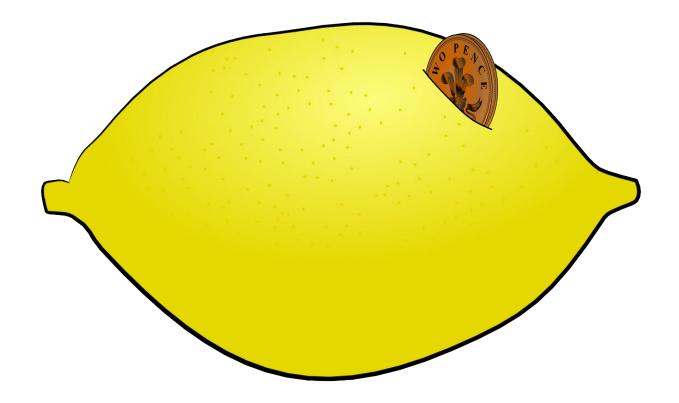


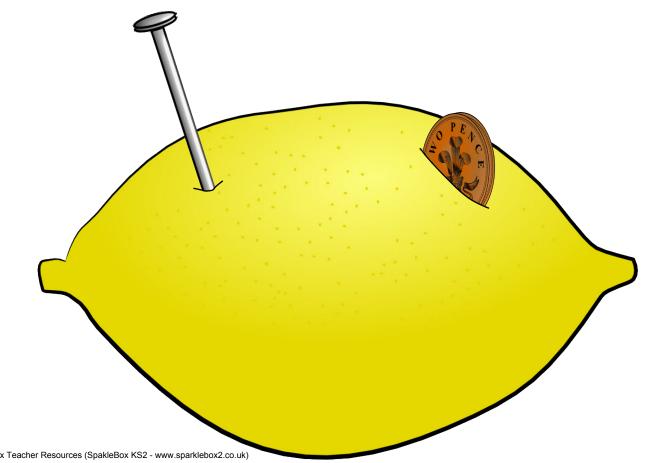


# Insert a coin into one side of the lemon.



Step 2

Push one of the nails into the other side of the lemon. Make sure that the coin and nail **do not touch**.

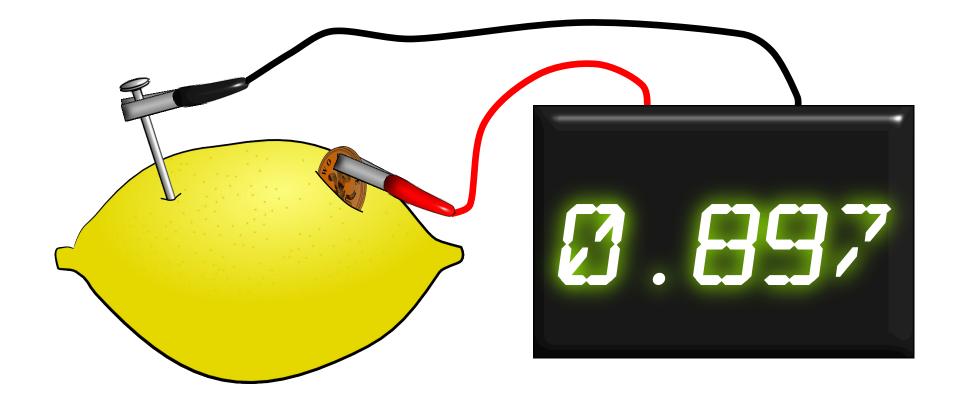


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This is one single battery cell. The copper coin and zinc nail act as electrodes and the juice in the lemon is the electrolyte.

Step 3

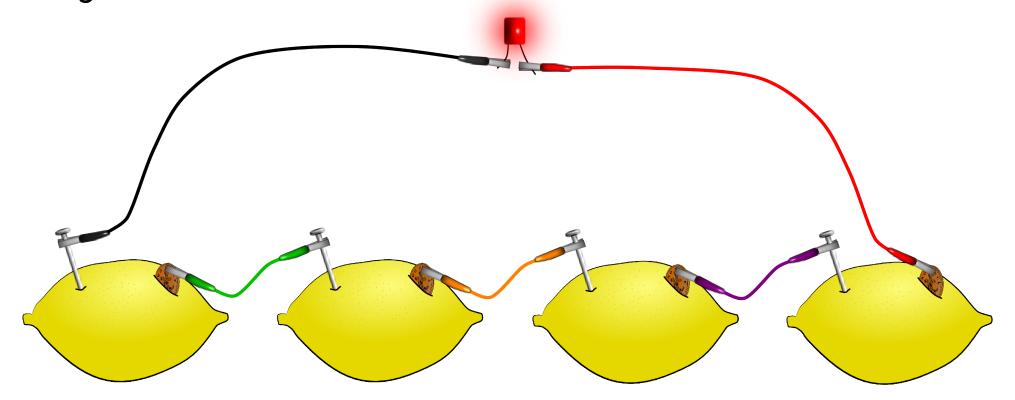
You can check that the lemon battery cell is producing a voltage by connecting a volt meter to the electrodes.



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# Step 4

To power a small bulb, you will need to create more battery cells. You can connect them together to create a higher voltage and current.



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