

# Intermetallic Compounds

An **intermetallic** (also called an **intermetallic compound**, **intermetallic alloy**, **ordered intermetallic alloy**, and a **long-range-ordered alloy**) is a type of metallic alloy that forms an ordered solid-state compound between two or more metallic elements. Intermetallics are generally hard and brittle, with good high-temperature mechanical properties. Ref: [Wikipedia](#)

YbSi      PdIn      Mg<sub>2</sub>Ge      PtAuSb<sub>4</sub>      PtPd      BaZn<sub>2</sub>Ge<sub>2</sub>

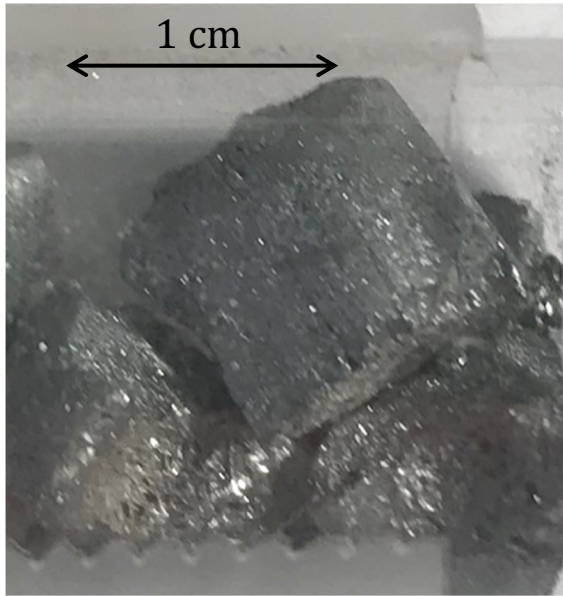
EuPd<sub>2</sub>Si<sub>2</sub>      PdIn<sub>2</sub>Se<sub>4</sub>      NdAsSe      AgSbTe<sub>2</sub>      Ce<sub>2</sub>SeTe

KInS<sub>2</sub>      Re<sub>2</sub>Te<sub>5</sub>      Eu<sub>3</sub>Sb<sub>2</sub>      Eu<sub>3</sub>P<sub>2</sub>      Nd<sub>3</sub>S<sub>4</sub>

The prices on the products are available upon request and depend on various factors such as cost of the production, quality, dimensions, etc.

CRYST<sup>+</sup>MAT request: [nzhigadlo@gmail.com](mailto:nzhigadlo@gmail.com)

## YbSi



Optical image of YbSi sample

YbSi belongs to the orthorhombic CrB-type structure with a space group  $Cmcm$ .

YbSi is the **heavy-electron antiferromagnet** compound. It shows a second-order **magnetic ordering transition** of the Yb moments **below 1.6 K**.

## InPd

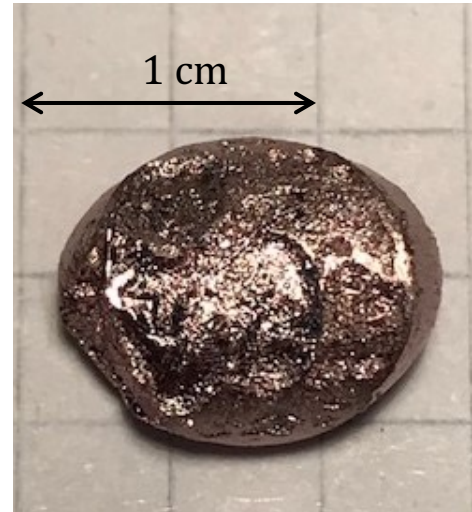
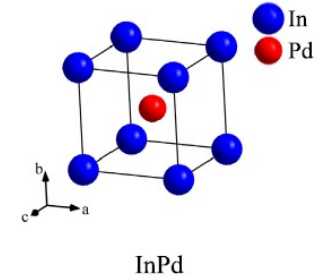


Image of InPd melted sample



InPd crystallizes in a primitive cubic crystal structure of the CsCl (B2) type with a space group  $Pm-3m$  (No 221).

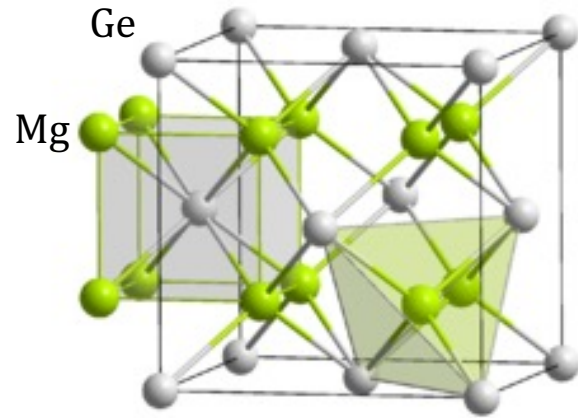
**InPd is a predominant hole-type conductor.**

Ref.: [Intermetallics 55 \(2014\) 56-65](#)

# Mg<sub>2</sub>Ge



Optical image of Mg<sub>2</sub>Ge crystals



Mg<sub>2</sub>Ge crystallizes in the cubic structure with a space group *Fm-3m*

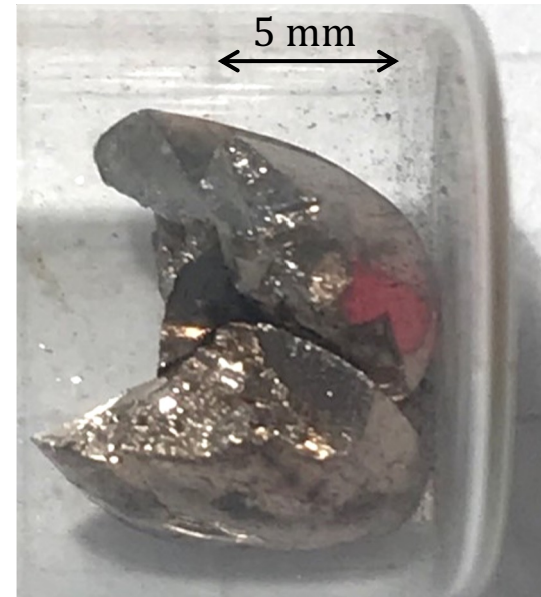
Mg<sub>2</sub>Ge belongs to the class of Zintl phases with a relatively large electronegativity difference of elements. In the case of Mg<sub>2</sub>Ge, four valence electrons are transferred from Mg to Ge. This leads to 4+4=8 valence electrons at the Ge atom.  
Ref.: [J. Alloy Comp. 235 \(1996\) 250](#).

PtAuSb<sub>4</sub>



Optical image of PtAuSb<sub>4</sub> sample

PtPd

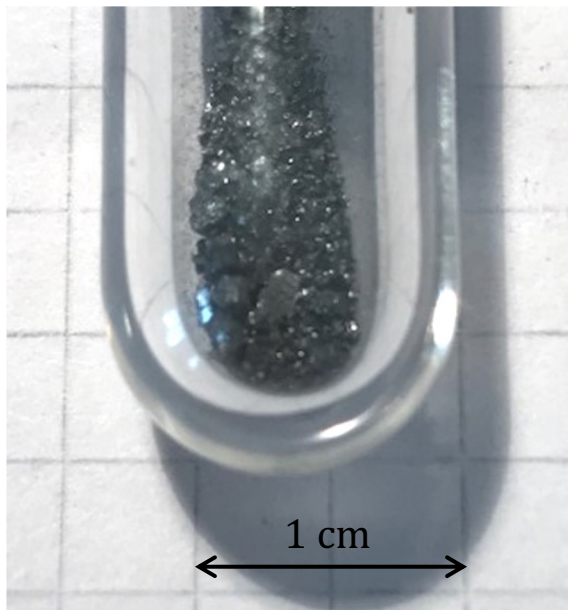


Optical image of PtPd alloy

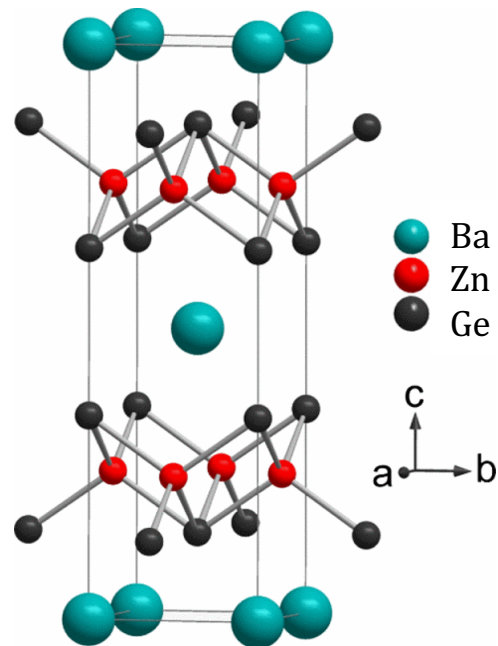
By employing first-principles calculations it was **predicted the existence of type-II Weyl phonons in ZnSe**, a well-known II-VI semiconductor.

Ref: [Phys. Rev. B 103, 094306 \(2021\)](#)

# BaZn<sub>2</sub>Ge<sub>2</sub>



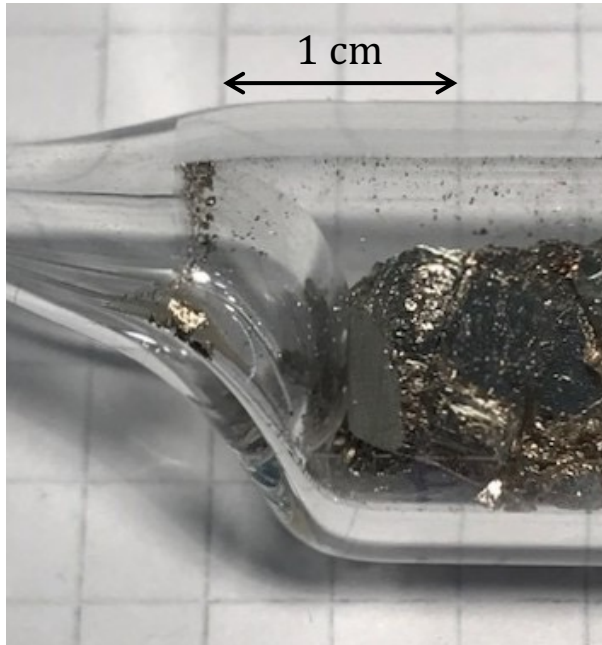
Optical image of BaZn<sub>2</sub>Ge<sub>2</sub> sample



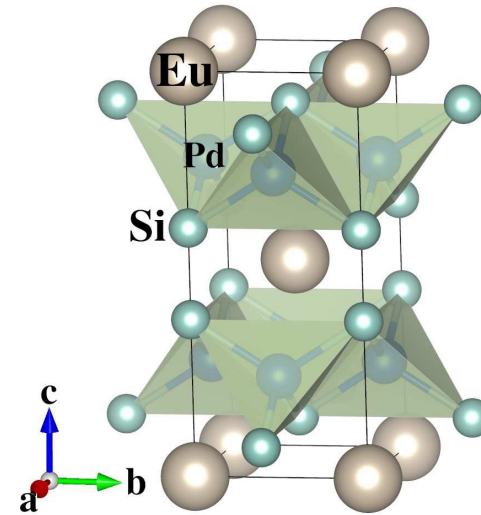
BaZn<sub>2</sub>Ge<sub>2</sub> crystallizes in the ThCr<sub>2</sub>Si<sub>2</sub> structure (tetragonal,  $I4/mmm$ ;  $a = 5.53 \text{ \AA}$ ,  $c = 10.55 \text{ \AA}$ ,  $Z = 2$ ).

Ref.: [Chem. Mater. 9 \(1997\) 1463-1466](#)

# EuPd<sub>2</sub>Si<sub>2</sub>



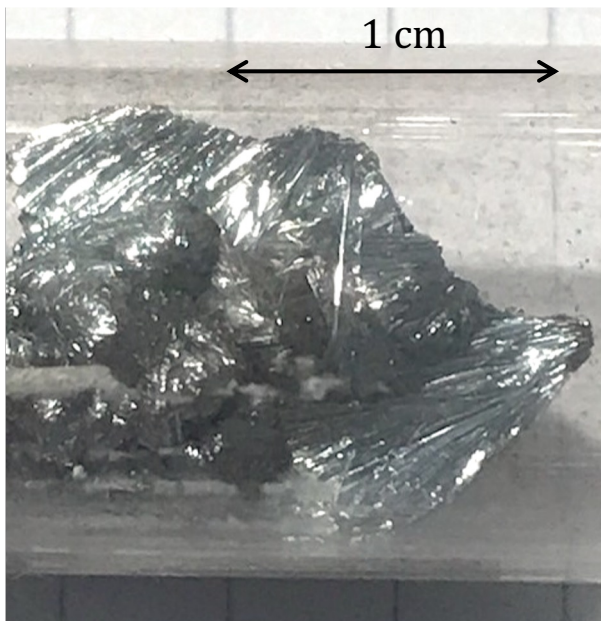
Single-crystalline ingot of EuPd<sub>2</sub>Si<sub>2</sub>



EuPd<sub>2</sub>Si<sub>2</sub> crystallizes in a tetragonal body-centered ThCr<sub>2</sub>Si<sub>2</sub>-type structure with space group *I*4/*mmm* (No. 139). It consists of layers PdSi<sub>4</sub> tetrahedra intercalated between Eu planes.

EuPd<sub>2</sub>Si<sub>2</sub> is the **mixed-valence compound** that exhibits a valence transition from Eu<sup>2+</sup> to Eu<sup>3+</sup> around 160 K.





Single-crystalline ingot of PdIn<sub>2</sub>Se<sub>4</sub>

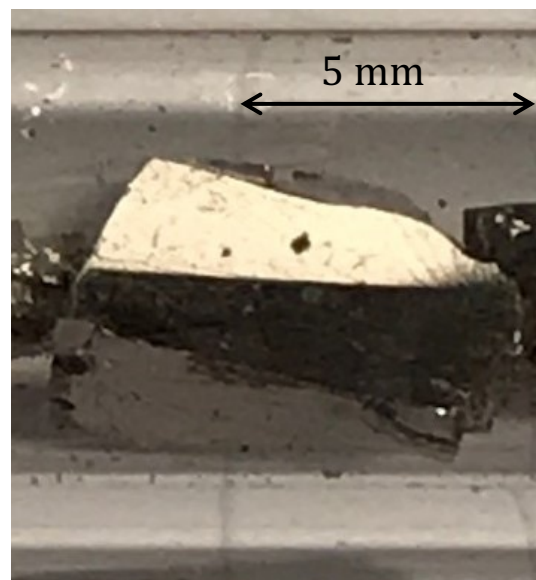
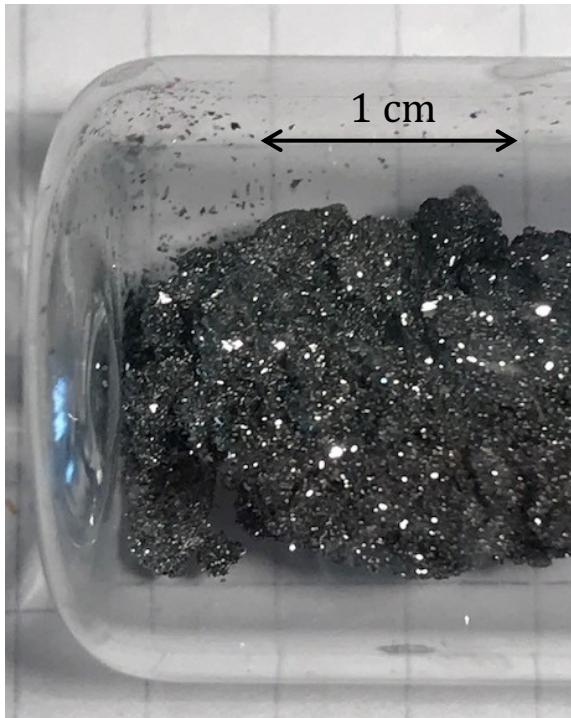


Image of Ce<sub>4</sub>(La<sub>0.76</sub>Y<sub>0.24</sub>)<sub>4</sub>B sample

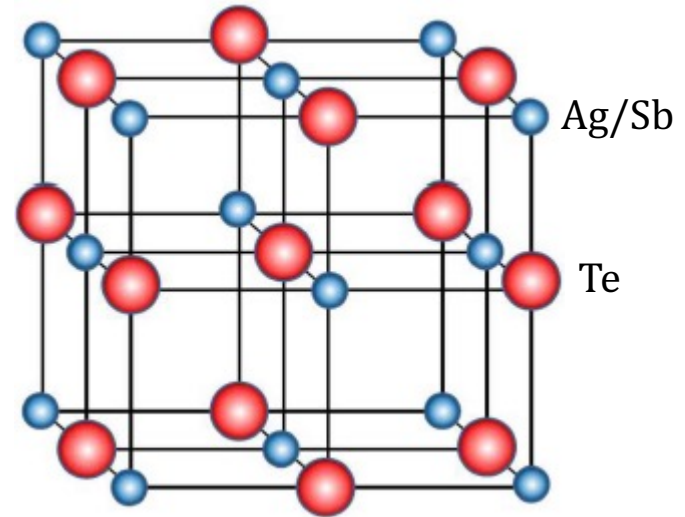
## NdAsSe



Optical image of NdAsSe sample

NdAsSe crystallizes in the monoclinic CeAsS type structure.

## AgSbTe<sub>2</sub>

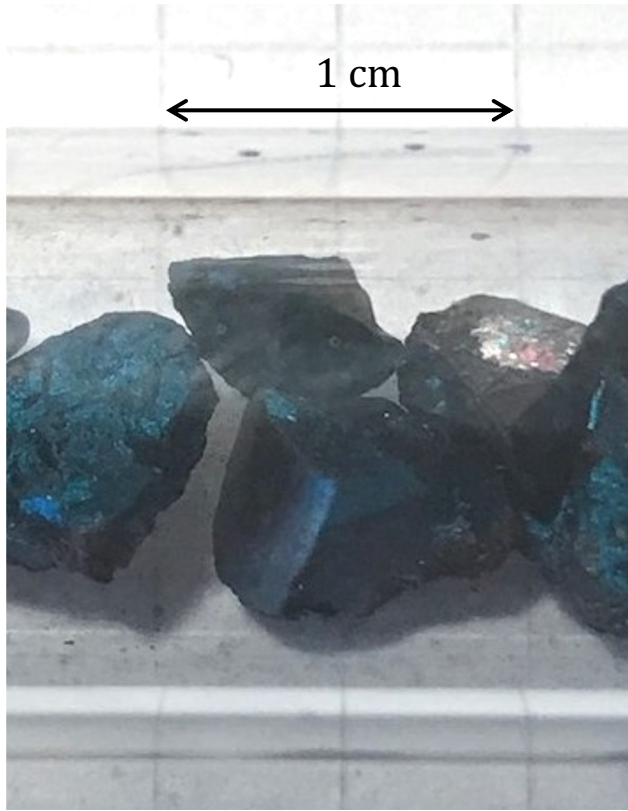


AgSbTe<sub>2</sub> crystallizes in a NaCl-type structure (cubic, Fm-3m, 225), where Ag and Sb randomly occupying the Na site whereas Te is located at the Ce position.

AgSbTe<sub>2</sub> is a promising **thermoelectric** material. The **band gap** is **~0.35 eV** at room temperature.

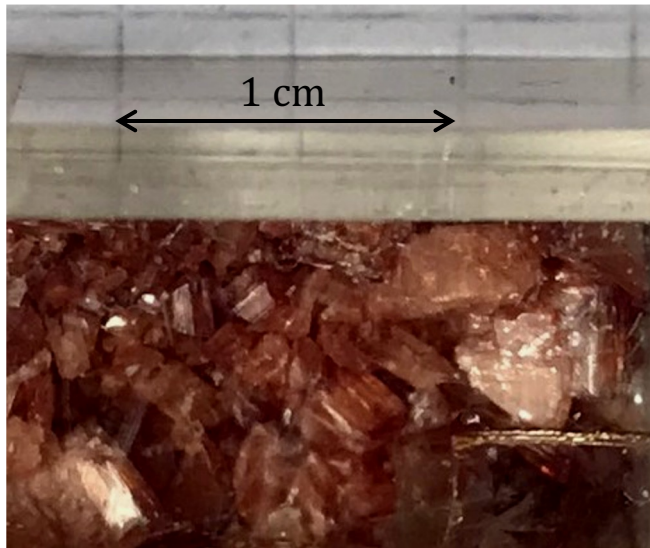


# $\text{Ce}_2\text{SeTe}$

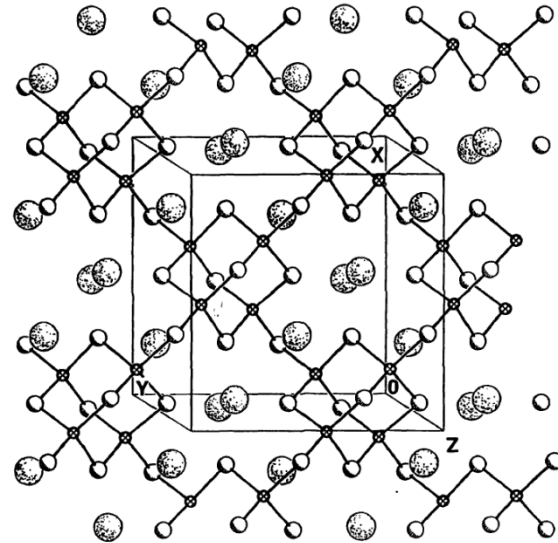


Optical image of  $\text{Ce}_2\text{SeTe}$  samples

# KInS<sub>2</sub>



Optical image of KInS<sub>2</sub> samples



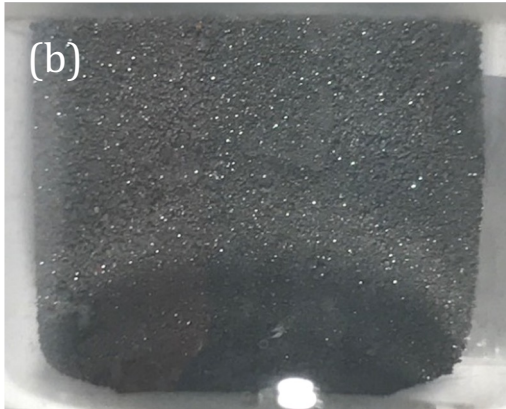
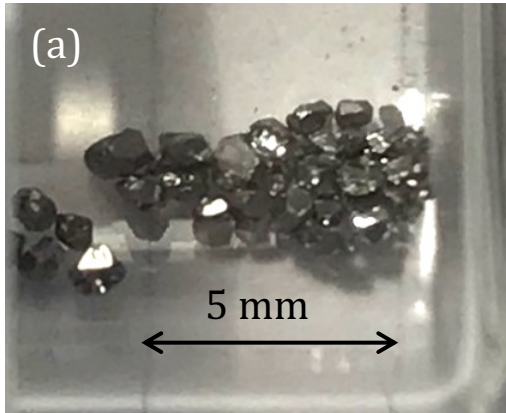
KInS<sub>2</sub> crystallizes in the monoclinic space group C2/c. The bonding is highly covalent and exhibits both 2D and 3D features.

**Limited information on the physical properties** of KInS<sub>2</sub> is available.

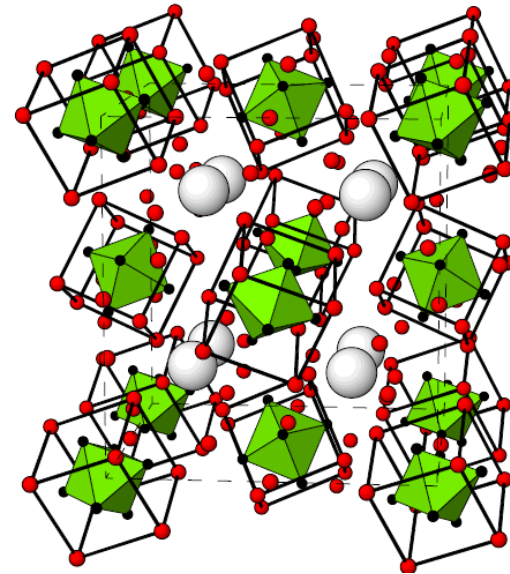
It has a band-gap of 3.0 eV.

Ref.: [J. Solid State Chem. 92, 2, 520-530 \(1991\)](#)

# $\text{Re}_2\text{Te}_5$



Images of  $\text{Re}_2\text{Te}_5$  samples:  
(a) crystalline, (b) powder.



$\text{Re}_2\text{Te}_5$  crystallizes in the orthorhombic structure with space group  $Pbca$  (No 61). The Re cluster surrounded by Te atoms. The large spheres represent the atoms that can possibly be inserted in these voids.

Ref.: [Proc. XVII Int. Conf. Thermoelec. Japan, May 24-29 \(1998\) 298](#)

$\text{Re}_2\text{Te}_5$  is a **semiconducting compound** with an energy gap of  $\sim 0.8$  eV.



Optical image of Eu<sub>3</sub>Sb<sub>2</sub> sample



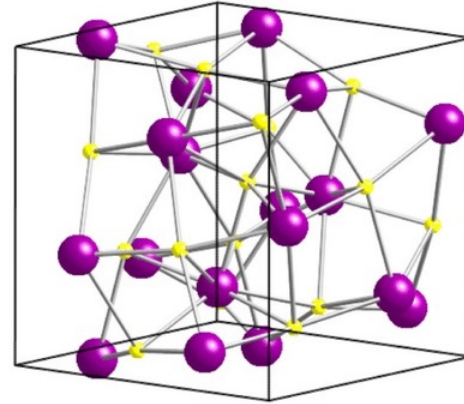
Optical image of Eu<sub>3</sub>P<sub>2</sub> sample

$\text{Eu}_3\text{P}_2$  is **ferromagnetic semiconductor** with Curie temperature of 25 K.

$^{151}\text{Eu}$  Mössbauer spectra of  $\text{Eu}_3\text{P}_2$  reveal that the compound contains both  $\text{Eu}^{2+}$  and  $\text{Eu}^{3+}$  ions. Above 150 K thermally activated electron hopping between these ions is observed. Ref.: [J. Phys. Chem. Solids 55, 2 \(1994\) 219-227](#)



Optical image of  $\text{Nd}_3\text{S}_4$  sample



$\text{Nd}_3\text{S}_4$  adopts the cubic symmetry with space group  $I-43d$  (No. 220)

Among the many Nd-S compounds only  $\text{Nd}_3\text{S}_4$  order ferromagnetically with  $T_c = 47$  K.