

# Chromium Speciation of Drinking Waters by IC-ICPMS

**Haihan Chen**, Jonathan Peters, Hui Guo, and Ruth Marfil-Vega

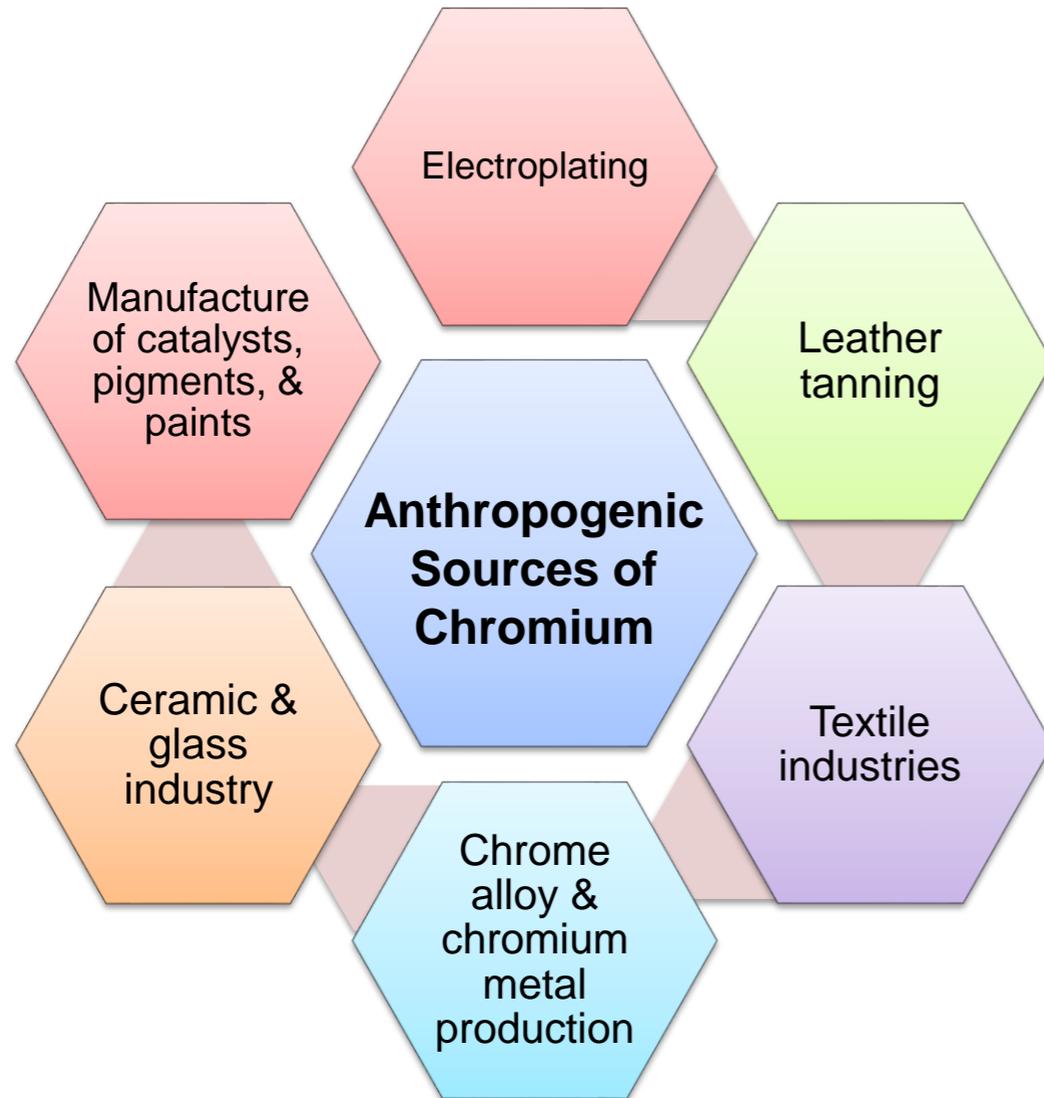
Shimadzu Scientific Instruments

# In today's presentation

1. Why Chromium speciation?
2. Current methods
3. New method
4. Results
5. Conclusions & Benefits
6. Q&A



# Chromium & Its Sources



- ❑ Naturally-occurring element found in rocks, animals, plants, and soil, where it exists in combination with other elements to form various compounds.
- ❑ Industrial releases to the air, water, and soil account for the majority of the anthropogenic releases.

(Johnson et al. The contemporary anthropogenic chromium cycle 2006)

- ❑ Occurs in the soluble state in drinking water and mainly presents as Cr(III) & Cr(VI).

# Why Chromium Speciation?

**Chromium(III)**

**MAIN FUNCTIONS**

- Assists insulin action

**DAILY RECOMMENDATION**

<b>35</b> µg	<b>25</b> µg
Men 19–50 Years	Women 19–50 Years
<b>30</b> µg	<b>20</b> µg
Men 51+ Years	Women 51+ Years

- ❑ Cr(III) is an essential human dietary element.
- ❑ Cr(VI) is an occupational carcinogen, and a reproductive toxicant, and can cause other health issues.

[https://www.atsdr.cdc.gov/csem/chromium/standards\\_and\\_regulations.html](https://www.atsdr.cdc.gov/csem/chromium/standards_and_regulations.html)

**Exposure to chromium-6 can cause:**

 Lung Cancer	 Liver Damage	 Reproductive Problems	 Developmental Harm
--	---	--	---

# Chromium Regulations

Table 1. Regulations and Guidelines for Chromium

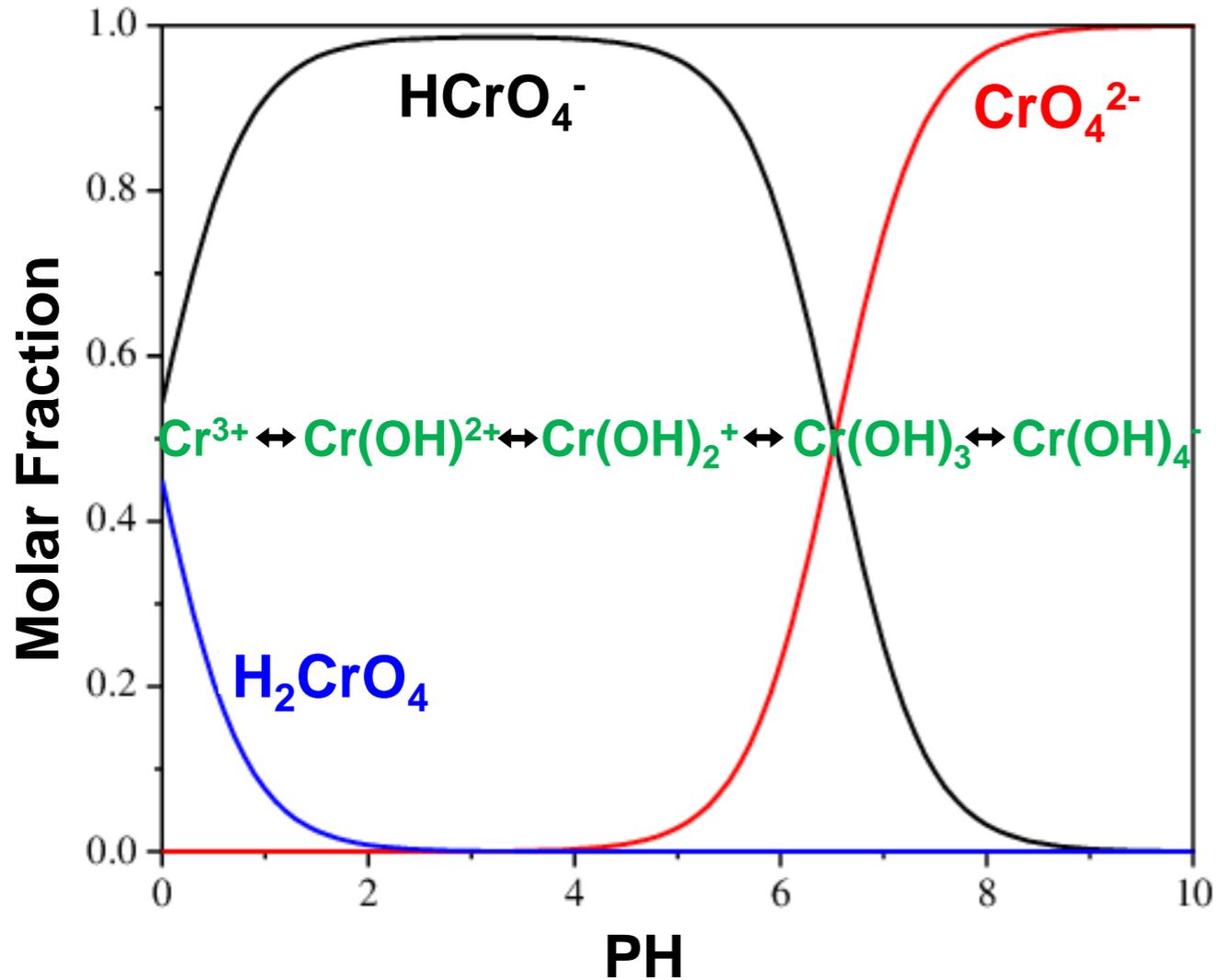
Agency	Focus	Level	Comments
American Conference of Governmental Industrial Hygienists	Air: workplace	10 µg/m <sup>3</sup> as Cr	Advisory; TWA* to avoid carcinogenic risk from insoluble Cr(VI) compounds
		50 µg/m <sup>3</sup> as Cr	TWA for water-soluble Cr(VI) compounds
		500 µg/m <sup>3</sup> as Cr	TWA for chromium metal and Cr(III) compounds
National Institute for Occupational Safety and Health	Air: workplace	1 µg/m <sup>3</sup> as Cr	Advisory; TWA (10-hour) for chromic acid and all Cr(VI) compounds
		500 µg/m <sup>3</sup> as Cr	Advisory; TWA (10-hour) for chromium metal and Cr(II) and Cr(III) compounds
Occupational Safety and Health Administration	Air: workplace	5 µg/m <sup>3</sup> as CrO <sub>3</sub> /m <sup>3</sup>	Regulation; PEL <sup>+</sup> for chromic acid and chromates, (8-hour TWA)
		500 µg/m <sup>3</sup> as Cr	PEL for Cr(II) and Cr(III) compounds (8-hour TWA)
		1,000 µg/m <sup>3</sup> as Cr	PEL for chromium metal and insoluble compounds (8-hour TWA)
Environmental Protection Agency	Air: environment	Not available	Chromium is listed as a hazardous pollutant
	Drinking water	100 µg/L	Regulation; current MCL <sup>+</sup> for total chromium

- ❑ Different regulations established and in development for Total Cr, Cr(III) and Cr(VI).
- ❑ **EPA** currently regulates the **total Cr**. Cr(VI) has been under review since 2008.
- ❑ Are new regulations and manufacturing practice about PFAS going to change Cr emissions from electroplating industries?



<https://www.cdc.gov/niosh/topics/hexchrom/default.html>

# Chromium Speciation as f(pH)



- Cr(III) typically exists as cationic aqua-hydroxo complexes
- Cr(VI) exists typically as an anionic chromate species
- Interconversion of Cr(III) & Cr(VI) depending on sample conditions (pH)

# Methods for Chromium Speciation

	EPA 7196	EPA 218.7	ISO/CD 24384	New method
Target	Dissolved Cr(VI)	Cr(VI)	Cr(III) & Cr(VI)	Cr(III) & Cr(VI)
Method	Colorimetric	IC	LC-ICP/MS	IC-ICP/MS
Samples	EP/TCLP extracts, groundwater, domestic and industrial waste (limited)	Drinking water	Wastewater, surface water, groundwater, or tap waters	Continue in this presentation!
DL	ppm	ppt	ppt	
Limitations	<ul style="list-style-type: none"> <li>• Interferences</li> <li>• Laborious sample prep</li> <li>• Lack of sensitivity</li> </ul>	<ul style="list-style-type: none"> <li>• Post-column derivatization required</li> </ul>	<ul style="list-style-type: none"> <li>• Chelating pre-treatment</li> <li>• Laborious sample prep</li> </ul>	

# Instrumentation

- ❑ All analyses were run on a IC coupled to ICPMS.
- ❑ The ICPMS is equipped with a collision cell that uses helium (He) to discriminate polyatomic interferences based on kinetic energy.
- ❑ Cr was analyzed at 52 m/z with He gas on to remove polyatomic interferences such as ArC and ClOH.

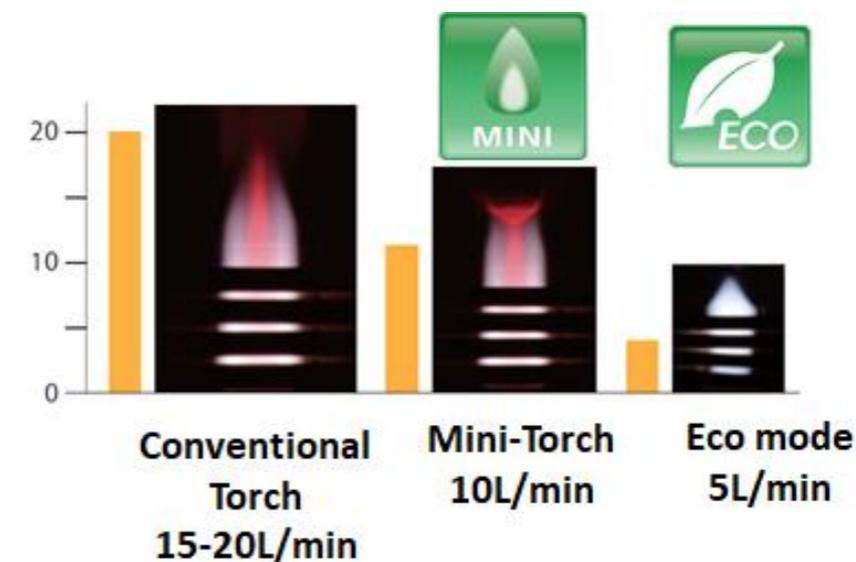


Shimadzu ICPMS-2030

# Instrumentation – Inductively Coupled Plasma Mass Spectrometry (ICPMS)

## Operating Conditions of ICPMS

Parameter	Setting	Parameter	Setting
Radio Freq. Power	1.20 kW	Mix Gas	0.00 L/min
Sampling Depth	5.0 mm	Cell Gas	6.0 mL/min
Plasma Gas	8.0 L/min	Cell Voltage	-21 V
Auxiliary Gas	1.10 L/min	Energy Filter	7.0 V
Carrier Gas	0.70 L/min	Chamber Temp.	5°C



- ❑ An LC fittings kit was used to connect IC tubing directly to the nebulizer.

# Instrumentation – Ion Chromatography (IC)



Shimadzu Prominence IC

- The IC was configured with an inert flow path.

## Optimization Studies for Chromatographic Separation of Cr(III) and Cr(VI)

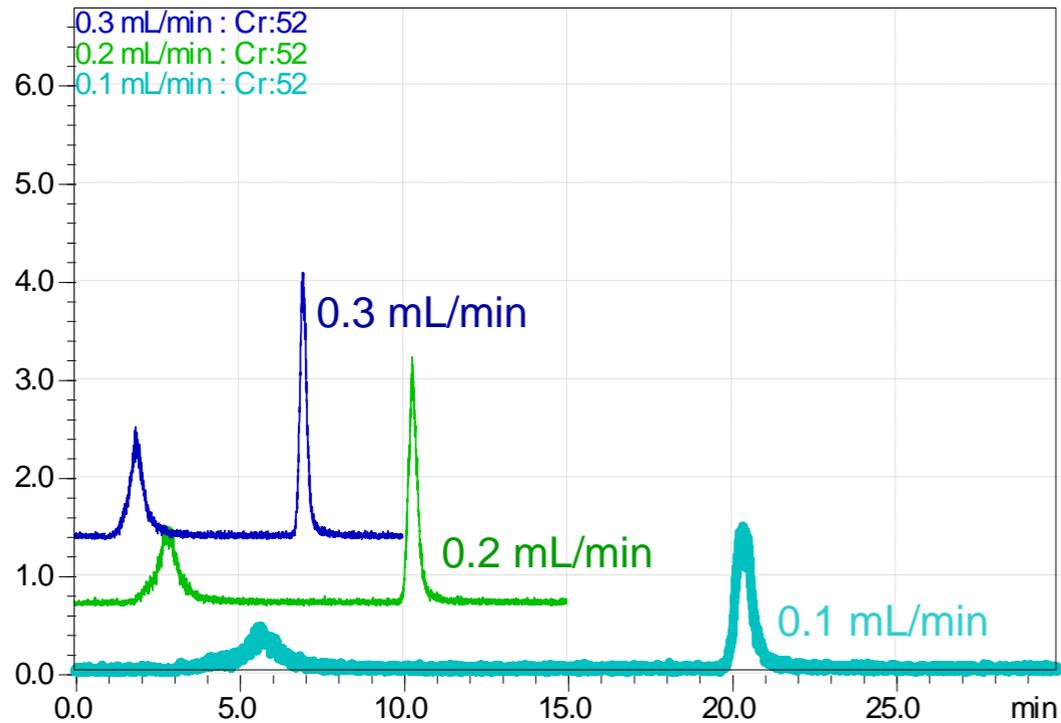
Parameter	Setting	Parameter	Setting
Column	Shodex™ VC-50 2D	Separation Scheme	Isocratic
Mobile Phase	6-9 mM HNO <sub>3</sub>	Column Temp.	30-50°C
pH	2-3	Injection Volume	5-100 µL
Flow Rate	0.1-0.3 mL/min	LC Vials	Plastic, 1.5 mL

# Method - Sample Preparation

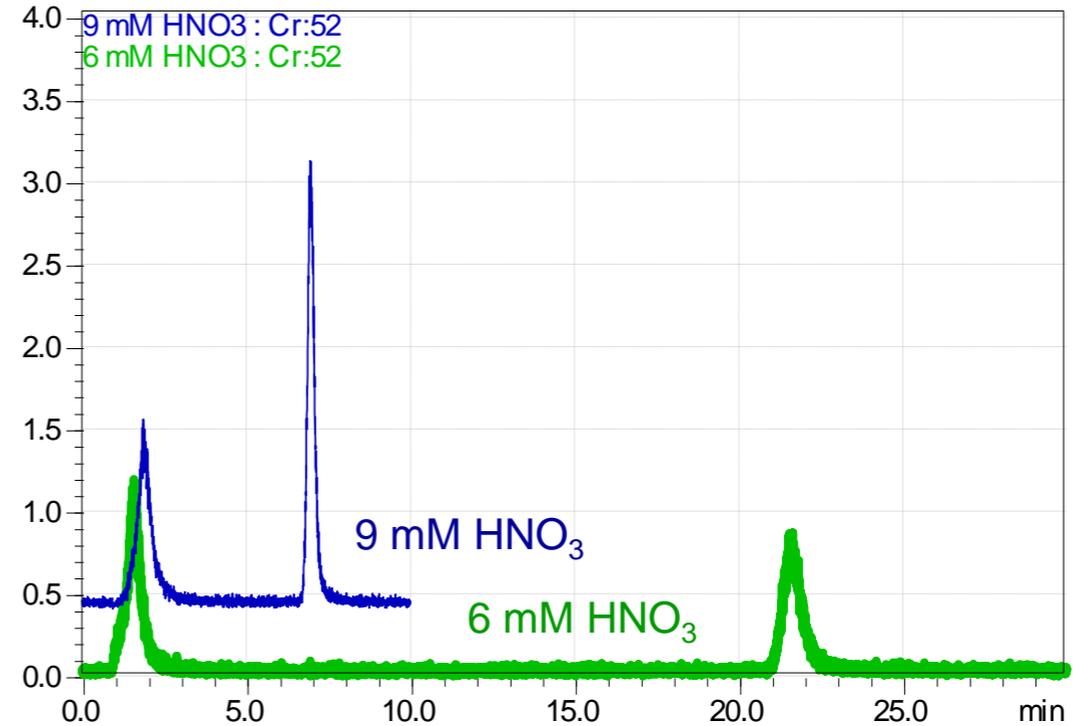
- ❑ Samples analyzed directly without any pretreatment
  - Tap water
  - Well water
  - Commercially available mixed standards of 22 elements
  - Spiked samples
- ❑ Chromium standards
  - Diluting 1000 mg/L stock solutions of trivalent and hexavalent chromium in deionized water
- ❑ Mobile phase  $\text{HNO}_3$ 
  - Made from trace metal grade concentrated  $\text{HNO}_3$



# Results – Optimized Chromatographic Conditions

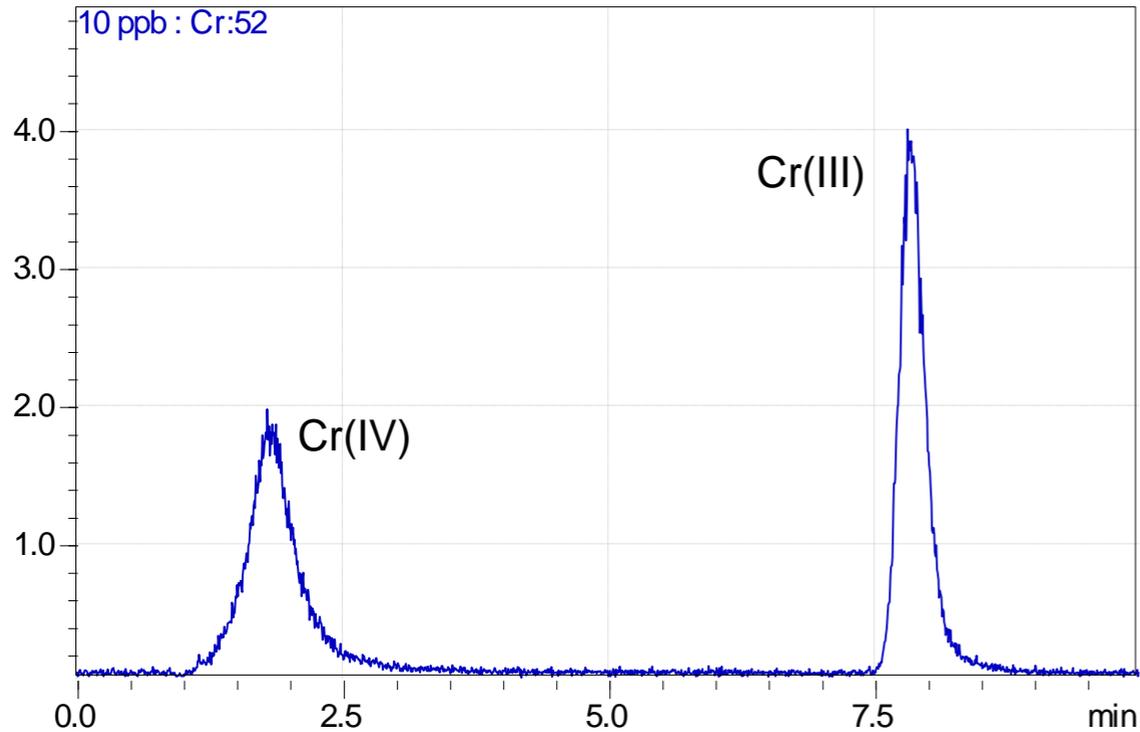


Chromatograms at different flow rates



Chromatograms with different concentrations of mobile phase

# Results – Optimized Chromatographic Conditions



Chromatogram of 10 ppb Cr(III) and Cr(VI)

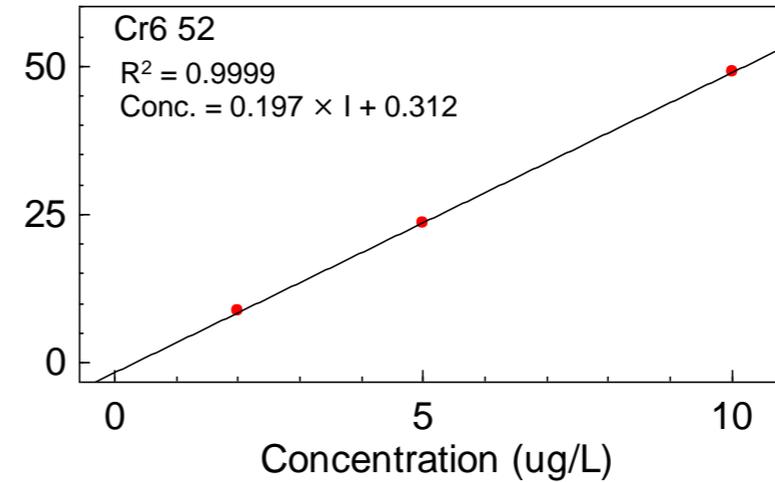
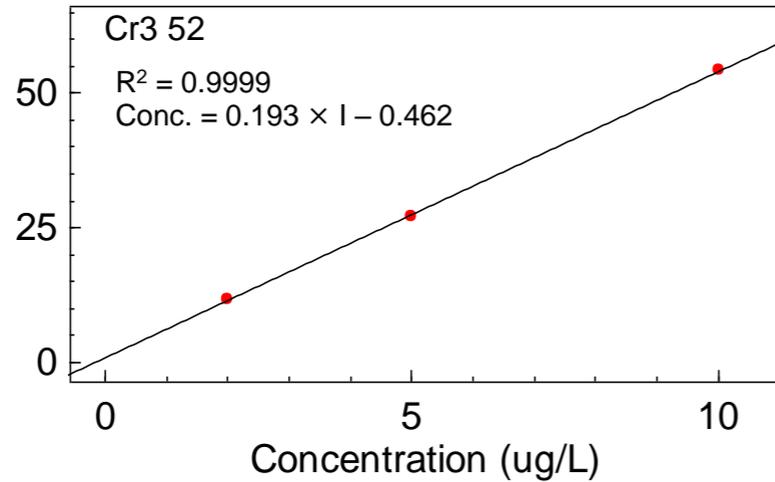
- ❑ Chromium species, Cr(III) and Cr(VI), are well separated and elute within 10 minutes.

## Operating Conditions of IC

Parameter	Setting
Column	Shodex™ VC-50 2D
Separation Scheme	Isocratic
Mobile Phase	9mM HNO <sub>3</sub>
Column Temp.	50°C
pH	2.0
Injection Volume	20 µL
Flow Rate	0.3 mL/min
LC Vials	Plastic, 1.5 mL

# Results – Calibration Curves

Calibration curves of Cr(III) and Cr(VI) in the range of 2-10 µg/L



## Characteristics for Chromium Speciation Determined

	Cr(III)	Cr(VI)
Retention Time (min)	7.79	1.84
Limit of Detection (µg/L)	0.20	0.35
Limit of Quantitation (µg/L)	0.67	1.67
Correlation Coefficient ( $R^2$ )	0.9999	0.9999
Equation	Conc. = $0.193 \times I - 0.462$	Conc. = $0.197 \times I + 0.312$
Residual Range (µg/L)	-0.005 - 0.001	-0.03 - 0.02
Relative Standard Error (RSE, %)	0.05	1.22

# Results – Sample Analysis

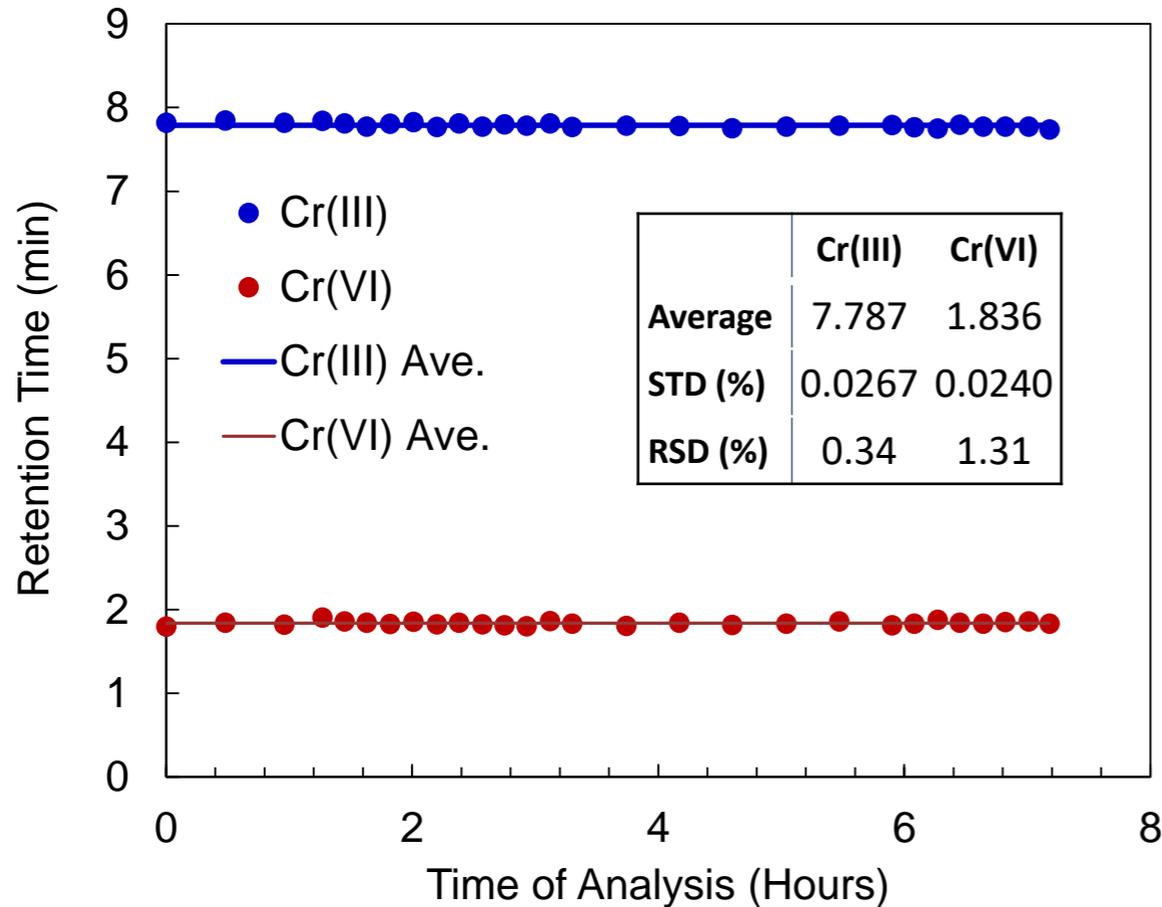
**Concentrations of Cr in  $\mu\text{g/L}$  in Original and Fortified Samples as well as Recovery Yields in Percent**

Sample	Cr(III)	Cr(VI)
Tap water	n.d.	n.d.
Fortified tap water	5.05	5.16
<b>Recovery (%)</b>	<b>101</b>	<b>103</b>
Well water	n.d.	0.738
Spiked well water	5.15	5.61
<b>Recovery (%)</b>	<b>103</b>	<b>97</b>
Commercial Standard	3.92	n.d.
Fortified commercial standard	9.06	4.77
<b>Recovery (%)</b>	<b>103</b>	<b>95</b>

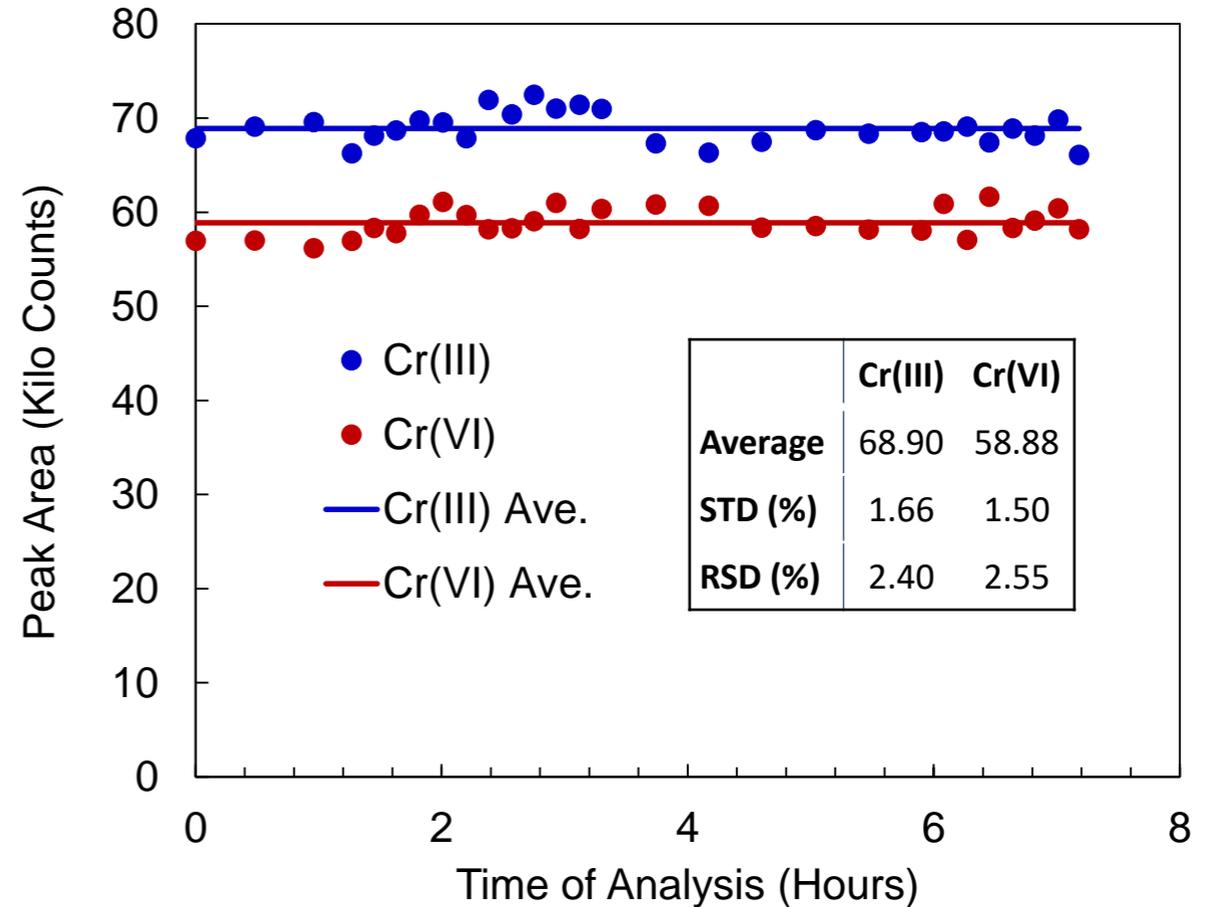
n.d. = not detected. Cr(III) < 0.20  $\mu\text{g/L}$ , Cr(VI) < 0.35  $\mu\text{g/L}$ .

# Results – Instrument Stability & Precision

Variation of Retention Time for Multiple Injection of 10 µg/L Cr(III) and Cr(VI) Standard Over 7 Hours



Variation of Peak Area for Multiple Injection of 10 µg/L Cr(III) and Cr(VI) Standard Over 7 Hours



# Conclusions

- ❑ ICPMS coupled with IC provides excellent sensitivity, precision, accuracy, stability, fast time response and high sample throughput for determination of chromium speciation in waters.
- ❑ The use of 9 mM nitric acid other than salt solutions as mobile phase reduces background signal and possible interference.
- ❑ The use of a column to separate both cations and anions enables fast separation of chromium speciation without any sample pretreatment with complexing agents.
- ❑ Eliminating sample preparation avoids any possible risk of contamination as well as maximizes sample throughput.





**Haihan Chen, Ph.D.**  
**Product Coordinator**  
**Shimadzu Scientific Instruments**  
**[hachen@shimadzu.com](mailto:hachen@shimadzu.com)**



**OneLabOneEarth.com**