analysis tab





GCxGC separation OK, but RI's and match factors are identical or within tolerance



It's unlikely that we'll ever ID every peak with 100% confidence, but we can get an idea of reasonable compositions.