Roll No.

(12/24)

15025

M. Sc. EXAMINATION

(For Batch 2021 & Onwards)

(Third Semester)

CHEMISTRY

MSc/Chem/3/DSC3-I

Inorganic Chemistry Special-III

Time: Three Hours

Maximum Marks: 70

P.T.O.

Note: Attempt Five questions in all, selecting one question from each Unit. Q. No. 1 is compulsory.

- 1. (i) Why is borazine often called "Inorganic Benzene"?
 - (ii) Explain the concept of molecular self-assembly.

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- (iii) Why does hemoglobin exhibit a sigmoidal oxygen-binding curve while myoglobin does not?
- (iv) What do you understand by CN and CO poisoning?
- (v) How is iron absorbed and regulated in the human body? $2\times 5=10$

Unit I

- 2. (a) What are the methods of synthesizing arsenic sulfides? Write their applications.
 - (b) What are borazines? Give their methods of preparation and chemical reactions. 5
 - (c) Discuss the factors that contribute to the ability of elements to undergo catenation.How does catenation vary across different groups in the periodic table?

- 3. (a) How does the intercalation of guest molecules or ions into host structures affect the properties and applications of materials? Provide examples to illustrate your answer.
 - (b) Explain the structural features, methods of preparation and chemical reactions of phosphazenes. How do the P-N bonds in phosphazenes differ from typical covalent bonds?

Unit II

- 4. (a) What are the fundamental principles of molecular recognition? How do these principles guide the design of molecules for selective binding?
 - (b) How does multiple recognition work in supramolecular chemistry and what are the advantages of using multiple binding sites in the design of receptors?

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- 5. (a) Describe the design principles behind supramolecular catalysts and how they mimic or differ from natural enzymes in terms of specificity and efficiency. 7
 - (b) What are supramolecular devices? How do they utilize molecular recognition and self-assembly to perform specific functions?

Unit III

6. (a) Draw the structure of Chlorophyll and briefly discuss the photosynthesis process.

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- (b) Write notes on the following:
 - (i) Ferredoxin
 - (ii) Ruberidoxin.
- 7. (a) Write notes on the following: 8
 - (i) Xanthine Oxidase
 - (ii) Carbonic Anhydrase.
 - (b) Describe the role of Vitamin-B₁₂ in the biological system in detail.

Unit IV

- 8. (a) What are essential and trace elements?

 Discuss their role in biological systems. 7
 - (b) Describe the different types of chelating agents used in chelate therapy, including their chemical properties and specific mechanisms of action.
- 9. Write down the causes, harmful effects and prevention of:

(a) Acid Rain

b) Particulate pollution.

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