

# Trace sulfur and hydrocarbon contaminants in beverage grade carbon dioxide

## **Application Note**

Food Testing & Agriculture

### Introduction

Beverage grade carbon dioxide is used in large quantities in the manufacture of carbonated beverages, mostly soft drinks. The purity of carbon dioxide used in the production of soft drinks by a major manufacturer in Belgium and France was questioned when a number of people became sick after consuming the beverage. While no particular chemical or biological cause was determined, the importance of carbon dioxide purity was highlighted.

Sulfur gases and certain hydrocarbons have been targeted for determination at very low levels in carbon dioxide. The sulfur gases include hydrogen sulfide, carbonyl sulfide, and sulfur dioxide. The hydrocarbons include acetaldehyde, benzene, and light hydrocarbons.

A two tasking GC system capable of analyzing the target compounds were developed: 1) a pulsed flame photometric detector (PFPD) used for the analysis of the sulfur gases down to about 0.1 ppm levels, and 2) trace hydrocarbons (<100 ppbv) in carbon dioxide determined by preconcentration using the sample preconcentration trap (SPT).



### Authors

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#### **Experimental**

A schematic of the system is shown in Figure 1. For sulfur gas analysis a 100 µL sample is introduced directly onto the capillary column. Permeation tubes were used to generate the sulfur gas standards with carbon dioxide as diluent gas.

For the determination of the trace organics, approximately 100 mL of the sample is drawn through the SPT trap. The trapped organics are then thermally desorbed to the capillary column where they are chromatographed. Dynamic dilutions of bottled gas standards served as the sources of hydrocarbon standards in carbon dioxide. Chromatographic parameters for these two applications are shown in Tables 1 and 2:

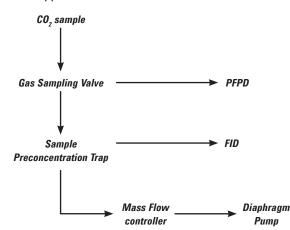


Figure	1: Se	chematic	of	beverage	grade	CO, system
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Sample	100 uL	
Column 1	Agilent CP-SilicaPLOT, 0.32 mm x 30 m, Part no. CP8567	
Carrier	Helium, 2 mL/min; 15 psi (T-4 EFC)	
Column Oven	30/10 min, 20 °C/min to 200 °C/10 min	
	200 °C, S filter, R647 PMT	
PFPD	H <sub>2</sub> : 13 mL/min Air 1: 17 mL/min Air 2: 10 mL/min	

Table 1: Sulfur Gases

Sample	100 mL	
Column	Agilent PoraPLOT Q, 0.53 mm x 25 m, Part no. CP7554	
Carrier	Helium, 2 mL/min; 15 psi (T-4 EFC)	
Column Oven	olumn Oven 50 °C/6 min, 20 °C/min to 200 °C/20 min	
SPT	Trap: Tenax TA (60/80)	
	Trap: 30 °C; Desorb: 190 °C	
FID	250 °C, Range 12	

**Table 1: Hydrocarbons** 

#### **Results**

#### **Sulfur Gases**

All tubing exposed to sample was constructed of Silcosteel to prevent adsorption of sulfur components, particularly  $H_2S$ , on tubing walls. Sample volumes greater than 100 µL were found to overload the Agilent CP-SilicaPLOT column resulting in split, broadened, or non-Gaussian peaks, therefore a sample volume of 100 µL was used. Figure 2 shows a chromatogram of approximately 0.2 ppm of COS,  $H_2S$ , and  $SO_2$  in  $CO_2$ .

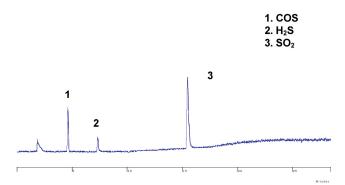


Figure 2: Sulfur Gases in Carbon Dioxide at 0.2 ppm

The peak area and retention time precision at 1.0 ppm were measured and are shown in Table 3.

Compound	Precision (n=6)				
	Area Counts %RSD	Retention Time %RSD			
COS	1.4	0.3			
H <sub>2</sub> S	5.2	0.1			
SO <sub>2</sub>	6.5	0.03			

Table 3: Measurement precision for Sulfur Gases, 1.0 ppm

#### Hydrocarbons

Measuring low levels of hydrocarbons in carbon dioxide is accomplished with the Sample Preconcentration Trap. The molecular weight of the target hydrocarbon dictates what trapping adsorbent is required. If acetaldehyde and simple aromatics are to be determined, then a Tenax TA adsorbent is adequate. If lower molecular weight components down to  $C_2$  hydrocarbons need to be monitored, then a multiphase trap such as the three phase trap, Tenax TA/Carbotrap B/ Carbonsieve S-III may be used. Ambient temperature is sufficient to trap all components.

Preconcentration of a 100 mL sample was used to obtain the required sensitivity. Figure 3 is a chromatogram of four aromatics from a 50 ppbv standard.

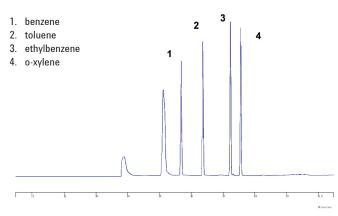


Figure 3: Preconcentration of 50 ppbv aromatics from CO<sub>2</sub>. Column: Agilent CP-Sil 5 CB, 0.53 mm x 30 m, 1.5 µm, Part no. CP8735.

Measurement precision for the aromatics for six runs averaged 1.2% for area and 0.02% for retention time.

Acetaldehyde is another target analyte in beverage grade CO<sub>2</sub>. It is trapped by the Tenax TA and injected as shown in Figure 4.

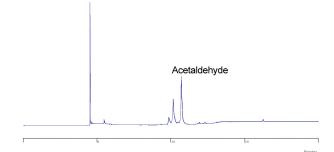


Figure 4: Preconcentration of 0.5 ppmv acetaldehyde from  $CO_2$ . Column: Agilent PoraPLOT Q, 0.53 mm x 30 m, Part no. CP7554.

#### Conclusion

Trace sulfur and hydrocarbon contaminants may be measured in beverage grade  $CO_2$  with a single gas chromatograph. Sulfur compounds are measured at levels down to approximately 0.05 to 0.1 ppm by introducing a small sample directly to a PLOT silica column and PFPD via UltiMetal deactivated or coated tubing.

Trace level hydrocarbons, in this case aromatics and acetaldehyde, are determined by preconcentrating a 100 mL sample on a preconcentrating trap followed by thermal desorption to a capillary column. Detection levels down to less than 1 ppbv are possible.

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