Roll No.

(12/24)

15024

M. Sc. EXAMINATION

(For Batch 2021 & Onwards)

(Third Semester)

CHEMISTRY

MSc/Chem/3/DSC2-I

Inorganic Chemistry Special-II

Time: Three Hours Maximum Marks: 70

Note: Attempt *Five* questions in all, selecting *one* question from each Unit. Q. No. 1 is compulsory.

- 1. (i) What are chain silicates and sheet silicates?
 - (ii) Discuss the structures of XeO₂F₂.
 - (iii) Give two examples of mononuclear polyhydrides.

- (iv) Why f^{14} lanthanide complexes are colourless? Explain.
- (v) What is the electronic structure of actinides? $2 \times 5 = 10$

Unit I

- 2. (a) Briefly discuss pyrophyllite by taking suitable examples. 5
 - (b) How metal carbide can help in synthesizing metallofullerenes?
 - (c) Discuss the synthesis of hyponitrous acid. 5
- What are the composition and structure of silicones and organosilicones? Also, discuss the preparation and properties of these.

Unit II

- 4. (a) Discuss the synthesis of transition metal hydrides. Also discuss chemical behavior towards the hydride compounds? 10
 - (b) Write a note on non-classical hydrides. 5

- 5. (a) Draw the structure of [HCr₂(CO)₁₀]- and also discuss the bonding in this complex.
 - (b) How HCo(CO)₄ complex can act as a pseudohalides?
 - (c) Briefly discuss the significance of M-H interactions in chemical reactions. 5

Unit III

- 6. (a) Write a short note on lanthanide shift reagents.
 - (b) Briefly explain the origin of colour in lanthanide complexes. 7
- 7. (a) What is effective magnetic moment (μ_{eff})? Calculate μ_{eff} of the $_{62}\mathrm{Sm}^{3+}$ and $_{69}\mathrm{Tm}^{3+}$ ions.
 - (b) Write short notes on the following: 7
 - (i) binary
 - (ii) ternary compounds of lanthanides.

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Unit IV

8.	(a)	Which actinide elements are found in
		earth's crust and what is their value in
		ppm? What are control rods? Give two
		examples. 8

- (b) Briefly discuss the oxidation state of actinides. Also explain actinides contraction in detail.
- 9. (a) Why the actinides have more tendency of forming complexes than lanthanides?

 Give an example of uranium cyclopentadienyl complex and draw its structure.
 - (b) Discuss radioactive behavior of uranium.