

## Chemical Hazardous Waste. Introduction

Hazardous waste is waste that, when present in quantities and concentrations that are high enough, pose a threat to human health or the environment if they are improperly stored, transported, treated or disposed. Details on this can be found within the O. Reg 437: General-Waste Management.

As part of the cradle-to-grave management system of the MOECP, hazardous waste requires special handling with respect to how it is collected, stored, transported, treated, recovered and disposed to reduce adverse effects to human health and the environment.

At Brock, Science Stores staff coordinate the chemical hazardous waste collection. An external contactor licensed to haul and dispose of hazardous waste collects said waste from the labs in the company of the Brock staff.

Lab users that cause the generation of hazardous waste shall follow the procedures outlined below in alignment with the safety and environmental protection legislation that apply (Environmental Protection Act, O. Reg 437: General- Waste Management). In following these procedures, the handling and disposal of hazardous waste can be made safely.

Most questions regarding when and how waste is collected and what type of resources are available can be responded by the sciences stores staff. In addition, staff from HSW who oversee these activities and can also respond any questions or concerns.

#### Contacts info:

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## Generation and Labelling

- Minimize volume and hazard upon ordering, when possible.
- Under no circumstances will hazardous chemicals/wastes be disposed of down a sink. Chemical wastes sent to the drain may react and cause significant damage which may result in loss of drains, impeding lab activities until they can be replaced. In addition, they may pollute the environment.
- Plan for weekly waste disposal. Designate a place for the waste. If a fume hoods is used for this, that hood cannot accommodate any other activity/work.
- Plan for safe segregation of waste by compatibility and reactivity before beginning an
  experiment. Incompatible and highly reactive chemicals must always be kept
  separately.
- Minimum segregation groups include organics vs inorganic, acids vs bases, halogenated vs non-halogenated. Segregate further according to below.





For flammables-1 gal



For flammables-2 gal



For flammables-5 gal



White, for nonflammablesvarious sizes

Figure 1. Waste containers for liquid hazardous waste.

- Use safety data sheets to determine segregation groups and/or the <u>compatibility chart</u> found at the end of this document or the EPA Combability Chart available online.
- Two types of containers (Fig. 1) are available for liquid hazardous waste: red, hard polypropylene (PP) spring-lidded safety containers with flame arrestor; and white, semitransparent, PP containers with screw cap;
  - Use the Red containers are for flammable waste.
  - Use the screw cap plastic white containers for aqueous waste solutions, not for flammable liquid waste, unless the

flammable is also very corrosive.

- Check the containers before using them. They
  must be intact and suitable to contain the
  chemical, and capable of a tight closure to
  prevent leaks during transport.
- Discard containers with a deteriorated flame arrestor, without a lid/cap or with a compromised integrity.
- Obtain the Yellow Chemical Waste Label (Fig. 2) sticker from Science Stores (MC E303) and affix it to every waste container prior to start adding wastes in.
  - Fill out all the parameters that asked for on the label.
  - Do not use abbreviations or short forms.
  - Represent the waste properly, i.e. if your mixture results in a product, list the product of the reaction and not individual reactants. This may have significant safety or legal implications.

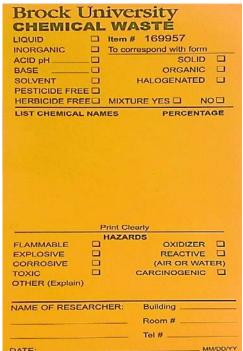


Figure 2. The **Yellow Chemical Waste Label**.



- Containers must be filled to the shoulder only. NEVER fill them completely. It may lead to leaks.
- Always maintain all containers closed except when pouring in, even if they are inside a hood. Vapors or fumes emanating unnecessarily pollute the environment or the lab air depending on where the container is situated.
- While containers must be kept closed, do not seal screw cap closures until the day of collection from the lab to avoid any pressure buildup.
- Screw cap containers must be closed tightly the day of collection from the lab to prevent leakage during its transport.
- When little waste is foreseen for the week, use the smallest container available to avoid carrying waste over to the following week.
- When preparing the waste for pickup, check that containers are not overfilled or overflowing. Containers that are overflowing or filled past the shoulder line will not be collected under regular pickup procedures. Since this may carry a safety implication or cause a spill cleanup surcharge ENSURE proper fill.
- Metal and glass containers are not permitted for hazardous waste, unless extraordinary situations exist. If that is the case, use a secondary container to catch any spill in the event of a breakage.



Figure 3. PP pail with lid for silica waste.

- When additional or replacement containers are needed, ask the Science Stores staff.
- For silica waste ask the Science Stores staff for a container. White, PP, wide mouthed 20 L pails with a lid (Fig. 3) are available for this type of waste.

## Hazardous waste registration and pick up from labs

For the waste to be picked up from your lab, the sciences stores staff must be notified.

- Use the Chemical Waste Disposal Record form (Fig. 6) to request a pickup. The form is available from Science Stores and can be downloaded at <a href="http://brocku.ca/mathematics-science/departments-and-centres/sciencestores/waste-disposal">http://brocku.ca/mathematics-science/departments-and-centres/sciencestores/waste-disposal</a>
- All containers which are filled to the shoulder level must be disposed every week. This contributes to your lab being compliant with the allowable limit of flammable liquids and to safety by reducing the existence of hazardous products.
- Each container registered for disposal must list the number which corresponds to the item # located on the upper right of the Yellow Chemical Waste Label.
- Submit the completed form to Science Stores by Tuesday at 3:00 pm either in person (MC E303), by fax (905 984 4864), or email (sciencestores@brocku.ca).



- External waste technicians and staff from Science Stores will pick up waste every
   Wednesday morning from every lab that registered waste.
- Wastes that are not properly labelled will not be picked up.
- Wastes that are overflowing or present any safety problem will not picked up under regular collection procedures and may be left behind until an arrangement for spill cleanup can be made. Since this carries a surcharge, follow waste procedures strictly to prevent unnecessary expenses.

## Excess solvent cans / other metal containers and glass waste

- Excess/unused clean metal containers from labs are collected to prevent clutter and tripping hazards and are recycled.
- Containers which had hazardous materials must be emptied and triple rinsed. Collect the rinsate as hazardous waste.
- Affix the Safe for Disposal (Fig. 4) label, which can be obtained from Science Stores.
- Place the labelled container in the hallway by the main entry door to your lab on Mondays during office hours.
- Custodial services will pick up containers that meet the above criteria.
- For the disposal of unused clean glass containers or clean pieces of glass, place them in the cardboard box shown in Fig. 5. Contact Custodial Services (CS) to obtain these boxes.



Figure 4. **Safe for Disposal** label for recycling metal containers.



Figure 5. Cardboard box for disposal of clean glass waste.

- When a container had had hazardous materials, it must be emptied and triple rinsed. Collect the rinsate as hazardous waste.
- When the box is ¾ full, close the plastic liner and the lid and write down Please Dispose on the box. CS will collect the box and replace it with a new one.

*Note*: wastes from pyrophoric and other reactive materials must follow special precautions which are not covered here. Contact your Principal Investigator for these.

## Hydrofluoric Acid waste

Hydrofluoric acid (HF) is a very corrosive and toxic substance, for which special precautions apply. If you are new to work with HF, please contact the Lab Safety Specialist at ext 6179 for an assessment of your fume hood and the especial instructions that apply.

Hydrofluoric acid waste is collected in specific type of containers (UN 3H1/Y) different to the ones used for other chemical hazardous wastes and rated for holding HF waste. They are provided by Science Stores. Two types of containers are available.



- For work with small volumes and a concentration of HF of up to 48%, containers are of 2L capacity, made of PE, with a secure lid.
- For work with large volumes and a concentration of HF up to 48%, containers are of 4 L capacity, made of HDPE, with a secure lid.
- Dispose HF waste according to the schedule and registration protocol used for other hazardous waste at the University.
- Containers may only be filled to the shoulder level.
- Waste containers must always be kept tightly closed and clean from any residues.
- If you apply any solution to neutralize residues of HF that there may be on the container surface, the remains of said solution must be cleaned before the waste is offered for disposal to the waste contractor.
- HF waste must not be mixed with any other material.
- Containers with HF waste may not be reused.

END



1	Inorganic Acids	1	]																						
2	Organic acids	х	2	]																					
3	Caustics	х	х	3																					
4	Amines & Alkanol amines	х	х		4	]																	•		
5	Halogenated Compounds	х	Г	х	х	5																			
6	Alcohols, Glycols & Glycol Ethers	Х					6	]																	
7	Aldehydes	х	х	x	х		х	7	]																
8	Ketone	Х		х	х			х	8																
9	Saturated Hydrocarbons									9															
10	Aromatic Hydrocarbons	Х									10	]													
11	Olefins	х				х						11	]												
12	Petroleum Oils												12	]											
13	Esters	х		х	х									13	]										
14	Monomers & Polymerizable Esters	х	х	х	х	х	х								14										
15	PhenoIs	П		х	х			х							х	15									
16	Alkylene Oxides	х	х	х	х		х	х							х	х	16	]							
17	Cyanohydrins	х	х	x	х	х		х									х	17							
18	Nitriles	х	х	х	х												х		18	]					
19	Ammonia	х	х					х	х					х	х	х	х	х		19					
20	Halogens			х			х	х	х	х	х	х	х	х	Х	х				х	20				
21	Ethers	х													х						х	21			
22	Phosphorus, Elemental	х	х	х																	х		22		
23	Sulfur, Molten									Х	х	х	х				х						х	23	
24	Acid Anhydrides	Х		х	х		х	х							х		х	Х	х	х					24
	X Represents unsafe combinations				F	Repre	esent	s saf	fe co	nbir	atio	ns													

Figure 1. Chemical Compatibility Charts and Associated Groups



## Group 1: Inorganic Acids

Chlorosulfonic acid Hydrochloric acid (aqueous) Hydrofluoric acid (aqueous) Hydrogen chloride (anhydrous) Hydrogen fluoride (anhydrous) Nitric acid Oleum Phosphoric acid

Sulfuric acid

## Group 2: Organic Acids

Acetic acid Butyric acid (n-) Formic acid Propionic acid

### Group 3: Caustics

Potassium hydroxide Sodium hydroxide Ammonium hydroxide

## Group 4: Amines and Alkanolamines

Aminoethyl ethanolamine

Aniline

Diethanolamine Diethylenetriamine Diisopropanolamine Dimethylamine Ethylenediamine Hexamethylene diamine Methyl-5-ethylpyridine

Monoethanolamine Monoisopropanolamine

Morpholine Pvridine

Triethanolamine Triethylamine

Triethylenetetramine

Trimethylamine

## Group 5: Halogenated Compounds

Allyl chloride

Carbon tetrachloride

Chlorobenzene

Chloroform

Chlorohydrines

crude Dichlorobenzene (o-)

Dichlorobenzene (p-)

Dichlorodifluoromethane

Dichloroethyl ether

Dichloropropane

Ethyl chloride

Ethylene dibromide

Ethylene dichloride

Methyl bromide

Methyl chloride

Methylene chloride

Monochlorodifluoromethane

Perchloroethylene

Propylene dichloride

1,2,4-Trichlorobenzene

1,1,1-Trichloroethane

Trichloroethylene

Trichlorofluoromethane

## Group 6: Alcohols, Glycols and Glycol Ethers

Allyl alcohol Amyl alcohol

1.4-Butanediol

Butyl alcohol

Butylene glycol

Corn syrup

Cyclohexyl alcohol Decyl alcohol (n, iso)

Dextrose solution

Diacetone alcohol

Diethylene glycol

Diethylene glycol dimethyl ether

Diethylene glycol monobutyl ether

Diethylene glycol monoethyl ether Diethylene glycol monomethyl ether

Diisobutyl carbitol Dipropylene glycol

Dodecanol

Ethoxylated dodecanol Ethoxylated pentadecanol Ethoxylated tetradecanol Ethoxylated tridecanol

Ethoxytriglycol

Ethyl alcohol Ethyl butanol

2-Ethylbuytl alcohol

2-Ethylhexyl alcohol

Ethylene glycol

Ethylene glycol monobutyl ether Ethylene glycol monoethyl ether

Ethylene glycol monomethyl ether

Furfuryl alcohol

Glycerine

Heptanol

Hexanol

Hexylene glycol

Isoamyl alcohol

Isooctyl alcohol

Methyoxytriglycol

Methyl alcohol

Methylamyl alcohol

Molasses, all

Nonanol

Octanol

Pentadecanol

Polypropylene glycol methyl ether \_

Propyl alcohols (n, iso)

Propylene glycol

Sorbitol

Tetradecanol

Tetraethylene glycol

Tridecyl alcohol

Triethylene glycol Undecanol

### Group 7: Aldehydes

Acetaldehyde

Acrolein (inhibited)

Butyraldehyde (n, iso)

Crotonaldehyde

Decaldehyde (n, iso)

2-Ethyl-3-proplyacrolein

Formaldehyde solutions

Furfural

Hexamethylenetetramine

Isooctyl aldehyde

Methyl butyraldehyde

Methyl formal

Paraformaldehyde

Valeraldehyde



### **Group 8: Ketones**

Acetone Acetophenone Camphor oil Cylcohexanone Diisobutyl ketone Isophorone Mesityl oxide Methyl ethyl ketone Methyl isobutyl ketone

#### Group 9: Saturated Hydrocarbons

Butane Cyclohexane Ethane Heptane Hexane Isobutane

Liquified natural gas Liquified petroleum gas

Methane Nonane n-Paraffins Pentane

## ▶ Petrolatum

Benzene

Cumene

Petroleum ethers Butvlene Petroleum naphtha 1-Decene Polybutene Dicylcopentadiene Propane Diisobuytlene Propylene butylene polymer Dipentene Dodecene

## Group 10: Aromatic Hydrocarbons

Liquified petroleum gas p-Cymene 1-Heptene Coal tar oil 1-Hexane Isobutylene Diethybenzene Dodecyl benzene Nonene Dowtherm 1-Octene Ethylbenzene 1-Pentene Naphtha, coal tar Polybutene

Naphthalene (includes molten) Propylene Tetrahydronaphthalene Propylene butylene polymer

Propylene tetramer (dodecene) Toluene 1-Tetradecene Triethyl benzene Xylene (m-, o-, p-) 1-Tridecene Turpentine

#### Group 12: Petroleum Oils

Asphalt Gasolines Jet Fuels & kerosene Mineral spirits Naphtha Oils

Absorption oil Clarified oil Crude oil Diesel oil Fuel oil Lubricating oil Mineral oil Mineral seal oil Motor oil Penetration oil Range oil Road oil Spindle oil Spray Oil

Transformer oil

Turbine oil

#### Group 13: Esters

Amyl acetate Amyl tallate

Butyl acetates (n, iso, sec) Butyl benzyl phthalate

Castor oil Croton oil Dibutyl phthalate Diethyl carbonate Dimethyl sulfate Dioctyl adipate Dioctyl phthalate Epoxidized vegetable oils

Ethyl acetate Ethyl diacetate

Ethylene glycol monoethyl ether

acetate

Ethylhexyl tallate

Fish oil Glycol diacetate Methyl acetate Methyl amyl acetate Neatsfoot oil Olive oil Peanut oil Propyl acetates (n, iso)

## Resin oil

1-Undecene

Soya bean oil Sperm oil Tallow Tanner's oil Vegetable oil Wax, carnauba

Group 11: Olefins

1-Dodecene

Ethylene

## Group 14: Monomers and Polymerizable esters

Acrylic acid (inhibited)

Acrylonitrile

Butadiene (inhibited) Butyl acrylate (n, iso) Ethyl acrylate (inhibited) 2-Ethylhexyl acrylate (inhibited) Isodecyl acrylate (inhibited) Isoprene (inhibited)

Methyl acrylate (inhibited) Methyl methacrylate (inhibited) o-Propiolactone

Styrene (inhibited) Vinyl acetate (inhibited)

Vinyl chloride (inhibited) Vinyl toluene

Vinylidene chloride (inhibited)



Group 15: Phenols

Carbolic oil Creosote, coal tar Cresols Nonylphenol Phenol

Group 16: Alkylene Oxides

Ethylene Oxide Propylene Oxide

Group 17: Cyanohydrins

Acetone cyanohydrin Ethylene cyanohydrin Group 18: Nitriles

Acetonitrile Adiponitrile

Group 19: Ammonia

Ammonium hydroxide

Group 20: Halogens

Bromine Chlorine Group 21: Ethers

Diethyl ether (ethyl ether) 1, 4, Dioxane Isopropyl ether

Tetrahydrofuran

Group 22: Phosphorus,

elemental

Group 23: Sulfur, molten

Group 24: Acid Anhydride

Acetic anhydride Propionic anhydride