

Comprehensive Characterization of Natural Rubber Samples using Thermal Field-Flow-Fractionation coupled with MALS and Triple Detection

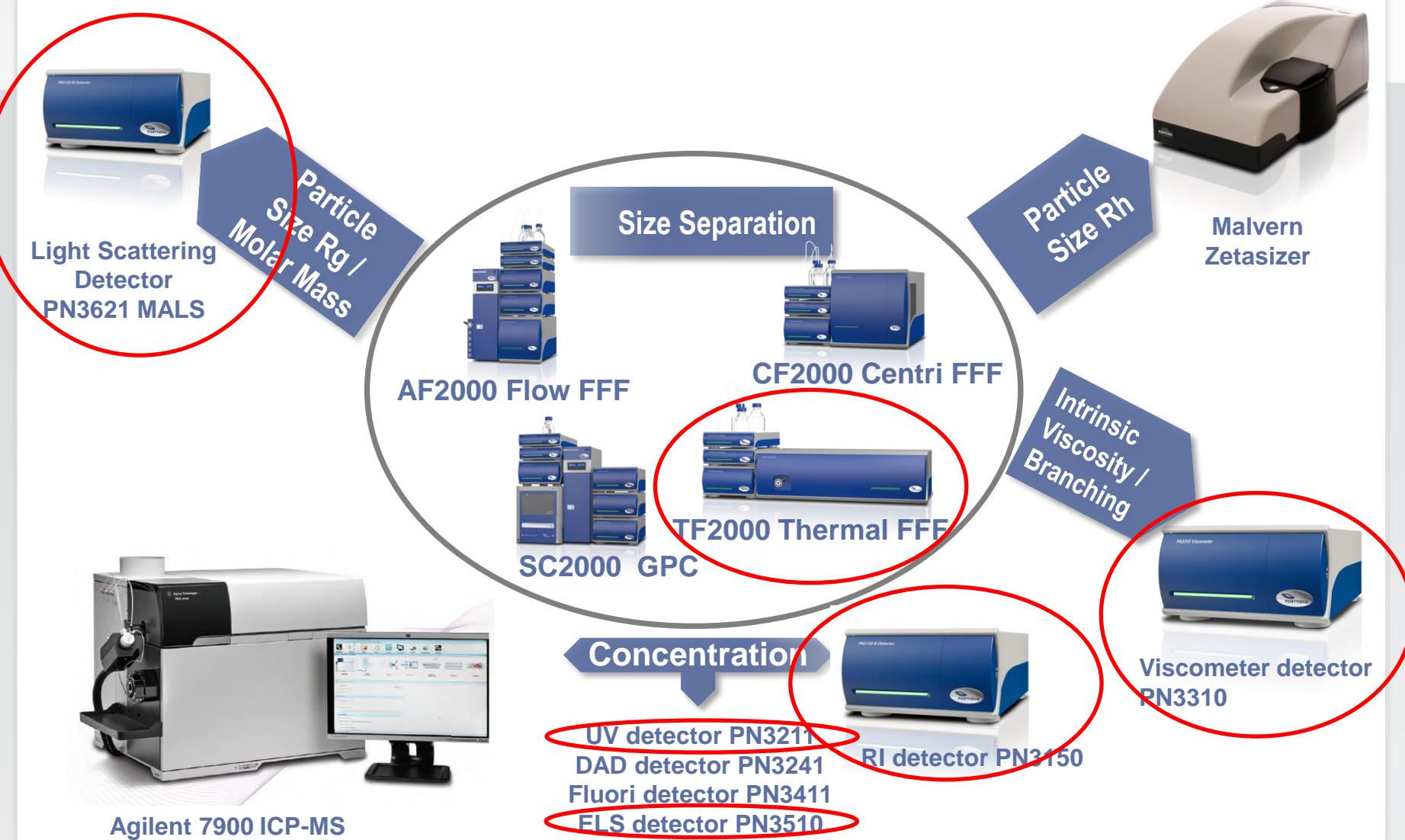


Dr. Gerhard Heinzmann, Postnova Analytics GmbH, Germany

International Symposium on GPC/SEC and Related Techniques,
Amsterdam, The Netherlands, September 26th-29th, 2016

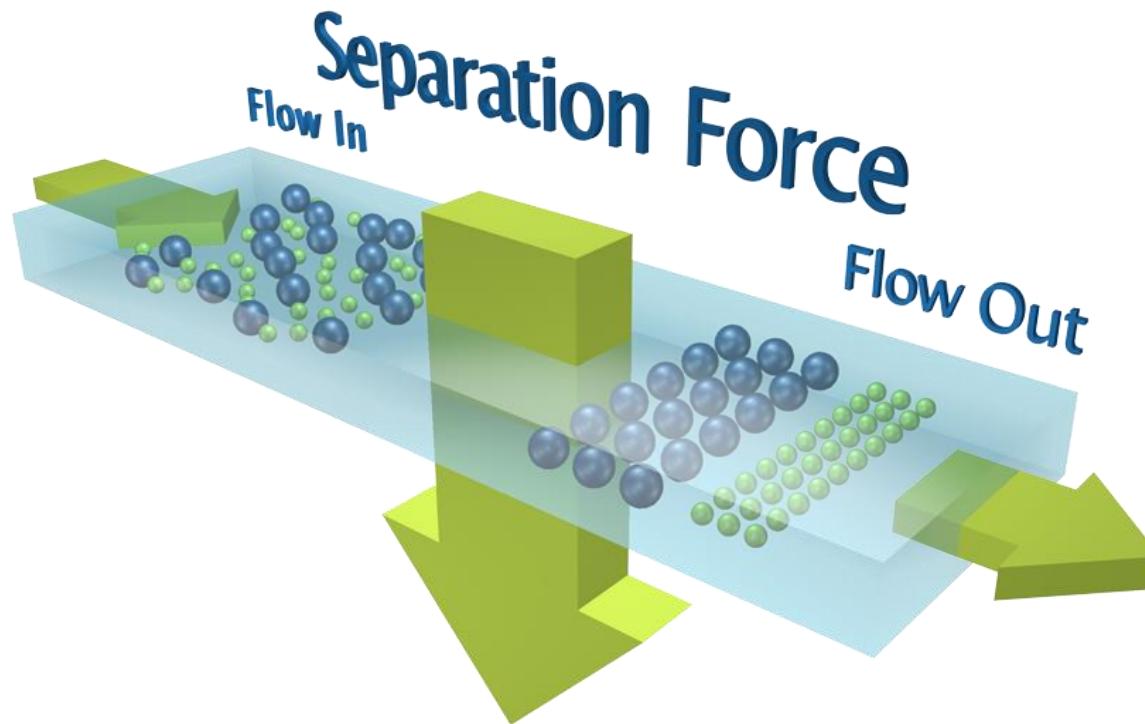
- FFF Separation Platform
- Thermal FFF (TF3)
- TF3-MALS and TF3-Triple/Tetra/Penta Detection
- Application Examples

Separation and Detection Platform



FFF Separation Principle

Separation Mechanism



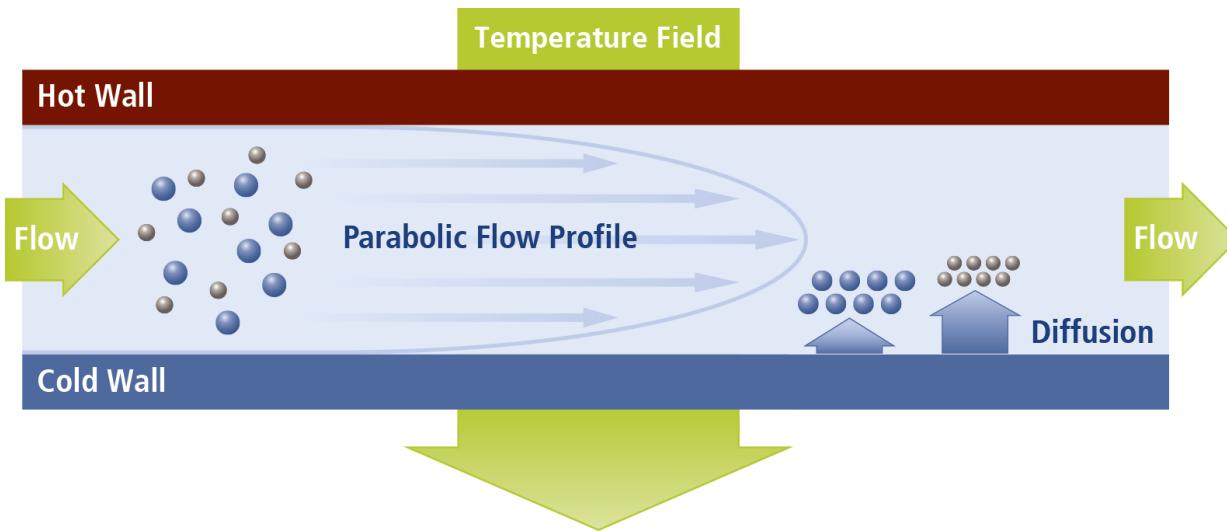
- Separation in a narrow ribbon-like channel
- Laminar flow inside the channel
- External field perpendicular to the solvent flow

Thermal FFF – Principle

TF2000: Separation

$$V_E \sim D T \Delta T / D$$

Thermal Diffusion Coefficient
-> Depends on chemical composition



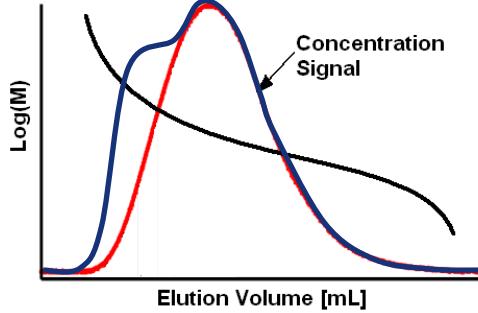
- Thermal gradient up to $\Delta 120^\circ\text{C}$
- Separation kDa up to several MDa
- Analysis time, 10 – 120min (no upper limit)
- Separation depends on Size and Chemical Composition ("2 Dimensional"?)



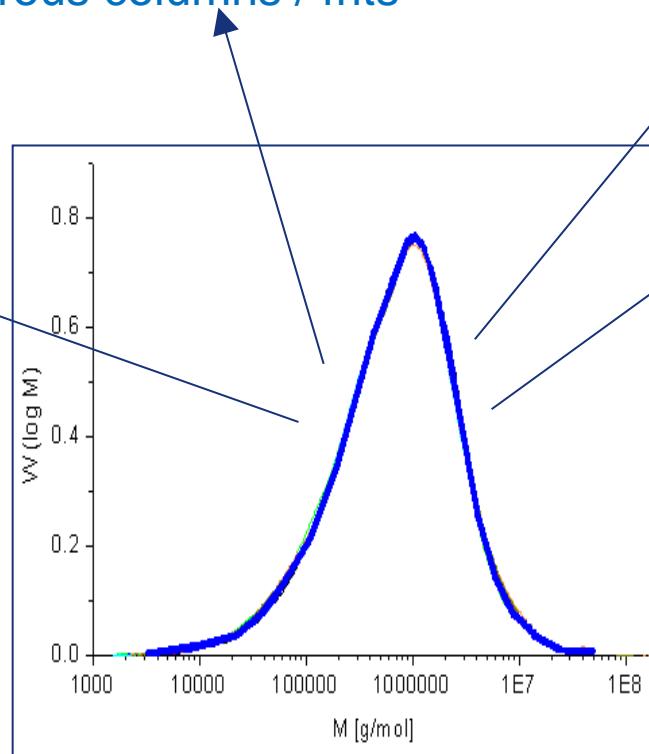
Thermal FFF vs. GPC/SEC

Advantages of Thermal FFF compared to GPC/SEC:

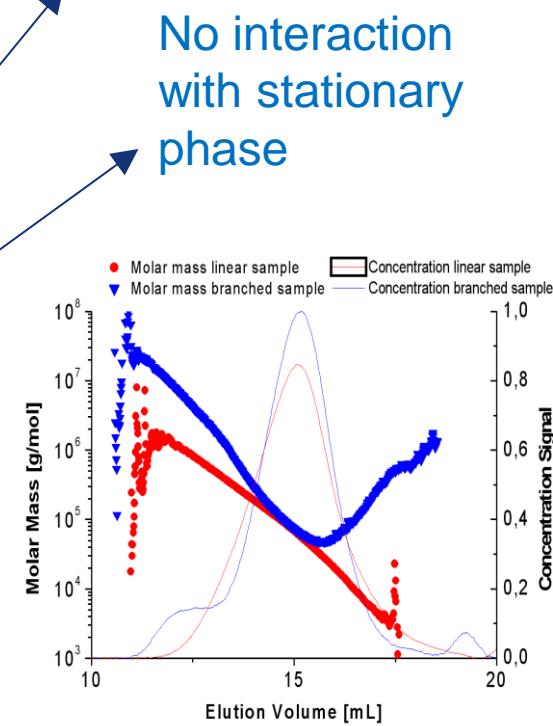
Excellent separation for high MW → extended 'Exclusion Limit'



No shear degradation in porous columns / frits



No filtration of gel material, no clogging of columns



FFF – Triple Detection

- MALS
- Viscometer
- Concentration Detector(s)

Setup of TF2000 Thermal FFF with Triple Detection (Penta Detection) MALS / Viscometer / 3x Concentration Detector



Detector Setup: UV => MALS => Visc => RI => ELSD

Conditions: Solvent = THF, Flow rate = 0.3 mL/min

Alternative Setup: UV => MALS => Split: Visc => Waste and RI => ELSD => Waste

Determination of Mw, Rg, IV and Structure



$$\text{LS signal} = K_{\text{LS}} * (\text{dn/dc})^2 * \text{Conc} * \text{Mw}$$

$$\text{RI signal} = K_{\text{RI}} * (\text{dn/dc}) * \text{Conc}$$

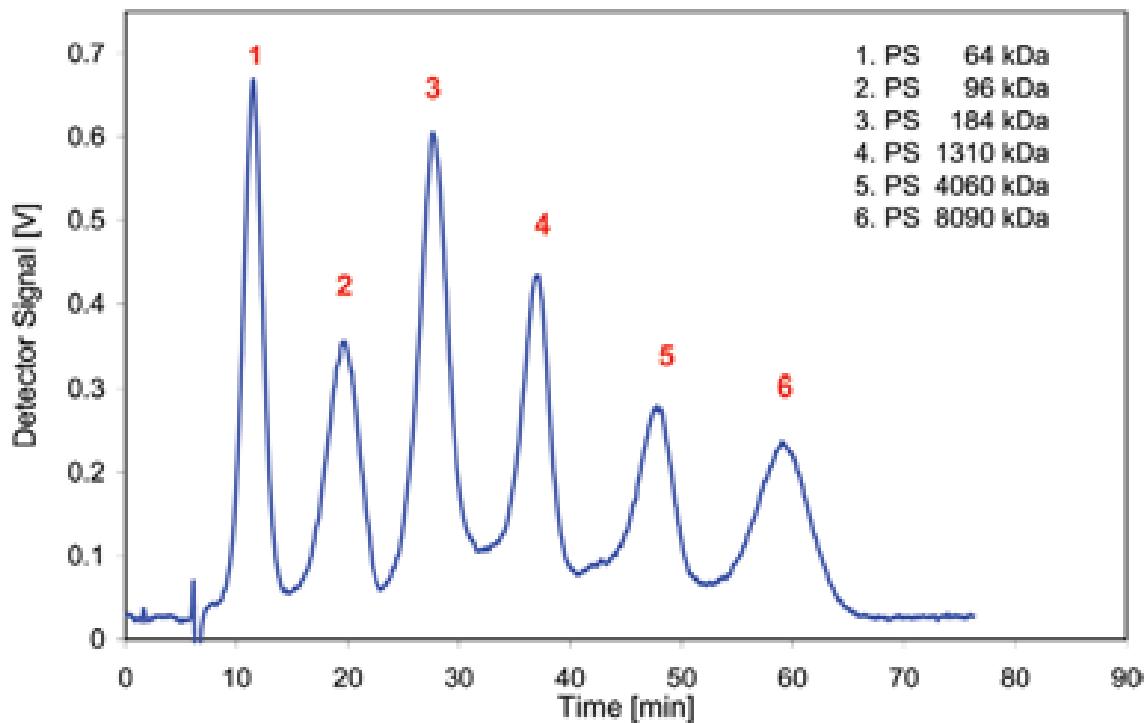
$$\text{UV signal} = K_{\text{UV}} * \epsilon * \text{Conc}$$

$$\text{Visco signal} = K_{\text{Visco}} * [\eta] * \text{Conc}$$

- **MW from MALS + RI**
- **Rg from MALS**
- **IV from Viscometer + RI**
- **Rh from Viscometer + RI + MALS**
- **Mark-Houwink plot (log IV vs. log Mw) shows structure and degree of branching**

Results

Separation of PS standards



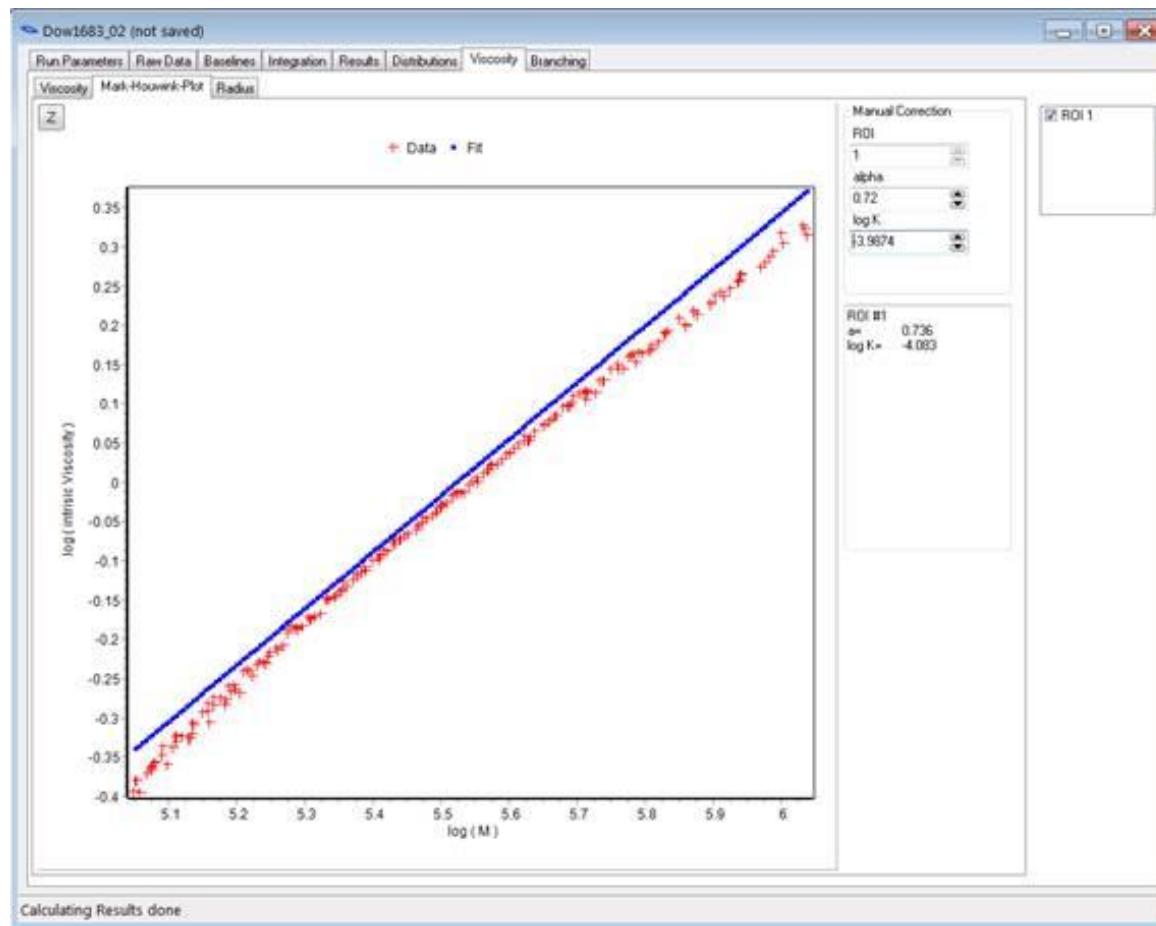
System

- TF2000 FFF System
- PN3150 RI Detector

Conditions

- Injection Volume: 20 μL
- Concentration: 2 mg/mL
- Temp. grad. $\Delta T = 90^\circ\text{K}$ to 0°K

DOW PS 1683 / $M_w = 250 \text{ kDa}$, $\text{PD} = 2.5$



System

- TF2000 FFF System
- PN3621 MALS
- PN3310 Viscometer
- PN3211 UV
- PN3150 RI
- PN3510 ELSD

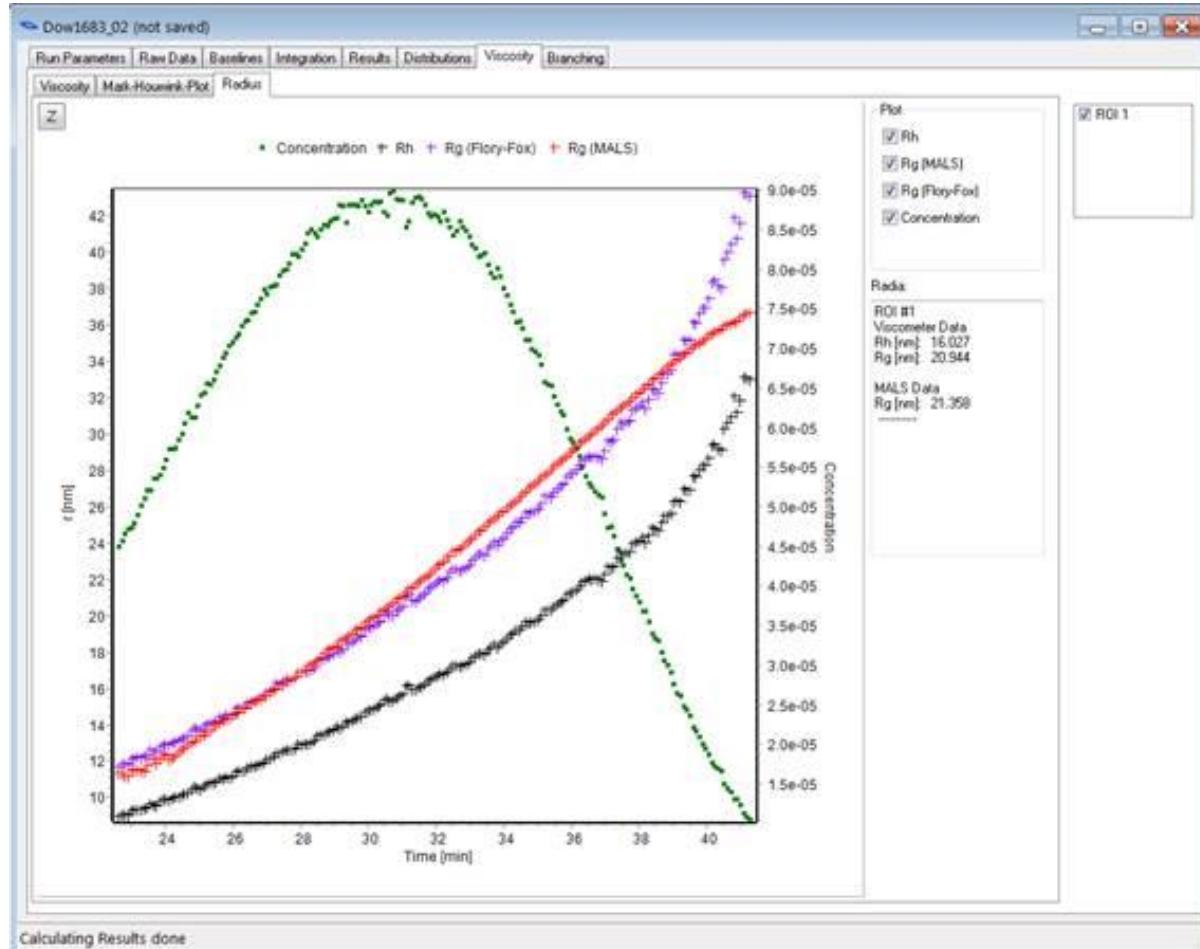
Conditions

- Injection Volume: $50 \mu\text{L}$
- Concentration: 5 mg/mL
- Temp. grad. $\Delta T = 90^\circ\text{K} \text{ to } 0^\circ\text{K}$

TFFF Applications – PS with Pentadetection



DOW PS 1683 / $M_w = 250 \text{ kDa}$, $\text{PD} = 2.5$

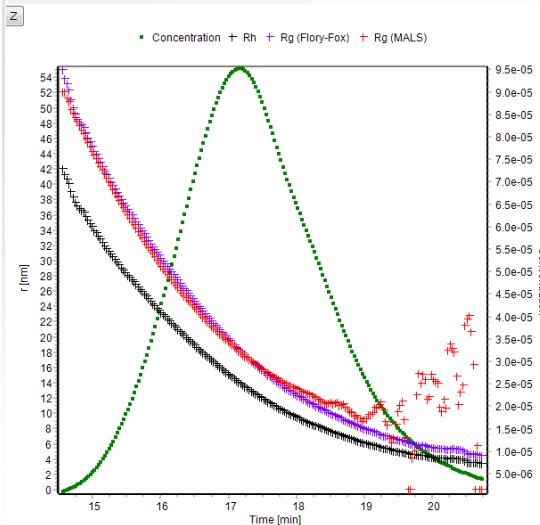


System

- TF2000 FFF System
- PN3621 MALS
- PN3310 Viscometer
- PN3211 UV
- PN3150 RI
- PN3510 ELSD

Conditions

- Injection Volume: 50 μL
- Concentration: 5 mg/mL
- Temp. grad. $\Delta T = 90^\circ\text{K}$ to 0°K

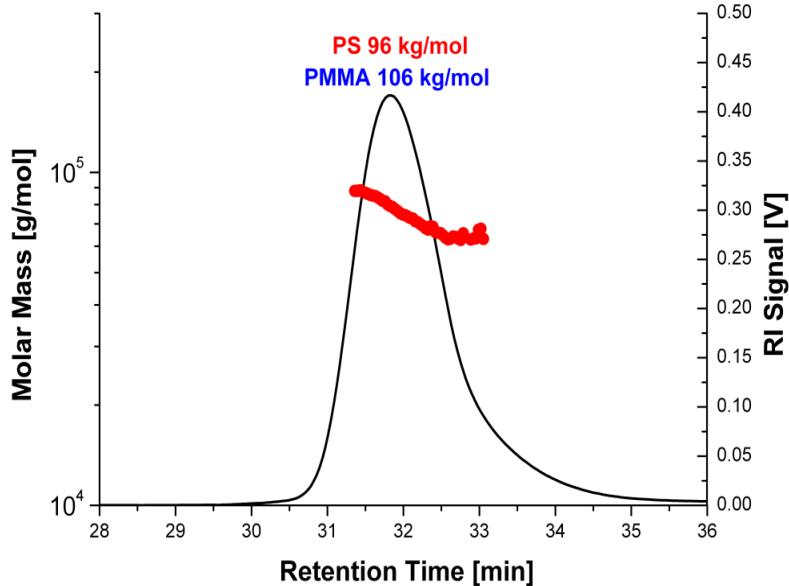


TFFF - Separation by Chemical Composition

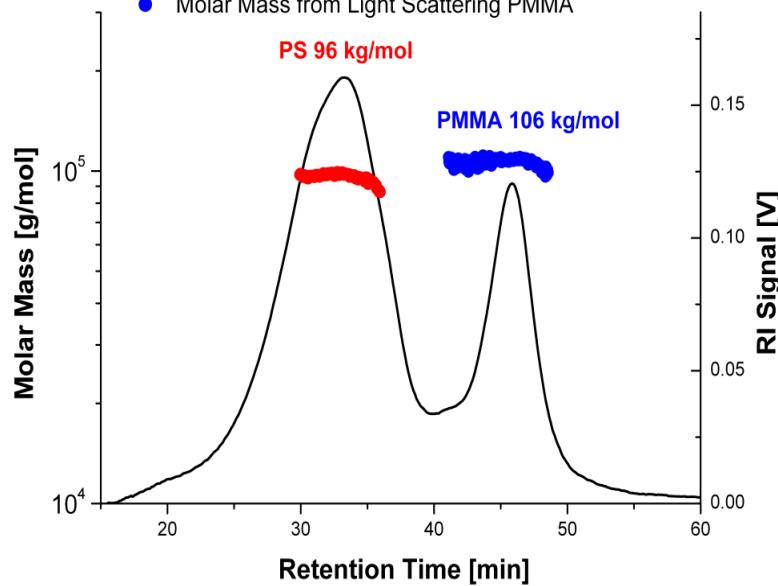


Separation of PS and PMMA-Standards with same R_h SEC vs. Thermal FFF (TF3)

- Molar Mass from Light Scattering PS + PMMA

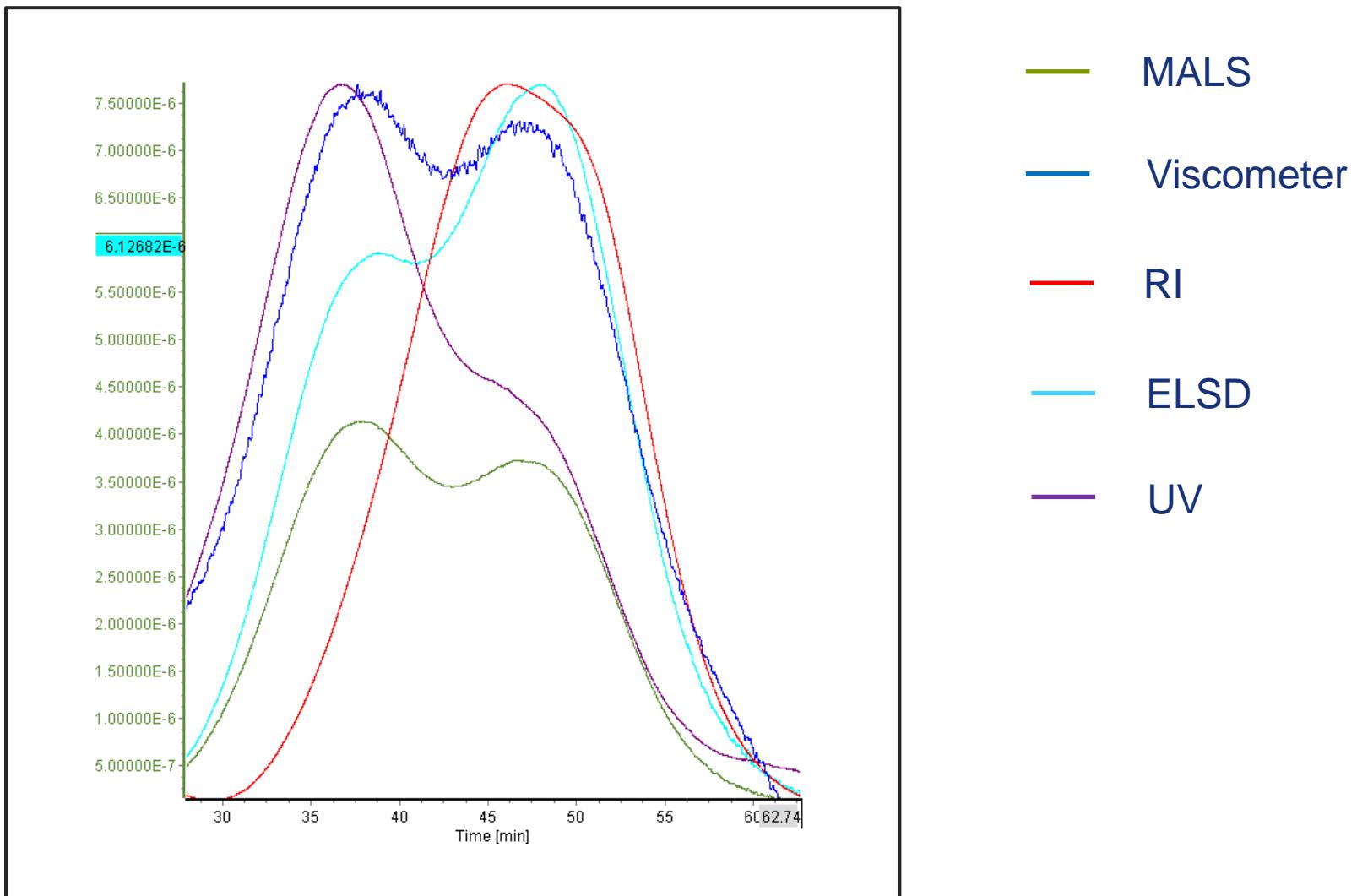


- Molar Mass from Light Scattering PS
- Molar Mass from Light Scattering PMMA



TF3 enables separation of molecules with the same hydrodynamic volume according to their chemical composition!

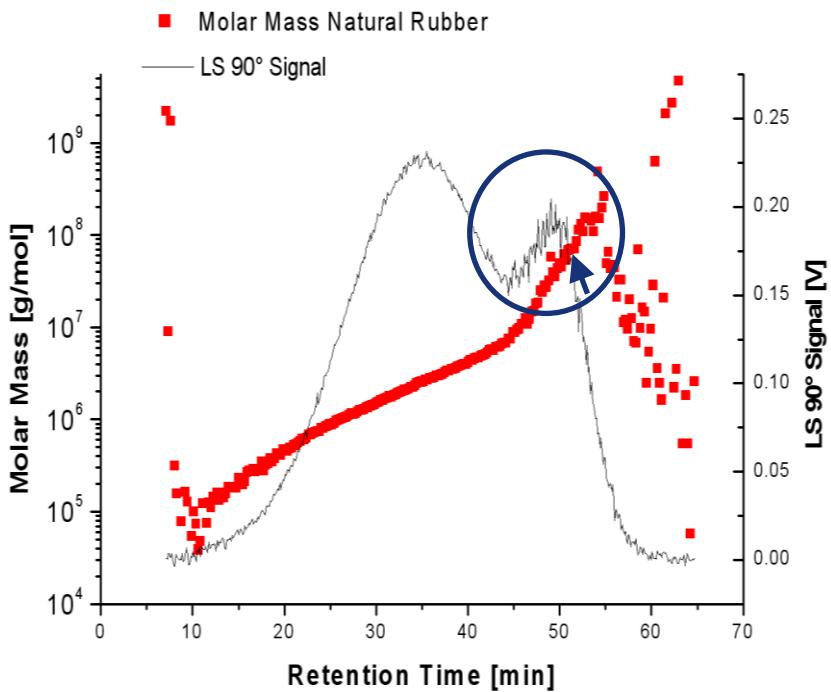
TFFF Applications – SBR Rubber with Pentadetection



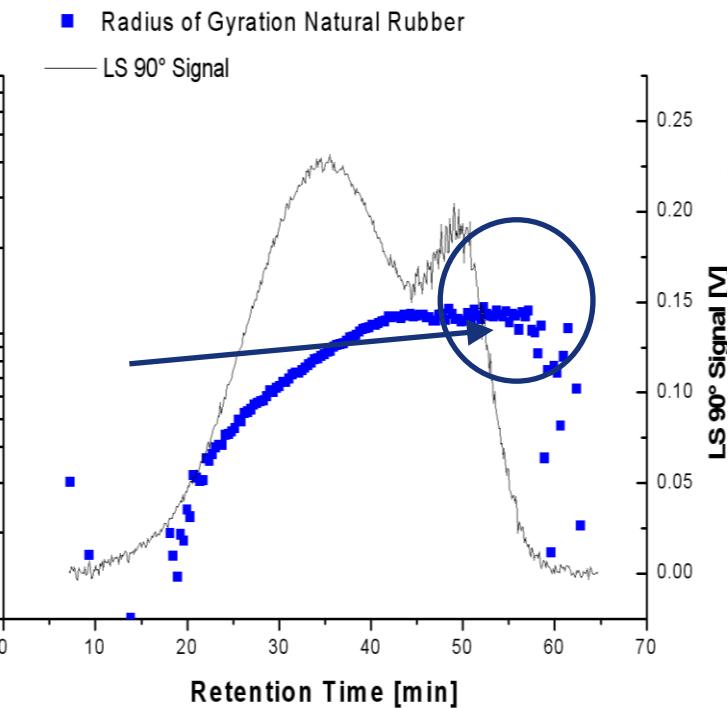
Application Example: Natural Rubber

TFFF Applications - Natural Rubber

Molar Mass



Radius of Gyration

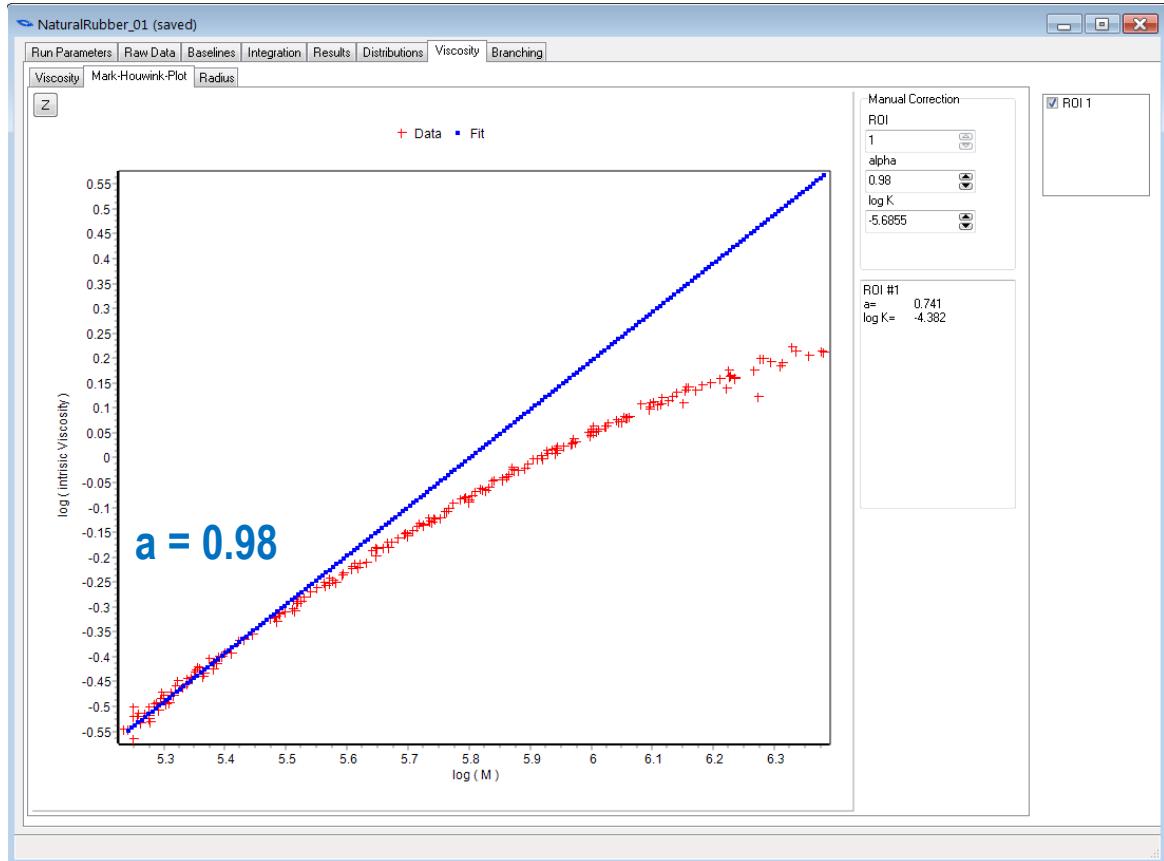


- **Nanoscale impurities with same size separated according to different chemical composition**

TFFF Applications – Natural Rubber with Pentadetection



Natural Rubber Sample (Polyisoprene)



Stiff chain structure in low MW area

System

- TF2000 FFF System
- PN3621 MALS
- PN3310 Viscometer
- PN3211 UV
- PN3150 RI
- PN3510 ELSD

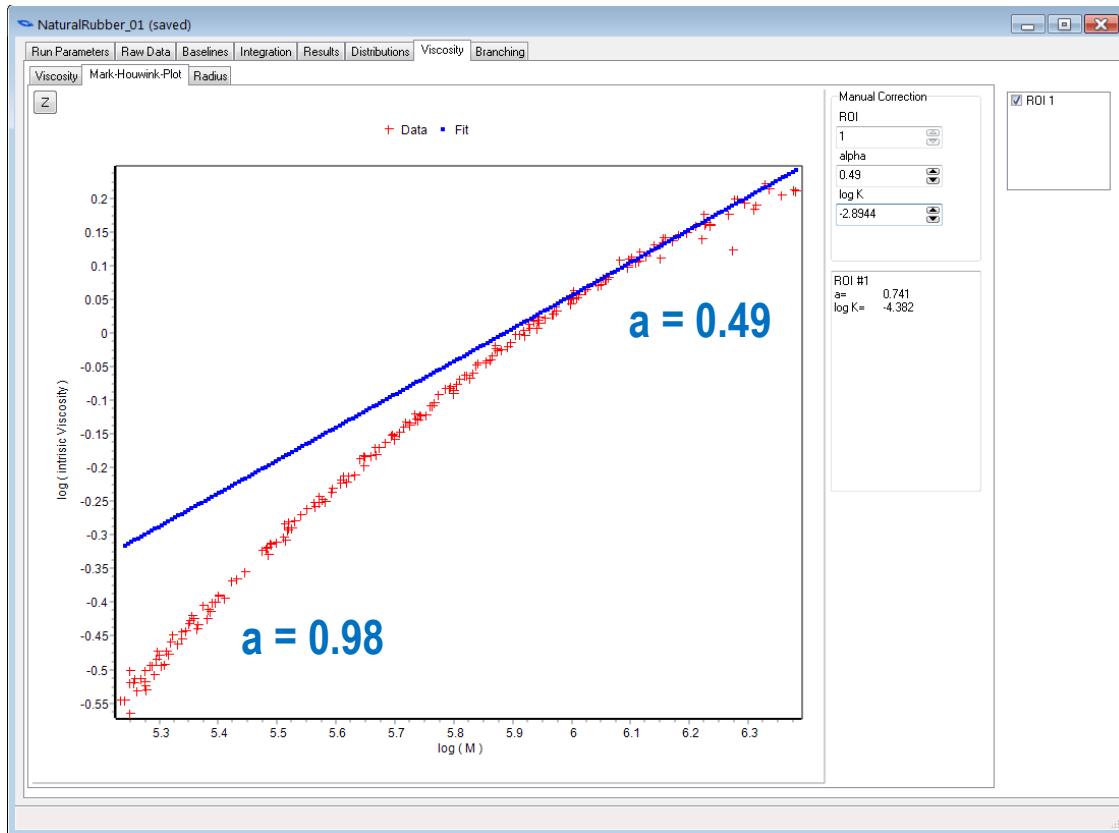
Conditions

- Injection Volume: 50 μL
- Concentration: 5 mg/mL
- Temp. grad. $\Delta T = 90^\circ\text{K} \text{ to } 0^\circ\text{K}$

TFFF Applications – Natural Rubber with Pentadetection



Natural Rubber Sample



System

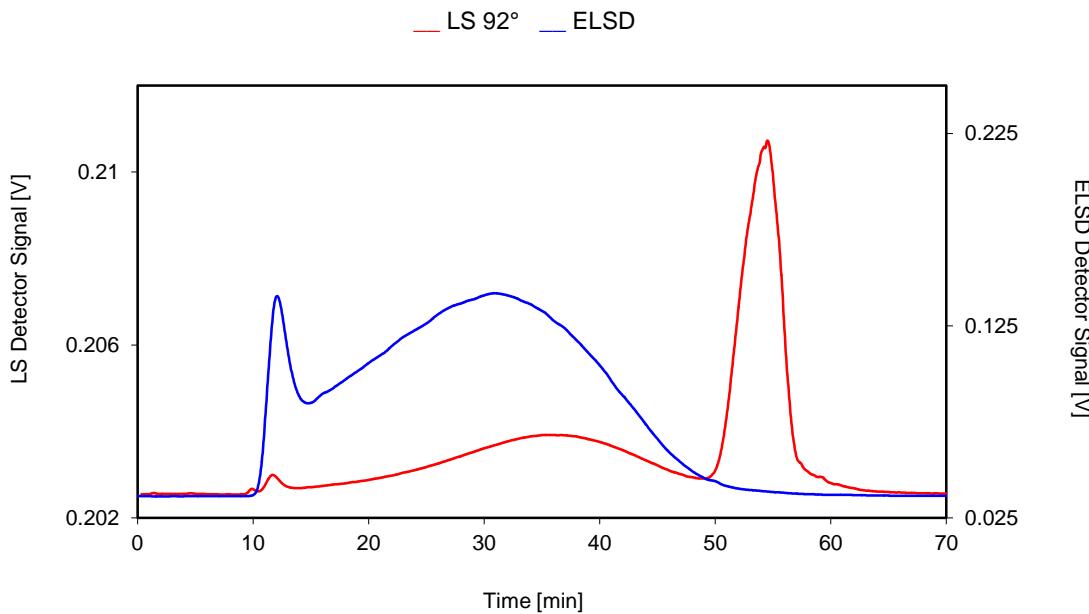
- TF2000 FFF System
- PN3621 MALS
- PN3310 Viscometer
- PN3211 UV
- PN3150 RI
- PN3510 ELSD

Conditions

- Injection Volume: 50 μL
- Concentration: 5 mg/mL
- Temp. grad. $\Delta T = 90^\circ\text{K} \text{ to } 0^\circ\text{K}$

More compact structure in high MW area (branching?)

Raw Data Fractogram of TF3 - MALS and ELSD



- The fractogram shows a system start peak at 11 min.
- The 1st peak – the main peak - was detected between 15 - 50 min by light scattering and ELSD detection.
- A 2nd peak was detected at 54 min predominantly by light scattering.

System

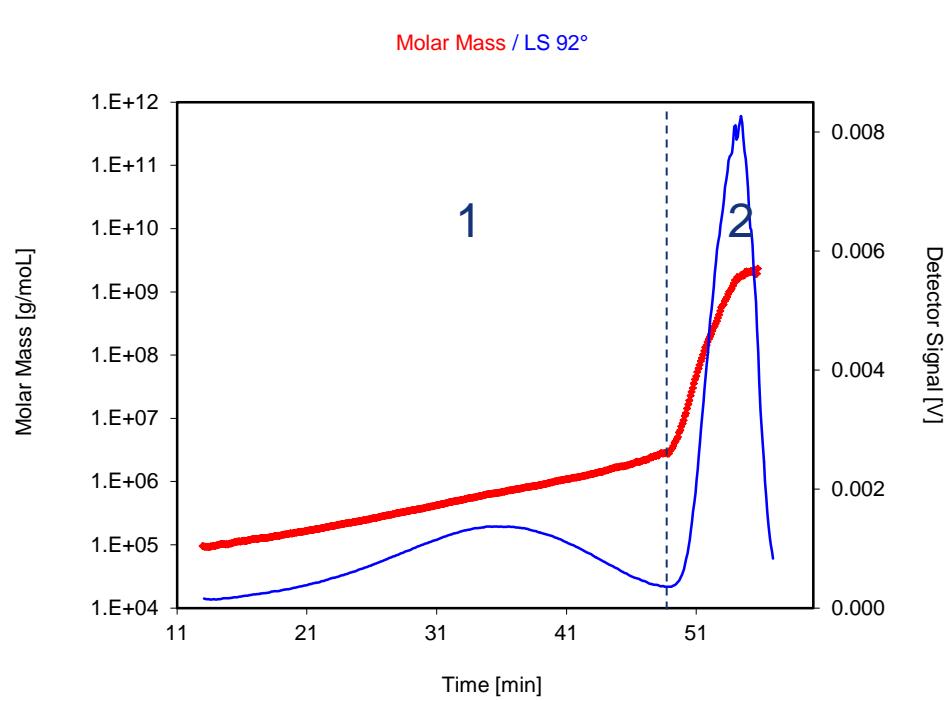
- PN5300 Auto Injector
- TF2000 Thermal FFF
- ELSD
- PN3621 MALS (92° SLS)

Conditions

- Injection Volume: 100 µL
- Concentration: 2.0 mg/mL
- LS 92°(red trace)
- ELSD (blue trace)

Overlay: Molar Mass and LS Signal

- The molar mass was calculated from MALS and ELSD data
- Literature value for $d_n/dc = 0.124 \text{ mL/g}$
- In the 1st peak the sample contains rubber material with a molar mass of appr. $4.7 \times 10^5 \text{ g/mol}$ (w-average) and in the 2nd peak of $3.6 \times 10^8 \text{ g/mol}$.



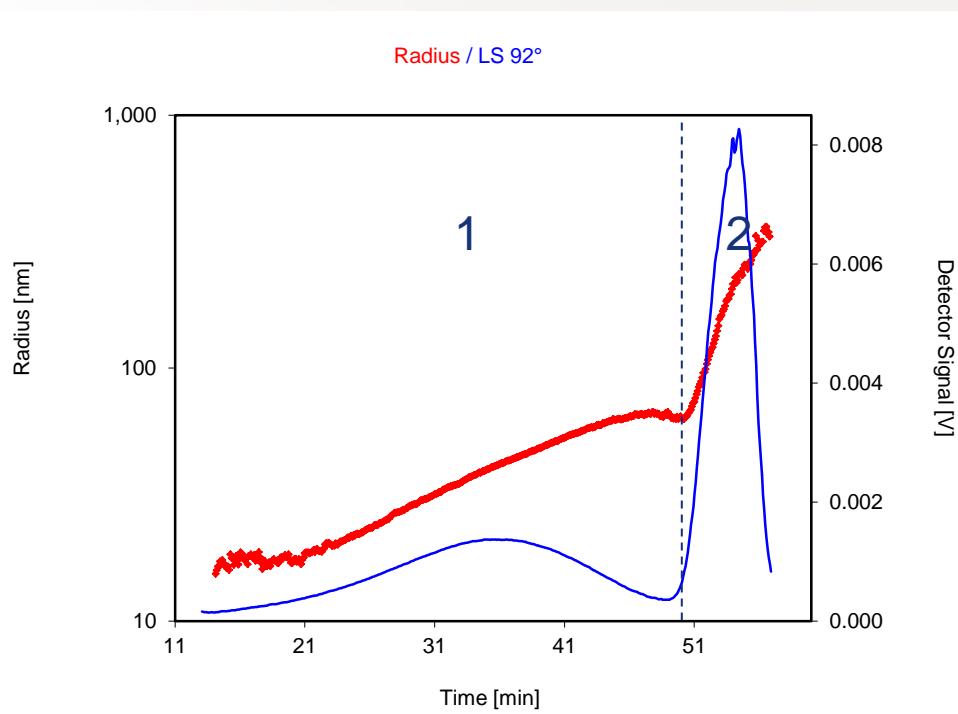
Conditions

- Molar Mass (red dots)
- LS Signal (blue trace)
- Fitting by Random Coil Model

		$M_w[\text{g/mol}]$
1. Peak 13.0 – 48.5 min	<i>n</i> -Average	2.7×10^5
	w-Average	4.7×10^5
	<i>z</i> -Average	8.1×10^5
2. Peak 48.5 – 56.0 min	<i>n</i> -Average	8.1×10^6
	w-Average	3.6×10^8
	<i>z</i> -Average	1.5×10^9

Overlay: Radius of Gyration and LS Signal

- The Radius of Gyration was calculated from MALS angular data
- The main/1st peak shows a Radius of Gyration of 42 nm (z-average) and the 2nd peak a Radius of Gyration of 218 nm.

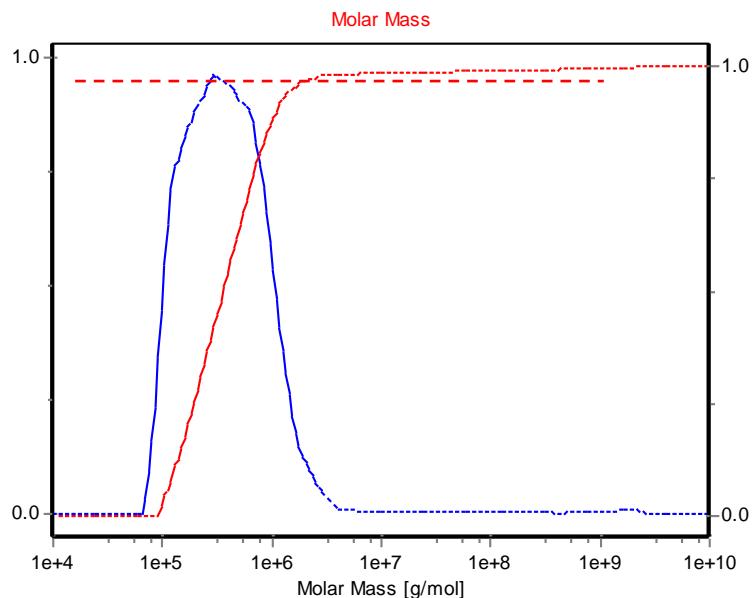


Conditions

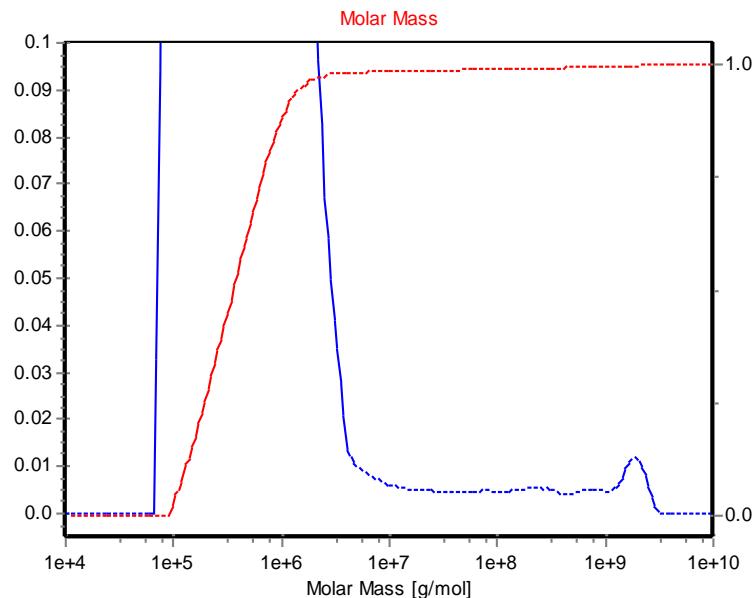
- Radius of Gyration (red dots)
- LS Signal (blue trace)
- Fitting by Random Coil Model

		R_g [nm]
1. Peak	<i>n</i> -Average	23
	<i>w</i> -Average	31
	z-Average	42
2. Peak	<i>n</i> -Average	65
	<i>w</i> -Average	107
	z-Average	218

Molar Mass Distribution

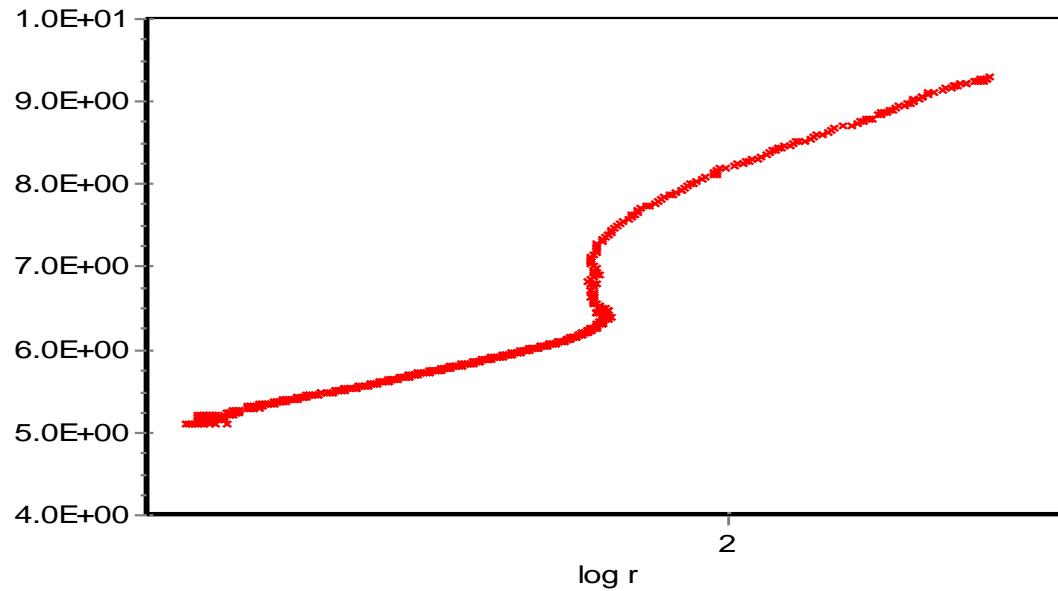


Zoom in Molar Mass Distribution



The sample shows a multimodal distribution. For the 1st fraction (13 – 48.5 min) the distribution is in the range of 9.2×10^4 – 2.8×10^6 g/mol and for the 2nd fraction (48.5 – 56 min) in the range of 2.7×10^6 – 2.3×10^9 g/mol. The sample contains 1.3 % of high molar mass material (gel). Calculation based on concentration detector signal.

Fractal Dimension



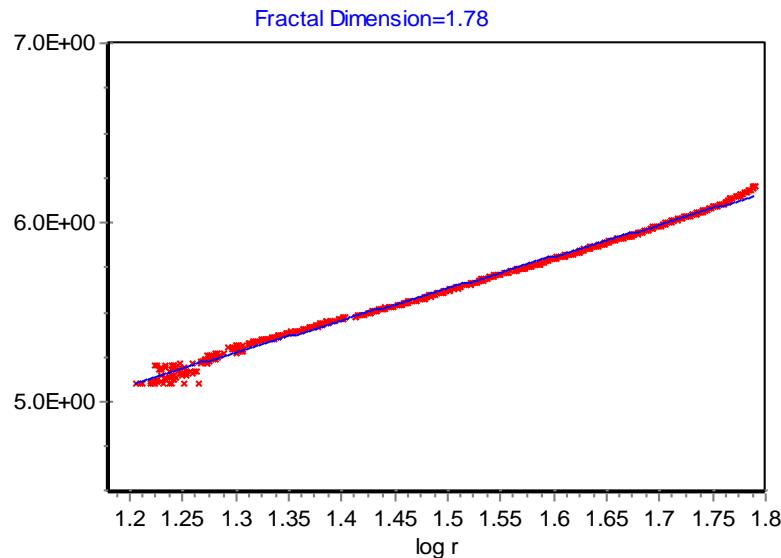
System

- PN5300 Auto Injector
- AF2000 FFF System
- PN3621 MALS Detector
- PN3150 RI Detector
- PN3510 ELSD

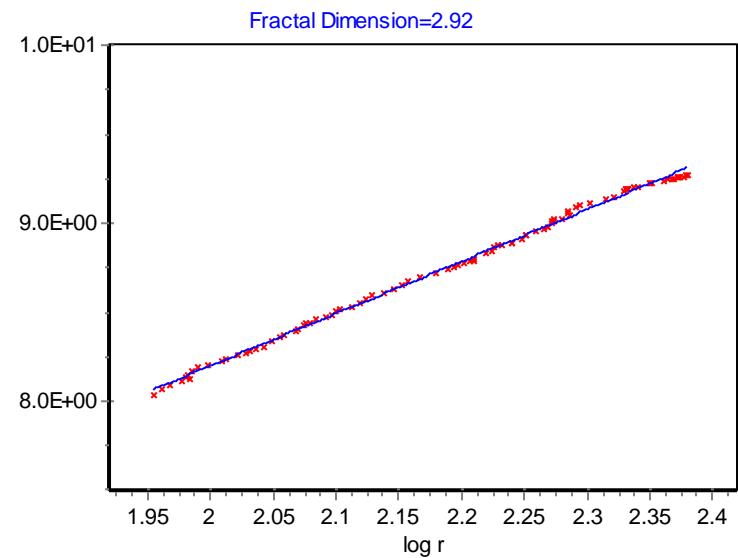
Conditions

- Injection Volume: 20 μL
- Concentration: 2 mg/mL

Fractal Dimension 17.5 - 44.6 min



Fractal Dimension 51.6 – 54.8 min



The Fractal Dimension was evaluated for 2 different regions.

For the 1st region of the fractogram (17.5 – 44.6 min) the Fractal Dimension is 1.78 and for the 2nd region (51.6 – 54.8 min) 2.92.

The Fractal Dimension change indicates that in the 2nd part of the fractogram there is material with a higher density, indicating cross linking or branches.

Thermal FFF is a Powerful Method for Polymer and Biopolymer Characterization

- Increased resolution for high molar mass species
- Huge flexibility in choice of fractionation power (gradient)
- No interaction with stationary phase / no artifacts
- No degradation due to shear stress during the separation
- No filtration by stationary phase or frits
- Separation according to molecular weight and size as well as according to chemical composition offers new horizons for further applications

Thank you for your Attention