

## Effect of hydrogen carrier gas on Py-GC/MS analysis of polymers Part 2 Hydrogenation of HDPE pyrolyzates in El source of MS

**[Background]** The previous Technical Note (PYA4-007E) revealed that the pyrolysis (Py)-GC/MS of high density polyethylene (HDPE) in a hydrogen ( $H_2$ ) atmosphere altered the peak intensity ratios of diolefin ( $C_n$ "), monoolefin ( $C_n$ ") and alkane ( $C_n$ ) from those obtained in a helium (He) atmosphere. The difference is attributed to the hydrogenation of unsaturated bonds formed during the pyrolysis process. This note examines the differences in the HDPE pyrolyzates formed when the pyrolysis is performed in either a He or a  $H_2$  atmosphere.

**[Experimental]** 200  $\mu$ g of HDPE powder was placed in a deactivated stainless steel sample cup (i.d.=4 mm, H=8 mm) and pyrolyzed at 600°C. The analytical system used for the Py-GC/MS analysis consisted of a Multi-Shot Pyrolyzer (EGA/PY-3030D) with an Auto-Shot Sampler (AS-1020E) interfaced directly to the split injector of a GC/MS system. An Ultra ALLOY® metal capillary column (UA+-5) was used as a separation column. The flow rate of He or H<sub>2</sub> carrier gas was 1 mL/min.

[Results] Pyrograms of  $C_{14}$  region of HDPE pyrolyzates along with MS spectra of  $C_{14}$  are shown in Fig. 1. Note that because  $H_2$  has a larger linear velocity in the column, the retention times for  $C_{14}$  peaks in  $H_2$  atmosphere are smaller than those in He. When the pyrolysis is done in a He atmosphere the peak intensity for the [M+2]+ (m/z 198) ion is extremely low. While, when the pyrolysis is done in a  $H_2$  atmosphere, the [M+2]+ peek is easily identified. This may be because the unsaturated bonds present in the pyrolyzates are partly hydrogenated during the separation process or on the surface of the MS ion source itself. Consequently, the peak intensity ratios observed in the two gases are significantly different. This work clearly illustrates that when  $H_2$  is used as a carrier gas, the hydrogenation of unsaturated pyrolyzates may occur.<sup>1)</sup>

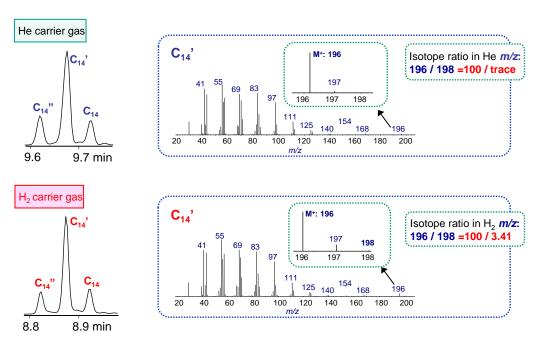


Fig. 1 Pyrograms of C<sub>14</sub> region of HDPE pyrolyzates and MS spectra of C<sub>14</sub>

Pyrolysis Temp.:  $600^{\circ}$ C, GC oven temp.: 40 (2 min) – 320  $^{\circ}$ C (20  $^{\circ}$ C/min, 3 min hold) Separation column: UA\*-5 (5% diphenyl 95% dimethylpolysiloxane), L=30 m, i.d.=0.25 mm, df=0.25  $\mu$ m Column flow rate: 1 mL/min, Split ratio: 1/100, Sample: HDPE (200  $\mu$ g)

1) Reproduced from A. Watanabe, et al., Anal. Chem., 88 (2016) 5462-5468

Keywords: H<sub>2</sub> carrier gas, saturated/unsaturated hydrocarbons, hydrogenation, Py-GC/MS, HDPE, MS ion source, isotope ratio

Products used: Multi-functional pyrolyzer, Auto-Shot Sampler, UA+-5

**Applications**: General polymer analysis

Related technical notes: PYA4-007E, PYA4-009E

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